

---

---

**Animal and vegetable fats and oils —  
Separation of lipid classes by capillary  
gas chromatography (fingerprint  
method)**

*Corps gras d'origines animale et végétale — Séparation des classes  
lipidiques par chromatographie en phase gazeuse sur colonne  
capillaire (méthode fingerprint)*

STANDARDSISO.COM : Click to view the full PDF of ISO/TS 22115:2021



STANDARDSISO.COM : Click to view the full PDF of ISO/TS 22115:2021



**COPYRIGHT PROTECTED DOCUMENT**

© ISO 2021

All rights reserved. Unless otherwise specified, or required in the context of its implementation, no part of this publication may be reproduced or utilized otherwise in any form or by any means, electronic or mechanical, including photocopying, or posting on the internet or an intranet, without prior written permission. Permission can be requested from either ISO at the address below or ISO's member body in the country of the requester.

ISO copyright office  
CP 401 • Ch. de Blandonnet 8  
CH-1214 Vernier, Geneva  
Phone: +41 22 749 01 11  
Email: [copyright@iso.org](mailto:copyright@iso.org)  
Website: [www.iso.org](http://www.iso.org)

Published in Switzerland

# Contents

	Page
Foreword .....	iv
<b>1 Scope</b> .....	<b>1</b>
<b>2 Normative references</b> .....	<b>1</b>
<b>3 Terms and definitions</b> .....	<b>1</b>
<b>4 Principle</b> .....	<b>1</b>
<b>5 Reagents</b> .....	<b>2</b>
<b>6 Apparatus</b> .....	<b>3</b>
<b>7 Sample</b> .....	<b>3</b>
7.1 Sampling .....	3
7.2 Preparation of test sample .....	3
<b>8 Procedure</b> .....	<b>3</b>
8.1 Preparation of the internal standard, tridecanoylglycerol (5.2.5), $c = 20$ mg/ml .....	3
8.2 Preparation of the individual standard solutions for determination of response factors .....	3
8.3 Silylation of the standards .....	4
8.4 Preparation of the sample .....	4
8.5 Sample solution .....	4
8.6 Silylation of the sample .....	4
8.7 Gas chromatography .....	4
8.8 Peak identification and integration .....	5
<b>9 Result of the determination</b> .....	<b>6</b>
9.1 Calculation of the response factor .....	6
9.2 Quantitative determination .....	6
<b>10 Precision of the method</b> .....	<b>7</b>
10.1 Interlaboratory test .....	7
10.2 Repeatability .....	7
10.3 Reproducibility .....	7
<b>11 Test report</b> .....	<b>7</b>
<b>Annex A (informative) Typical chromatograms</b> .....	<b>8</b>
<b>Annex B (informative) Results of an interlaboratory test</b> .....	<b>18</b>
<b>Bibliography</b> .....	<b>22</b>

## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see [www.iso.org/directives](http://www.iso.org/directives)).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see [www.iso.org/patents](http://www.iso.org/patents)).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation of the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT), see [www.iso.org/iso/foreword.html](http://www.iso.org/iso/foreword.html).

This document was prepared by Technical Committee ISO/TC 34, *Food products*, Subcommittee SC 11, *Animal and vegetable fats and oils*, in collaboration with the European Committee for Standardization (CEN) Technical Committee CEN/TC 307, *Oilseeds, vegetable and animal fats and oils and their by-products — Methods of sampling and analysis*, in accordance with the Agreement on technical cooperation between ISO and CEN (Vienna Agreement).

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at [www.iso.org/members.html](http://www.iso.org/members.html).

# Animal and vegetable fats and oils — Separation of lipid classes by capillary gas chromatography (fingerprint method)

## 1 Scope

This document specifies a method for the semi-quantitative analysis of oils, fats and oil/fat-related samples (deodistillates).

It is applicable to the screening of oils, fats and oil/fat-related samples to obtain main (e.g. triglycerides) and minor (e.g. sterols, sterol esters, tocopherols, wax esters, fatty alcohols, glycerol) component information in one single analysis. For a truly quantitative analysis of pre-identified compound classes, specific methods are more appropriate.

The method can also be used as a useful qualitative screening tool for the relative comparison of sample compositions.

## 2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 661, *Animal and vegetable fats and oils — Preparation of test sample*

## 3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at <https://www.iso.org/obp>
- IEC Electropedia: available at <http://www.electropedia.org/>

## 4 Principle

The hydroxylated compounds are transformed into silyl derivatives. This operation has no effect on the apolar (non-hydroxylated) compounds also present in the sample. The sample prepared is analysed by gas chromatography (GC) on a high-temperature capillary column with a low film thickness, with an on-column injector and flame-ionization detector.

For quantitative purposes, the compounds are quantified in the presence of an internal standard (1,2,3-tridecanoylglycerol) and the response factors are determined from a reference standard from each class.

## 5 Reagents

**WARNING — Attention is drawn to the regulations which specify the handling of hazardous substances. Technical, organizational and personal safety measures shall be followed.**

During the analysis, unless otherwise stated, use only reagents of recognized analytical grade and distilled or demineralized water or water of equivalent purity.

**5.1 Silylation reagent mixture:** Methylimidazole (CAS 616-47-7) and MSHFBA (CAS 53296-64-3).

Prepare a mixture of silylation reagents: 1 ml of MSHFBA + 50 µl of Methylimidazole.

NOTE This mixture cannot be stored more than one week due to moisture absorption. Indeed, silyl derivatives are moisture sensitive.

**5.2 Reference substances<sup>1)</sup>**

**5.2.1 Oleic acid** (CAS 112-80-1).

**5.2.2 1-Monoolein** (CAS 111-03-5).

**5.2.3 1,3-Diolein** (CAS 2465-32-9).

**5.2.4 Triolein** (CAS 122-32-7).

**5.2.5 Tridecanoylglycerol** (CAS 621-71-6).

**5.2.6 Eicosanol** (CAS 629-96-9).

**5.2.7 α-Tocopherol** (CAS 10191-41-0), **γ tocopherol** (CAS 54-28-4), **δ tocopherol** (CAS 119-13-1).

**5.2.8 Cholesterol** (CAS 57-88-5).

**5.2.9 Cholesterol palmitate** (CAS 601-34-3).

**5.2.10 Squalene** (CAS 111-02-4).

**5.2.11 Plant sterol mix<sup>2)</sup>** (CAS 474-67-9, 474-62-4, 83-48-7, 83-46-5).

**5.3 Isooctane**, trace organic analysis grade, purity 99 % min.

**5.4 Chloroform**, chromatographic quality.

---

1) Suitable suppliers are Sigma-Aldrich ([www.sigmaaldrich.com](http://www.sigmaaldrich.com)) or Larodan ([www.larodan.com](http://www.larodan.com)). This information is given for the convenience of users of this document and does not constitute an endorsement by ISO of this supplier.

2) A suitable supplier is Larodan ([www.larodan.com](http://www.larodan.com)). This information is given for the convenience of users of this document and does not constitute an endorsement by ISO of this supplier.

## 6 Apparatus

Usual laboratory equipment and, in particular, the following.

**6.1 Volumetric flask**, 100 ml and 10 ml.

**6.2 Oven**, at  $103\text{ }^{\circ}\text{C} \pm 1\text{ }^{\circ}\text{C}$

**6.3 Conical glass sample vials**, 10 ml capacity.

**6.4 Gas chromatograph for capillary columns**, equipped with an on-column injector or equivalent device, a temperature-programmable oven and a flame ionization detector (FID).

**6.5 Fused silica capillary column**, capable of being programmed up to  $400\text{ }^{\circ}\text{C}$  ("high temperature" type) for which the following characteristics are advised: 100 % dimethylpolysiloxane or 95 % dimethyl/5 % diphenyl polysiloxane stationary phase, length 30 m, internal diameter 0,32 mm or 0,25 mm, film thickness 0,1  $\mu\text{m}$ . Other columns of similar polarity and selectivity may also be used.

**6.6 Microsyringe**, 5  $\mu\text{l}$  to 10  $\mu\text{l}$  capacity, suitable for on-column injection in gas chromatography.

**6.7 Analytical balance**, reading accuracy 0,001 g.

**6.8 Ultrasonic bath**.

**6.9 Nitrogen evaporator**.

## 7 Sample

### 7.1 Sampling

Sampling is not part of the method specified in this document. A recommended sampling method is given in ISO 5555<sup>[1]</sup>.

It is important that the laboratory receive a sample which is truly representative and has not been damaged or changed during transport or storage.

### 7.2 Preparation of test sample

Prepare the test sample in accordance with ISO 661.

## 8 Procedure

### 8.1 Preparation of the internal standard, tridecanoylglycerol (5.2.5), $c = 20\text{ mg/ml}$

Weigh 200 mg tridecanoylglycerol (5.2.5) into a 10 ml volumetric flask, fill it up with isooctane (5.3).

### 8.2 Preparation of the individual standard solutions for determination of response factors

A 5 mg/ml standard solution in isooctane (5.3) is prepared for each compound (see Table 1) to determine the response factor for each component group.

**Table 1 — List of the standards for the calculation of the response factors**

Class determined	Standard
Alcohols	Eicosanol (5.2.6)
Free fatty acids	Oleic acid (5.2.1)
Hydrocarbons	Squalene (5.2.10)
Monoglycerides	Monoolein (5.2.2)
Tocopherols	$\alpha$ -Tocopherol (5.2.7)
Sterols	Cholesterol (5.2.8)
Diglycerides	Diolein (5.2.3)
Sterol esters	Cholesterol palmitate (5.2.9)
Triglycerides	Triolein (5.2.4)

### 8.3 Silylation of the standards

Introduce 100  $\mu$ l of the internal standard solution, tridecanoylglycerol (5.2.5),  $c = 20$  mg/ml, and 200  $\mu$ l of the individual standard solution (each component separately) into a silylation tube (6.3) and evaporate the solvent to dryness under nitrogen (6.9). Introduce 100  $\mu$ l of the silylation mixture (5.1). Keep at 103 °C for 15 min (6.2), then take up into 5 ml isooctane (5.3).

It is recommended when preparing the following solution to check the method precision (optional).

In a 100 ml volumetric flask (6.1), exactly weigh about 50 mg of each standard (5.2), including 50 mg of the internal standard solution, tridecanoylglycerol (5.2.5), and dilute to 100 ml with chloroform (5.4).

This solution can be stored at ambient temperature for six months.

To keep the solution homogenous over time, the solution has to be placed into an ultrasonic bath for 10 min before use.

Introduce 0,5 ml of the solution into a silylation tube (6.3) and dry under nitrogen (6.9). Introduce 100  $\mu$ l of the silylation mixture (5.1). Keep at 103 °C for 15 min (6.2), then take up into 10 ml isooctane (5.3).

### 8.4 Preparation of the sample

If the sample is solid at room temperature, completely melt the sample in a 103 °C oven (6.2) and homogenize it.

### 8.5 Sample solution

In a 10 ml volumetric flask (6.1), exactly weigh about 100 mg of sample. Introduce exactly 1 ml of internal standard solution (see 8.1) and dilute to 10 ml with isooctane (5.3).

If the sample is not totally soluble in isooctane, chloroform or toluene may be used instead of isooctane.

### 8.6 Silylation of the sample

Introduce 0,5 ml of the sample solution (see 8.5) into a silylation tube (6.3) and evaporate the solvent to dryness under nitrogen (6.9). Introduce 100  $\mu$ l of the silylation mixture (5.1). Keep at 103 °C for 15 min (6.2), then take up into 10 ml isooctane (5.3).

### 8.7 Gas chromatography

Install the column in the gas chromatograph and check the working conditions by injecting the solvent, isooctane (5.3). The baseline should be straight with a small positive drift. If the drift is high, proceed to condition the column. For a negative drift, check the connections of the column.

If the column is being used for the first time, it is necessary to condition the column by heating it in the oven using a temperature gradient up to 370 °C (depending on the oven temperature chosen for the analysis) in 4 h. Maintain the temperature for 2 h.

A 1 m long, 0,5 mm diameter, empty fused silica capillary column is recommended as a pre-column.

The conditions in [Table 2](#) for the gas chromatograph have been found to give useful chromatograms.

The solvent vent mode is recommended if the inlet in use is capable of using it.

Optimize the temperature programme and the velocity of the carrier gas flow so that chromatograms similar to the ones presented in [Figures A.1](#) to [A.10](#) are obtained.

**Table 2 — Gas chromatographic conditions**

Function	Conditions
Column	HP1 (30 m long, 0,25 mm i.d., 0,10 µm film thickness)
Oven temperature	Initial temperature 60 °C, programmed at 20 °C/min to 170 °C, 10 °C/min to 360 °C (hold 10 min)
Carrier gas	Linear velocity 87,7 cm/sec (carrier gas used H <sub>2</sub> )
Detector temperature	370 °C
Injector	Cold on-column
Injection volume	1 µl of the solution prepared in <a href="#">8.6</a>

## 8.8 Peak identification and integration

Identify the peaks with the standard solution prepared in [8.2](#). Typical chromatograms are presented in [Annex A](#). The relative retention times of identified compounds or families of compounds are listed in [Table 3](#).

When the compound of interest is present in the form of a single peak, integration according to the conventional criteria is used. By family, the resolution level is not always obtained as there can be some minor compounds. It is then recommended to integrate the set of peaks together or to combine them in a sum.

**Table 3 — Indicative relative retention times of identified compounds or families of compounds**

Relative retention time (RRT) to tridecanoylglycerol	Determination or class of compounds
< 0,53	Alcohols ( <i>silyl derivative</i> )
0,57 to 0,63	Free fatty acids ( <i>silyl derivative</i> )
0,87	Squalene
0,82 to 0,86	Monoglycerides ( <i>silyl derivative</i> )
0,90	Delta-tocopherol ( <i>silyl derivative</i> )
0,94	Gamma-tocopherol ( <i>silyl derivative</i> )
0,99	α-Tocopherol ( <i>silyl derivative</i> )
0,98 to 1,05	Sterols ( <i>silyl derivative</i> )
1,00	Tridecanoylglycerol ( <i>internal standard</i> )
1,30 to 1,36	Diglycerides ( <i>silyl derivative</i> )
1,52 to 1,57	Sterol esters
> 1,72	Triglycerides

## 9 Result of the determination

### 9.1 Calculation of the response factor

Calculate the response factor,  $F$ , for each family representative compound  $x$  as given in [Formula \(1\)](#):

$$F_x = \frac{A_{IS} \cdot m_x}{m_{IS} \cdot A_x} \quad (1)$$

where

$F_x$  is the response factor of compound  $x$  associated with the family X;

$m_{IS}$  is the mass of tridecanoylglycerol (internal standard), in mg;

$m_x$  is the mass of reference compound  $x$ , in mg;

$A_{IS}$  is the area of tridecanoylglycerol (internal standard);

$A_x$  is the area of reference compound  $x$ .

Examples of indicative response factors are listed in [Table 4](#).

If very different response factors are observed from those given in [Table 4](#), for example, this can occur in the case of squalene, it is recommended that an alternative internal standard molecule is used that is chemically similar to the compound in question.

**Table 4 — Indicative response factors of standards relative to tridecanoylglycerol (internal standard)**

Class determined	Standard	Response factor
Alcohols	Eicosanol	0,70
Free fatty acids	Oleic acid	1,04
Hydrocarbons	Squalene	0,89
Monoglycerides	Monolein	1,19
Tocopherols	$\alpha$ -Tocopherol	0,90
Sterols	Cholesterol	0,75
Internal standard	Tridecanoylglycerol	1,00
Diglycerides	Diolein	1,03
Sterol esters	Cholesterol palmitate	1,05
Triglycerides	Triolein	1,14

### 9.2 Quantitative determination

The mass fraction of the compounds from the family X is calculated as given in [Formula \(2\)](#):

$$w_x = \frac{F_x \cdot m_{IS} \cdot \sum A_x \cdot 100}{A_{IS} \cdot m} \quad (2)$$

where

- $w_x$  is the mass fraction of the compounds of the family X (in mg/100 mg);
- $F_x$  is the response factor of compound x associated with the family;
- $m_{IS}$  is the mass of tridecanoylglycerol (internal standard), in mg;
- $A_{IS}$  is the area of tridecanoylglycerol (internal standard);
- $\sum A_x$  is the sum of the areas of the compounds of the family X.
- $m$  is the sample mass, in mg.

## 10 Precision of the method

### 10.1 Interlaboratory test

Details of the test and the precision of the method are summarized in [Annex B](#). The values derived from this interlaboratory test might not be applicable to concentration ranges and matrices other than those given.

### 10.2 Repeatability

The absolute difference between two independent single test results, obtained with the same method on identical test material in the same laboratory by the same operator using the same equipment within a short interval of time, shall in not more than 5 % of cases exceed the value of  $r$  given in [Table B.1](#).

### 10.3 Reproducibility

The absolute difference between two single test results, obtained with the same method on identical test material in different laboratories by different operators using different equipment, shall in not more than 5 % of cases exceed the value of  $R$  given in [Table B.1](#).

## 11 Test report

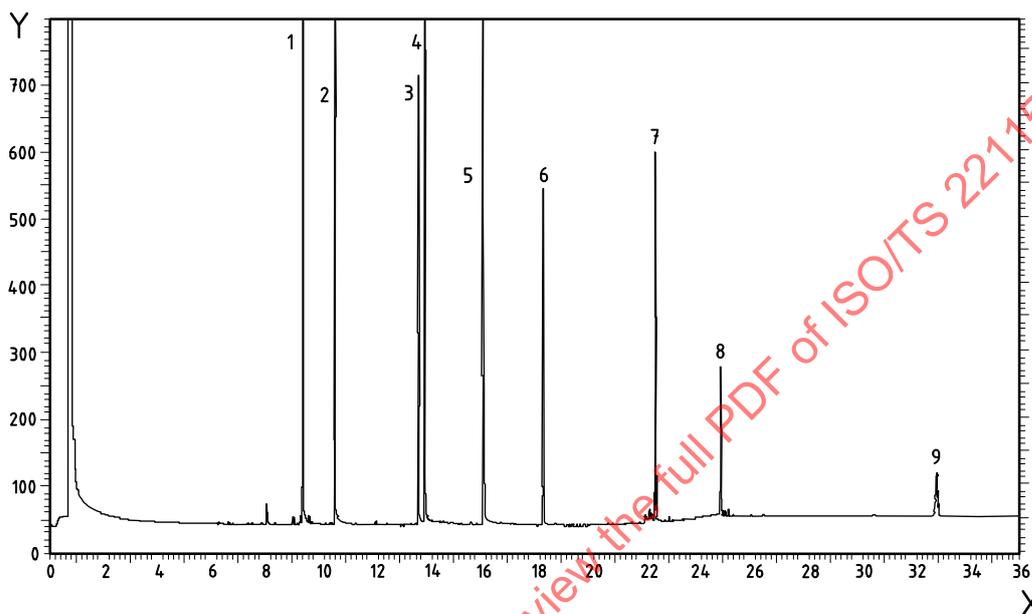
The test report shall contain at least the following information:

- a) all information necessary for the complete identification of the sample;
- b) the sampling method used, if known;
- c) the test method used, with reference to this document, i.e. ISO/TS 22115:2021;
- d) all operating details not specified in this document, or regarded as optional, together with details of any incidents occurred when performing the method, which can have influenced the test result(s);
- e) the test result obtained;
- f) if the repeatability has been checked, the final quoted result obtained.

## Annex A (informative)

### Typical chromatograms

See [Figures A.1](#) to [A.10](#) for example chromatograms.



#### Key

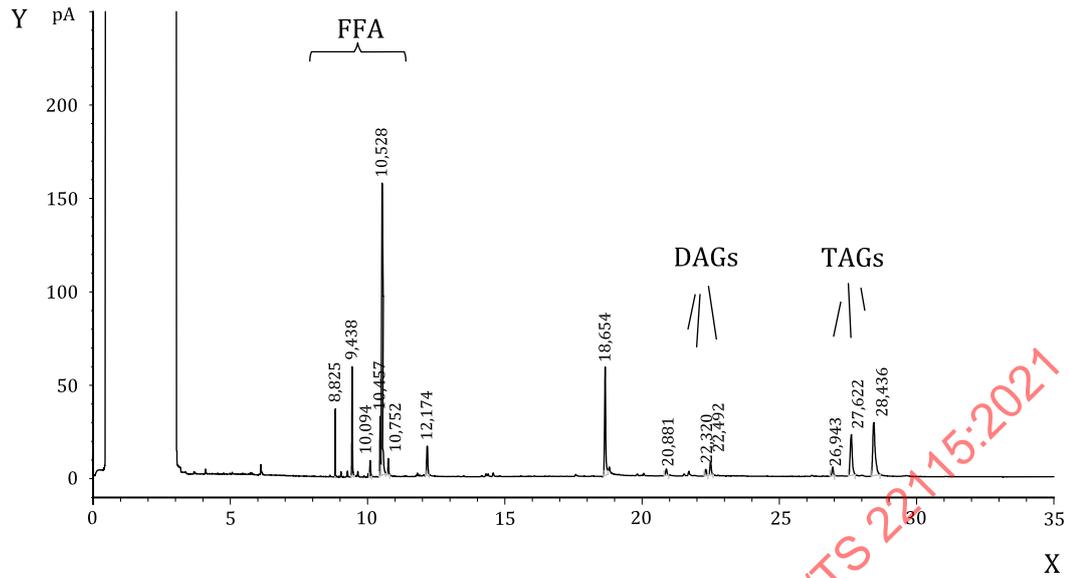
X minutes

Y millivolts

Identification of peaks:

Retention time (min)	Compound	Retention time (min)	Compound
9,387	Oleic acid [1]	18,317	Tridecanoylglycerol (IS) [6]
10,593	Eicosanol [2]	22,487	Diolenin [7]
13,683	Monoolein [3]	24,917	Cholesterol palmitate [8]
13,933	Squalene [4]	32,940	Triolein [9]
16,093	$\alpha$ -Tocopherol [5]		

**Figure A.1 — Gas chromatogram obtained from the reference solution**

**Key**

X minutes

FFA free fatty acids

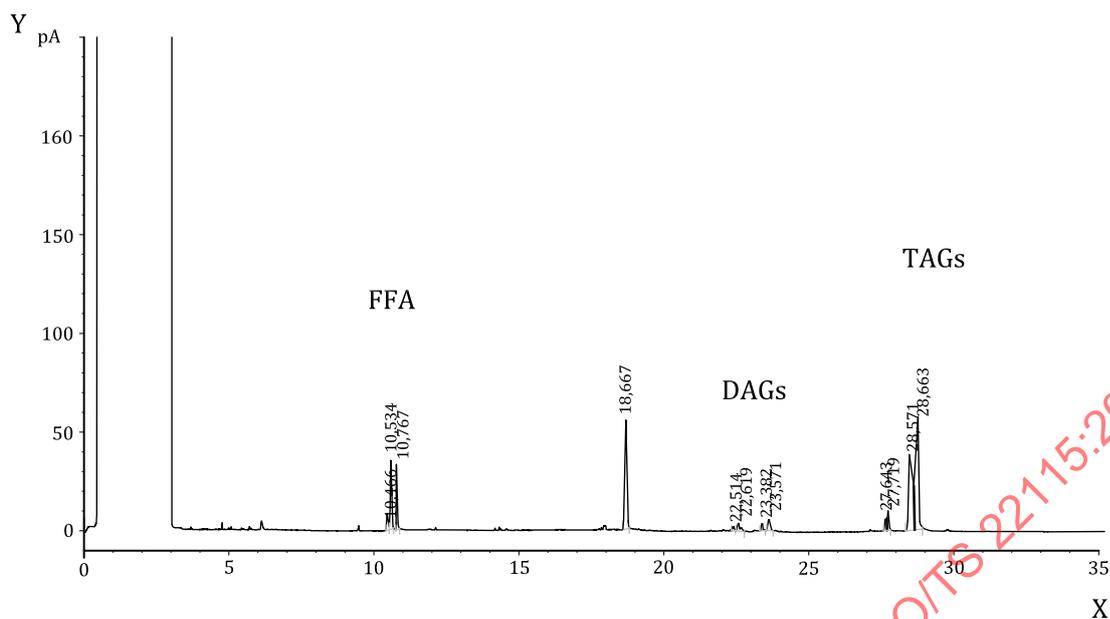
DAG diglycerides

TAG triglycerides

Identification of peaks:

Retention time (min)	Compound	Retention time (min)	Compound
8,825	Myristic acid	18,654	Tridecanoylglycerol (IS)
9,438	Palmitic acid	22,320 to 22,492	C18-Diglycerides
10,528	Oleic acid	26,943 to 28,436	Triglycerides
10,752	Stearic acid		

**Figure A.2 — Gas chromatogram obtained from an olive pomace oil**



**Key**

X minutes

DAG diglycerides

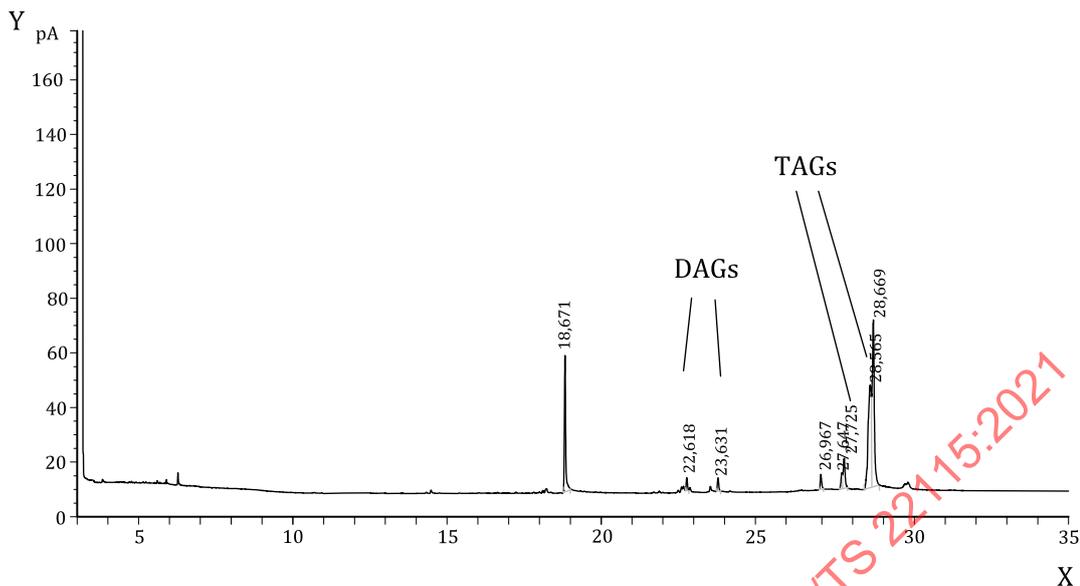
FFA free fatty acids

TAG triglycerides

Identification of peaks:

Retention time (min)	Compound	Retention time (min)	Compound
10,534	Oleic acid	22,514 to 23,382	Diglycerides
10,767	Stearic acid	27,643 to 28,663	Triglycerides
18,667	Tridecanoylglycerol (IS)		

**Figure A.3 — Gas chromatogram obtained from a crude shea butter**



**Key**

X minutes

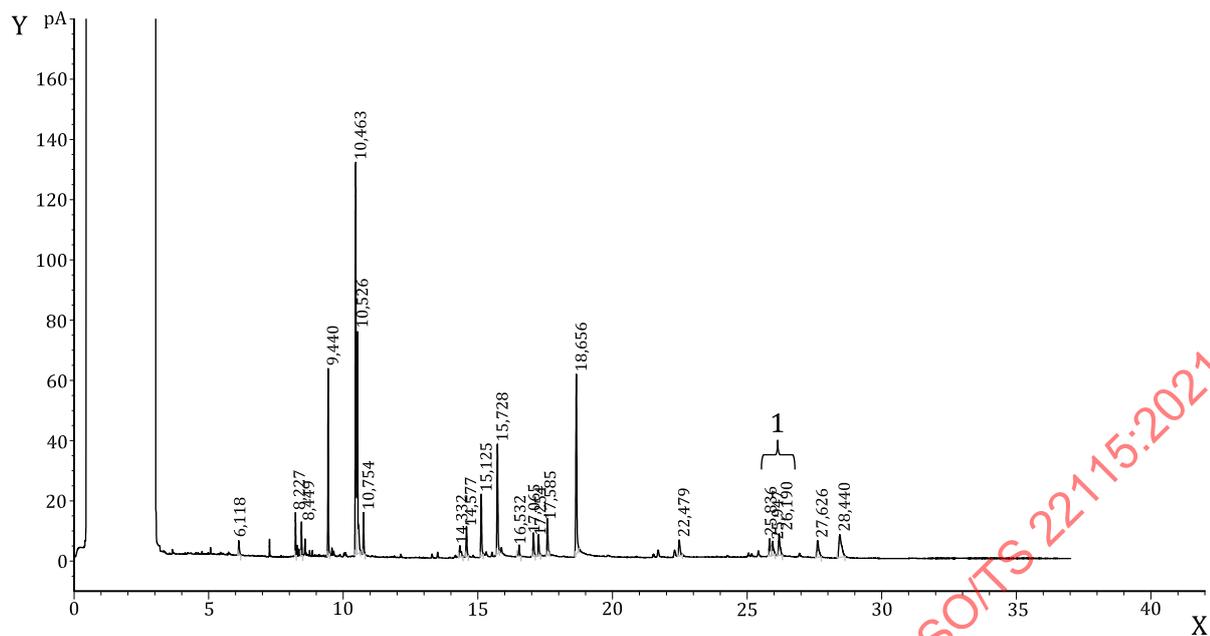
DAG diglycerides

TAG triglycerides

Identification of peaks:

Retention time (min)	Compound	Retention time (min)	Compound
18,671	Tridecanoylglycerol (IS)	27,967 to 28,669	Triglycerides
22,618 to 23,631	Diglycerides		

**Figure A.4 — Gas chromatogram obtained from a refined shea butter**



**Key**

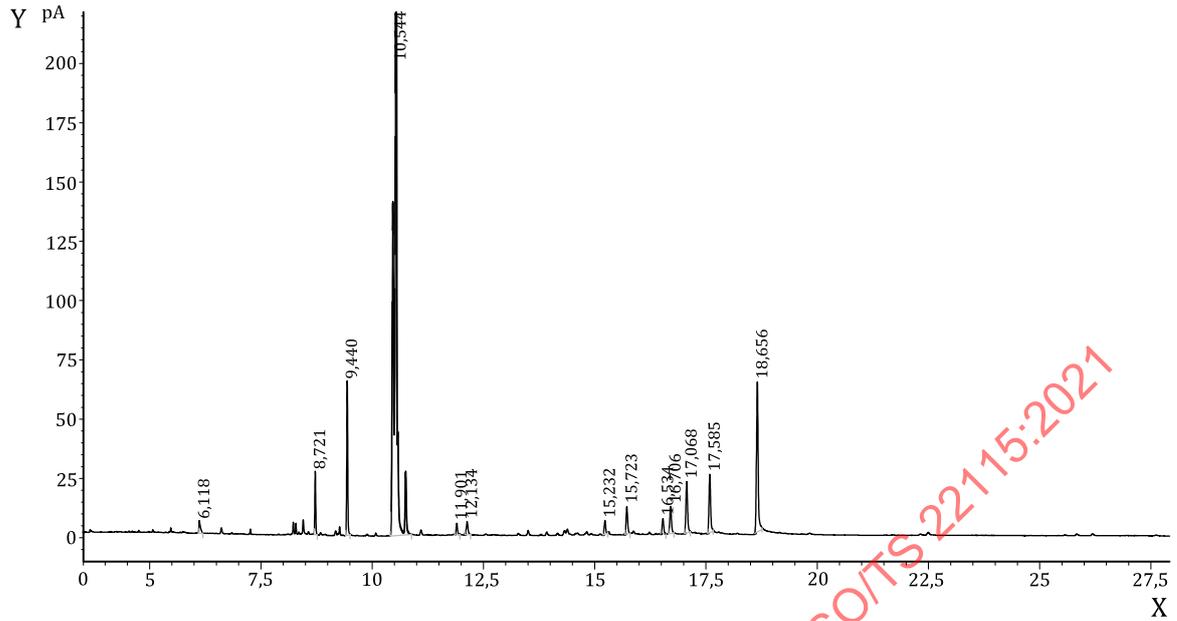
X minutes

1 sterol esters

Identification of peaks:

Retention time (min)	Compound	Retention time (min)	Compound
9,440	Palmitic acid	17,065	Campesterol
10,526	Oleic acid	17,254	Stigmasterol
14,332	C18-Monoglycerides	17,585	Beta-Sitosterol
14,577	Squalene	18,656	Tridecanoylglycerol (IS)
15,125	Delta-Tocopherol	22,479	C18-Diglycerides
15,728	Gamma-Tocopherol	25,836 to 26,190	Sterol Esters
16,532	Alpha-Tocopherol	28,440	C18-Triglycerides

**Figure A.5 — Gas chromatogram obtained from a soybean oil deodistillate**

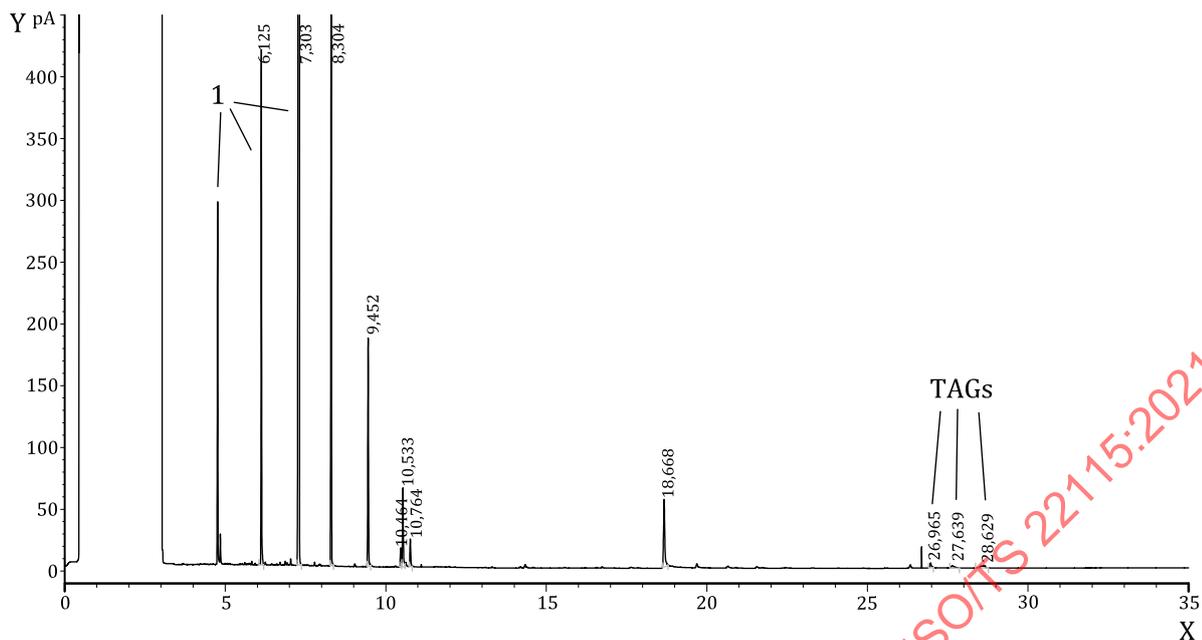
**Key**

X minutes

Identification of peaks:

Retention time (min)	Compound	Retention time (min)	Compound
8,721	Myristic acid	16,534	Alpha-Tocopherol
9,440	Palmitic acid	16,706	Brassicasterol
10,544	Oleic acid	17,068	Campesterol
15,232	Delta-Tocopherol	17,585	Beta-Sitosterol
15,723	Gamma-Tocopherol	18,656	Tridecanoylglycerol (IS)

**Figure A.6 — Gas chromatogram obtained from a rapeseed oil deodistillate**



**Key**

X minutes

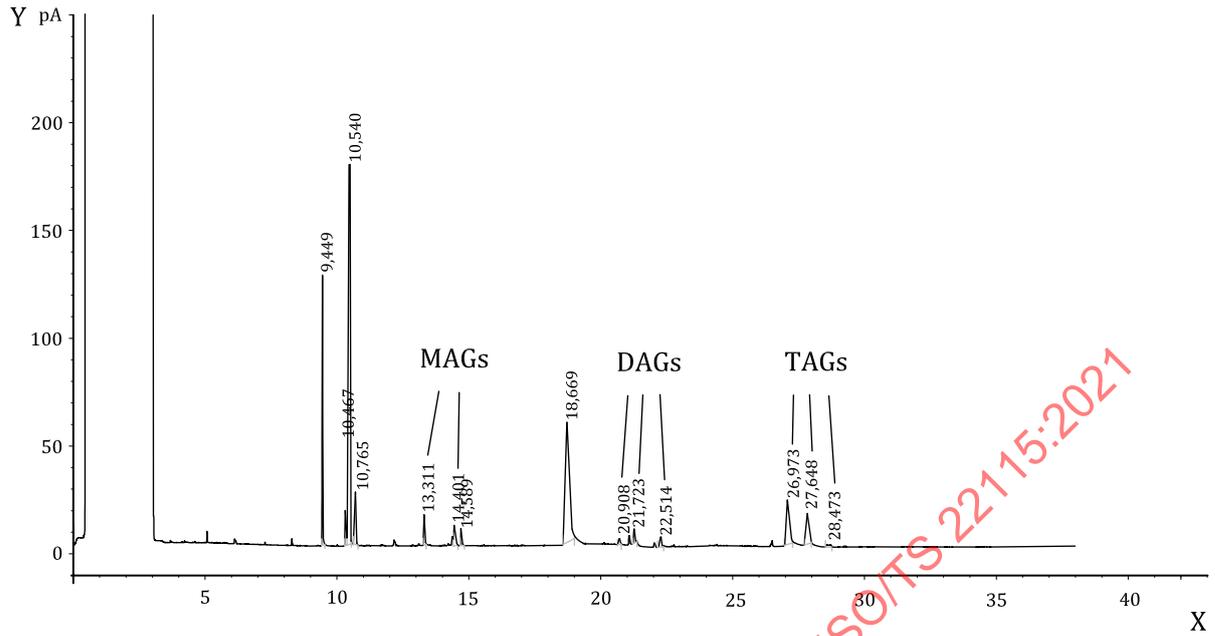
TAG triglycerides

1 fatty alcohols

Identification of peaks:

Retention time (min)	Compound	Retention time (min)	Compound
5,0 to 7,3	Fatty alcohols	10,533	Oleic acid
8,304	Lauric acid	18,668	Tridecanoylglycerol (IS)
9,452	Palmitic acid	26,965 to 28,629	Triglycerides

**Figure A.7 — Gas chromatogram obtained from a coco fat deodistillate**



**Key**

X minutes

DAG diglycerides

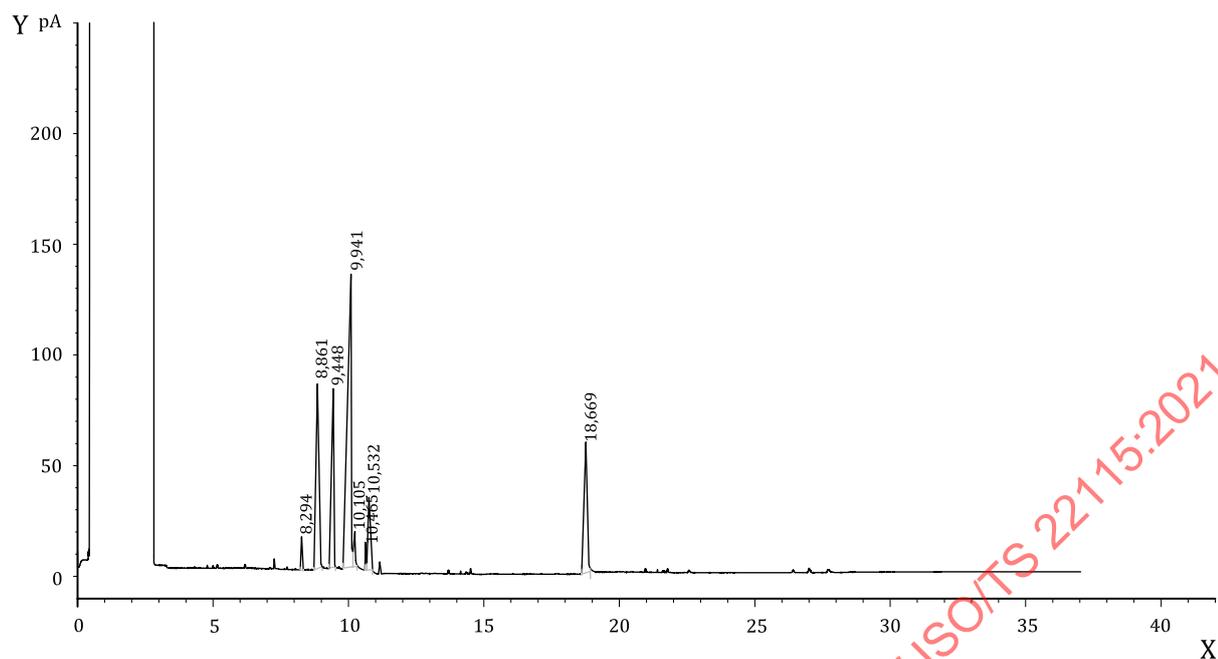
MAG monoglycerides

TAG triglycerides

Identification of peaks:

Retention time (min)	Compound	Retention time (min)	Compound
9,449	Palmitic acid	14,589	Squalene
10,540	Oleic acid	18,669	Tridecanoylglycerol (IS)
13,311	C16-Monoglycerides	20,908 to 22,514	Diglycerides
14,401	C18-Monoglycerides	26,973 to 28,473	Triglycerides

**Figure A.8 — Gas chromatogram obtained from a palm fat deodistillate**

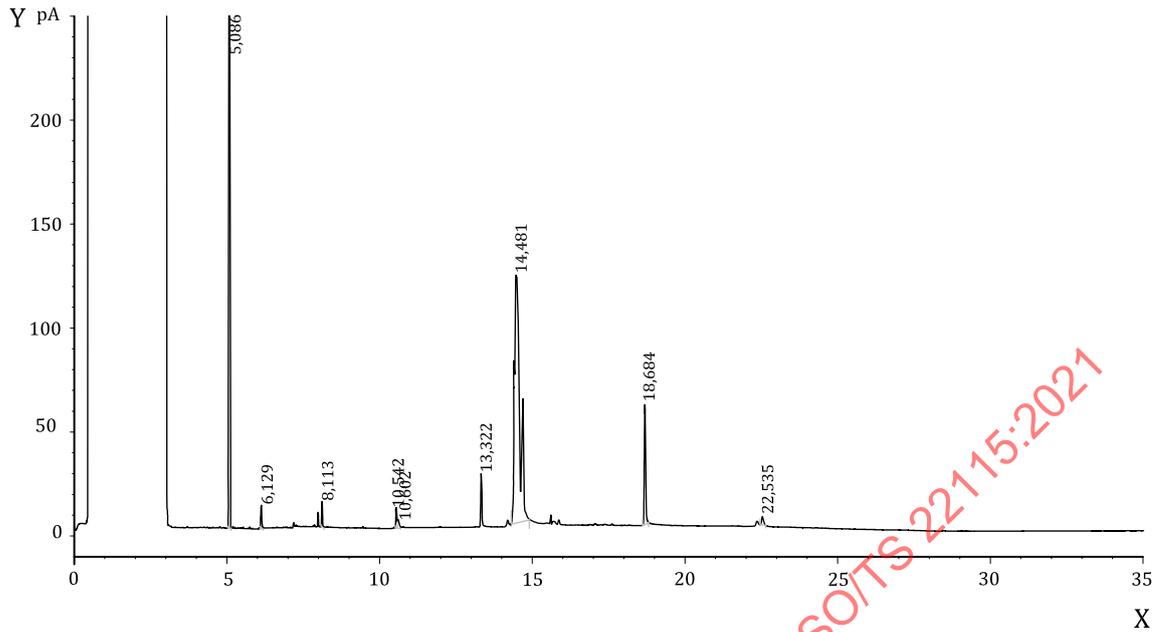
**Key**

X minutes

Identification of peaks:

Retention time (min)	Compound	Retention time (min)	Compound
8,294	Lauric acid	9,941	Stearic acid
8,861	Myristic acid	10,532	Oleic acid
9,448	Palmitic acid	18,669	Tridecanoylglycerol (IS)

**Figure A.9 — Gas chromatogram obtained from a palm fat deodistillate-olein fraction**

**Key**

X minutes

Identification of peaks:

Retention time (min)	Compound	Retention time (min)	Compound
8,113	Octadecane	14,481	C18-Monoglycerides
10,542	Oleic acid	18,684	Tridecanoylglycerol (IS)
13,322	C16-Monoglycerides	22,535	C18-Diglycerides

**Figure A.10 — Gas chromatogram obtained from an emulsifier (90 % monoglyceride content)**

## Annex B (informative)

### Results of an interlaboratory test

An international collaborative test involving 6 laboratories (out of 19 applicants) in 5 countries was carried out on 5 oil samples. The test was organized by Bunge Katalin Kóvári Innovation Center and the results obtained were subjected to statistical analysis in accordance with ISO 5725-1<sup>[2]</sup> and ISO 5725-2<sup>[3]</sup> to give the precision data shown in [Tables B.1](#) to [B.9](#).

**Table B.1 — Summary of statistical results of free fatty acids for sample A to sample E**

	Sample				
	A	B	C	D	E
Number of participating laboratories, $n_p$	6	6	6	5	6
Number of laboratories retained after eliminating outliers, $n_p$	5	6	6	5	5
Number of individual test results of all laboratories on each sample, $n_z$	2	2	2	2	2
Mean value, $\bar{w}$ , %	57,00	58,77	16,52	1,24	71,43
Repeatability standard deviation, $s_r$ , %	4,03	4,75	1,37	0,12	0,72
Repeatability coefficient of variation, $C_{V,r}$ , %	7,07	8,09	8,29	9,63	1,01
Repeatability limit, $r$ , (= $s_r$ 2,8), %	11,29	13,31	3,83	0,33	2,01
Reproducibility standard deviation, $s_R$ , %	7,49	7,57	1,48	1,40	6,47
Reproducibility coefficient of variation, $C_{V,R}$ , %	13,13	12,89	8,95	112,93	9,06
Reproducibility limit, $R$ , (= $s_R$ 2,8), %	20,96	21,20	4,14	3,92	18,12

**Table B.2 — Summary of statistical results of total glycerols (sum of mono-, di- and triacyl glycerols) for sample A to sample E**

	Sample				
	A	B	C	D	E
Number of participating laboratories, $n_p$	6	6	6	6	6
Number of laboratories retained after eliminating outliers, $n_p$	5	6	6	5	5
Number of individual test results of all laboratories on each sample, $n_z$	2	2	2	2	2
Mean value, $\bar{w}$ , %	8,14	6,57	7,62	18,60	7,72
Repeatability standard deviation, $s_r$ , %	1,02	0,23	0,60	3,74	3,54
Repeatability coefficient of variation, $C_{V,r}$ , %	12,57	3,53	7,88	20,10	45,91
Repeatability limit, $r$ , (= $s_r$ 2,8), %	2,87	0,65	1,68	10,46	9,92
Reproducibility standard deviation, $s_R$ , %	3,67	2,68	3,41	8,12	12,81
Reproducibility coefficient of variation, $C_{V,R}$ , %	45,09	40,86	44,78	43,67	166,05
Reproducibility limit, $R$ , (= $s_R$ 2,8), %	10,28	7,51	9,55	22,74	35,88