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**Natural gas — Determination  
of composition and associated  
uncertainty by gas chromatography —**

**Part 5:  
Isothermal method for nitrogen,  
carbon dioxide, C<sub>1</sub> to C<sub>5</sub> hydrocarbons  
and C<sub>6+</sub> hydrocarbons**

*Gaz naturel — Détermination de la composition et de l'incertitude associée par chromatographie en phase gazeuse —*

*Partie 5: Méthode isotherme pour l'azote, le dioxyde de carbone, les hydrocarbures C<sub>1</sub> à C<sub>5</sub> et C<sub>6+</sub>*



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## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 [www.iso.org/directives](http://www.iso.org/directives).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received [www.iso.org/patents](http://www.iso.org/patents).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation on the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the WTO principles in the Technical Barriers to Trade (TBT), see the following URL: Foreword - Supplementary information

The committee responsible for this document is ISO/TC 193, *Natural Gas*, Subcommittee SC 1, *Analysis of Natural Gas*.

This second edition cancels and replaces the first edition (ISO 6974-5:2000).

ISO 6974 consists of the following parts, under the general title *Natural gas — Determination of composition and associated uncertainty by gas chromatography*:

- *Part 1: General guidelines and calculation of composition*
- *Part 2: Uncertainty calculations*
- *Part 3: Determination of hydrogen, helium, oxygen, nitrogen, carbon dioxide and hydrocarbons up to C<sub>8</sub> using two capillary columns and one packed column*
- *Part 4: Determination of nitrogen, carbon dioxide and C<sub>1</sub> to C<sub>5</sub> and C<sub>6+</sub> hydrocarbons for a laboratory and on-line measuring system using two columns*
- *Part 5: Isothermal method for nitrogen, carbon dioxide, C<sub>1</sub> to C<sub>5</sub> hydrocarbons and C<sub>6+</sub> hydrocarbons*
- *Part 6: Determination of helium, oxygen, nitrogen, carbon dioxide and C<sub>1</sub> to C<sub>8</sub> hydrocarbons using three capillary columns*

## Introduction

This part of ISO 6974 describes a method for the analysis of natural gas that is commonly used for online process applications, but can be applied to laboratory instruments. The compositional data obtained are used for the calculation of calorific value, density and Wobbe index.

It is assumed that the natural gas does not contain any oxygen at source and that any oxygen which may be present is due to contamination during sampling.

The primary use of this chromatographic method is the calculation of calorific value (CV) according to ISO 6976. It is based on a column switching technique in which multiple columns, chosen for their separating ability for particular groups of components, are switched under automatic control.

Only one injection is necessary and the first phase of the method involves accelerated backflush of C<sub>6</sub>+ (which is measured as a recombined "pseudo component" rather than by the summation of individual component measurements). Lighter components (nitrogen, methane, carbon dioxide and ethane) are stored on the appropriate separating column while the heavier, C<sub>3</sub> to C<sub>5</sub> hydrocarbons are eluted. The lighter components are then separated by redirecting carrier gas on to the appropriate column.

A Thermal Conductivity Detector (TCD) is used for measurement of the above components.

When the method is first set up, the repeatability of measurement is established by repetitive analysis of a cylinder of test gas, commonly a typical natural gas. For each component, a control chart showing the mean value, and the bounds representing 2 and 3 standard deviations, is drawn up. Subsequently, this test gas is analysed after each calibration of the analyser, and the results are compared with the data in the control charts. The performance of the analyser is assessed by this procedure.

Any change in the method setup can give rise to differences in component responses and hence (where applied) to calculated uncertainties. In these circumstances fitting data to an existing control chart is not a suitable procedure, and the operations that were undertaken when the method was first set up shall be repeated.

This part of ISO 6974 provides one of the methods that may be used for determining the compositions of natural gas in accordance with ISO 6974-1 and ISO 6974-2.

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# Natural gas — Determination of composition and associated uncertainty by gas chromatography —

## Part 5:

## Isothermal method for nitrogen, carbon dioxide, C<sub>1</sub> to C<sub>5</sub> hydrocarbons and C<sub>6+</sub> hydrocarbons

### 1 Scope

This part of this International Standard describes a gas chromatographic method for the quantitative determination of the content of nitrogen, carbon dioxide and C<sub>1</sub> to C<sub>5</sub> hydrocarbons individually and a composite C<sub>6+</sub> measurement, which represents all hydrocarbons of carbon number 6 and above in natural gas samples. It is applicable to the analysis of gases containing constituents within the working ranges given in [Table 1](#).

**Table 1 — Component working ranges**

Component		Mole fraction	
		%	
		Min.	Max.
Nitrogen	N <sub>2</sub>	0,1	22
Carbon dioxide	CO <sub>2</sub>	0,05	15
Methane	CH <sub>4</sub>	34	100
Ethane	C <sub>2</sub> H <sub>6</sub>	0,1	23
Propane	C <sub>3</sub> H <sub>8</sub>	0,05	10
iso-Butane	i-C <sub>4</sub> H <sub>10</sub>	0,01	2,0
n-Butane	n-C <sub>4</sub> H <sub>10</sub>	0,01	2,0
neo-Pentane	neo-C <sub>5</sub> H <sub>12</sub>	0,005	0,35
iso-Pentane	i-C <sub>5</sub> H <sub>12</sub>	0,005	0,35
n-Pentane	n-C <sub>5</sub> H <sub>12</sub>	0,005	0,35
Hexanes +	C <sub>6+</sub>	0,005	0,35

NOTE 1 The working ranges in [Table 1](#) are those for which the method has been shown to be satisfactory, and are offered for guidance. However, there is no reason why wider ranges should not be used, provided that the successful measurement of such wider ranges has been demonstrated.

NOTE 2 Hydrocarbons above n-pentane are expressed as the “pseudo-component” C<sub>6+</sub> which is measured as one composite peak and calibrated as such. The properties of C<sub>6+</sub> are calculated from an extended analysis of the individual C<sub>6</sub> and higher hydrocarbons.

NOTE 3 Oxygen is not a normal constituent of natural gas and would not be expected to be present in gas sampled to an online instrument. If any oxygen is present as a result of air contamination, it will be measured with the nitrogen. The resulting measured (nitrogen + oxygen) value will be in error to a small extent because of the slight difference between the detector responses of oxygen and nitrogen.

NOTE 4 The helium and argon contents are assumed to be sufficiently small and unvarying that they need not be analysed for.

NOTE 5 The gas sample shall not contain any hydrocarbon condensate and/or water.

## 2 Normative references

The following documents, in whole or in part, are normatively referenced in this document and are indispensable for its application. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 6974-1, *Natural gas — Determination of composition and associated uncertainty by gas chromatography — Part 1: General guidelines and calculation of composition*

ISO 6974-2, *Natural gas — Determination of composition and associated uncertainty by gas chromatography — Part 2: Uncertainty calculations*

## 3 Principle

[Figure 1](#) is a flowchart showing the steps involved in the analytical process. It is based on more detailed flowcharts in ISO 6974-1 and ISO 6974-2, simplified to represent the procedure described in this part. References are given at each step to the relevant clause in this part and, where appropriate, to the relevant clauses in ISO 6974-1 and ISO 6974-2.

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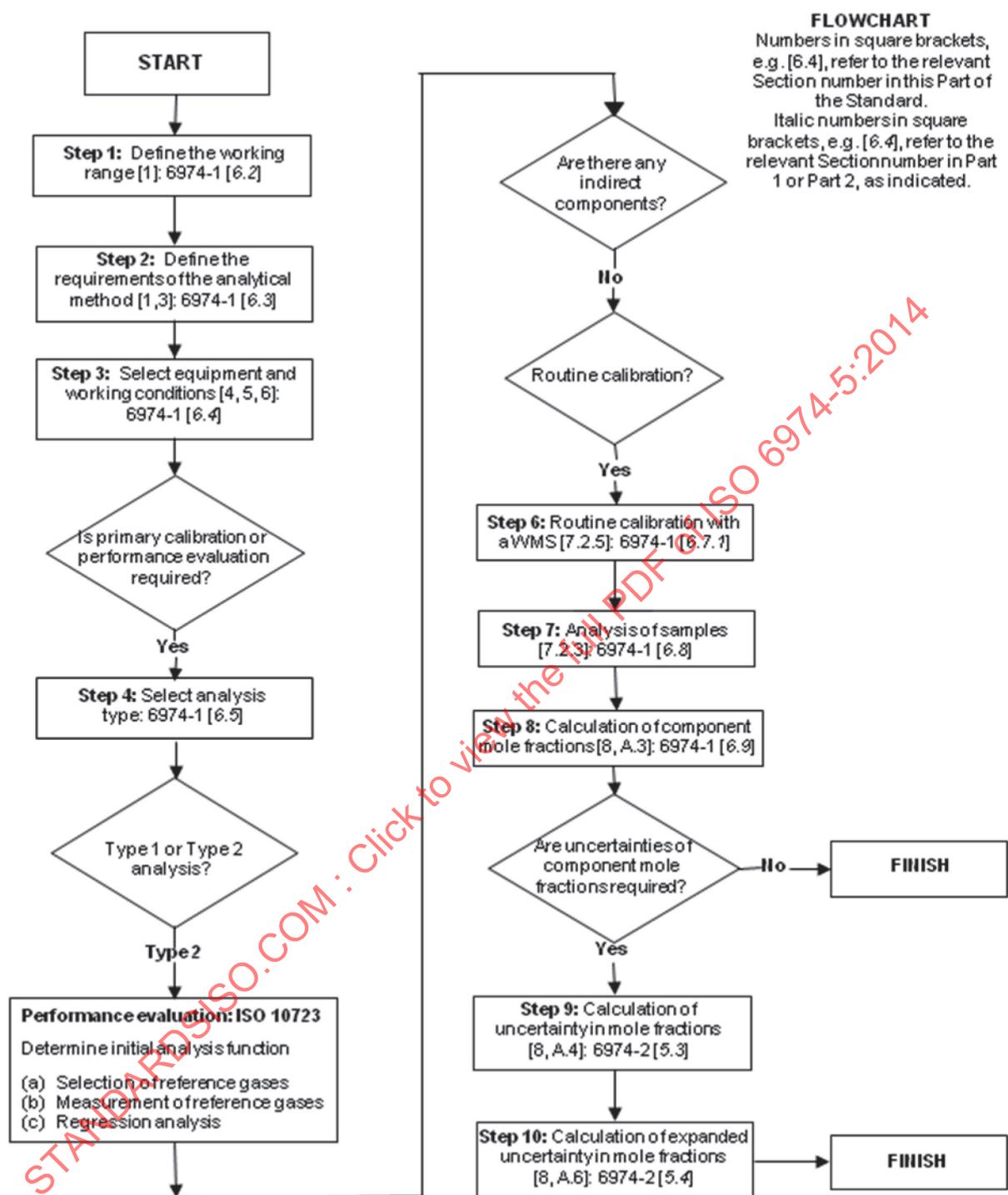


Figure 1 — Operational flowchart

NOTE The steps referred to in [Figure 1](#) are identical to the steps in flowcharts A and B in ISO 6974-1 Step 5 refers to the use of relative response factors for indirectly measured components. Indirect components are not used in this part of ISO 6974, so step 5 is not used.

The chromatographic method uses a column switching/backflush arrangement, configured as shown in [Figure 2](#). The sample is injected onto a boiling-point column which is divided into short and long sections (columns 1 and 2). The long section (column 2) provides separation of C<sub>3</sub> to C<sub>5</sub> hydrocarbons, while C<sub>6</sub> and heavier hydrocarbons are retained on the short section (column 1), from which they are

backflushed and measured by the detector as a single peak. Two six-port valves can handle the sample injection and backflushing operations, or they may be dealt with together by a single 10-port valve.

Nitrogen, carbon dioxide, methane and ethane pass rapidly and unresolved through the boiling-point column onto a porous polymer bead column (column 3), suitable for their separation. A six-port valve either connects this column or by-passes it during measurement of C<sub>3</sub> to C<sub>5</sub> components.

The separations that occur in the columns are as follows:

Column 1 Retains C<sub>6</sub>+ components ready for backflushing as one composite peak.

Column 2 Separates Propane, iso-Butane, n-Butane, neo-Pentane, iso-Pentane and n-Pentane, (which elute after C<sub>6</sub>+ has left column 1).

Column 3 Stores and separates Nitrogen, Methane, Carbon Dioxide and Ethane which elute after n-Pentane has left column 2.

## 4 Materials

4.1 **Carrier gas**, Helium (He), ≥99,995 % pure, free from oxygen and water.

4.2 **Auxiliary gases**, compressed air, for valve actuation (If consumption is low, carrier gas may be used as an alternative for valve actuation).

4.3 **Reference materials.**

4.3.1 **Reference gases**, according to ISO 6974-1.

4.3.2 **Gas mixture containing n-Pentane and 2,2-Di-Me-butane**, used to check valve timings (see [Annex B](#)).

## 5 Apparatus

5.1 **Gas chromatograph**, capable of isothermal operation and equipped with TCD.

5.2 **Column oven**, temperature range 70 °C to 105 °C, capable of being maintained to within ±0,1 °C.

5.3 **Valve oven**, controlled over the temperature range 70 °C to 105 °C, or valves fitted in the column oven.

5.4 **Pressure regulator**, to give suitable carrier gas flow rates

5.5 **Injection device**, V1, six-port sample injection valve

5.6 **Backflush valve**, V2, six-port, to allow rapid backflush of C<sub>6</sub>+ components. As described in [section 3](#), a single 10-port valve may be used for both these tasks. The operating principle is the same.

5.7 **Column isolation valve**, V3, six-port. This directs the carrier gas through the porous polymer bead column (column 3), or by-passes it.

5.8 **Columns**, The columns must satisfy the performance requirements given in [7.2.4](#). The following packing materials and column dimensions, given as examples, should be satisfactory, for use with conventional and readily available injection valves and TCDs. Any alternative combination of columns which provide similar separations and satisfy the performance requirements may be used. Micro-packed

or even capillary columns can be chosen, with appropriately sized injection and detector systems, in which case packing or coating details would be different.

## 5.9 Tube and packing.

### 5.9.1 Configuration 1

**5.9.1.1 Column 1**, 28 % DC200/500 on 45/60 mesh chromosorb P-AW, 0,75 m (2,5 foot) long, 2 mm i.d. (1/8 in o.d.).

**5.9.1.2 Column 2**, 28 % DC200/500 on 45/60 mesh chromosorb P-AW, 5,2 m (17 foot) long, 2 mm i.d. (1/8 in o.d.).

**5.9.1.3 Column 3**, 15 % DC200/500 on 50/80 mesh Porapak N, 2,4 m (8 ft) long, 2 mm i.d. (1/8 in o.d.).

### 5.9.2 Configuration 2.

**5.9.2.1 Column 1**, oxy-dipropionitrile on Porasil C, 0,3 m (1 foot) long, 0,75 mm i.d. (1/16 in o.d.).

**5.9.2.2 Column 2**, 20 % SF-96 on 80-100 mesh chromosorb W, 2,1 m (7 foot) long, 0,75 mm i.d. (1/16 in o.d.).

**5.9.2.3 Column 3**, HayeSep N, 2,1 m (7 foot) long, 0,75 mm i.d. (1/16 in o.d.).

**5.10 Method of packing**, any method which results in uniform column packing may be used.

NOTE The following method is suitable.

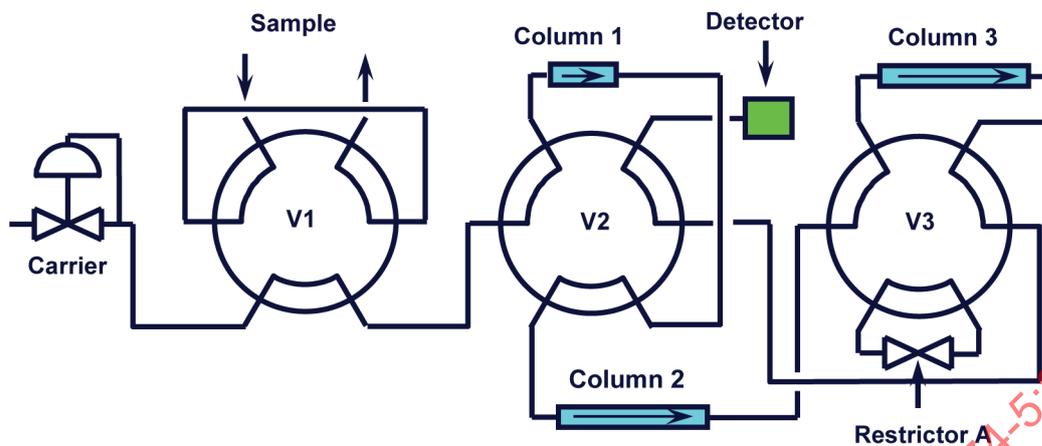
The column outlet is closed with a sintered disc or glass wool plug. A reservoir containing rather more packing than is needed to fill the column is connected to the inlet and a pressure of 0,4 MPa of nitrogen is applied to this reservoir. The flow of packing into the column is assisted by vibration. When the column is full, allow the pressure to decay slowly before disconnecting the reservoir.

**5.11 Thermal Conductivity Detector (TCD)**, with a time constant of not greater than 0,1 s, and internal volume appropriate for the column sizes and flow rate used.

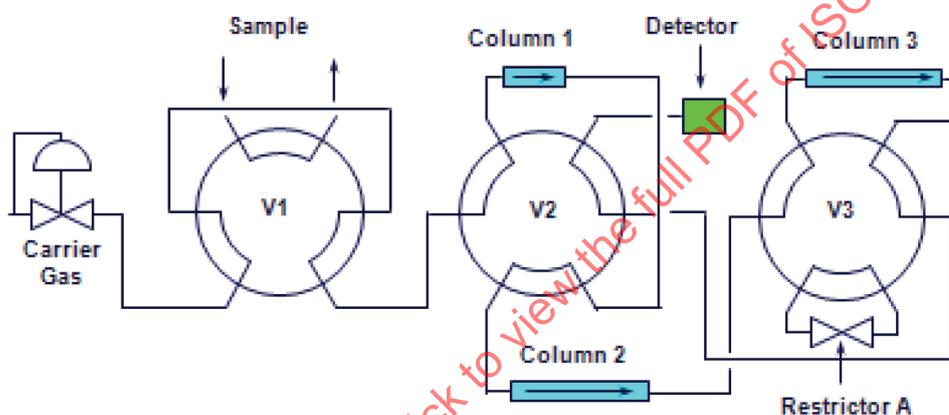
**5.12 Controller/Peak Measurement System.** Wide range (0 V to 1 V), capable of measuring peaks on a sloping baseline. Be enabled to control automatic operation of the valves according to a sequence selected by the operator.

**5.13 Auxiliary valves, tubing and other accessories**, to control the flow of sample gas to the chromatograph and for shutting off this flow for a defined period of time before injection.

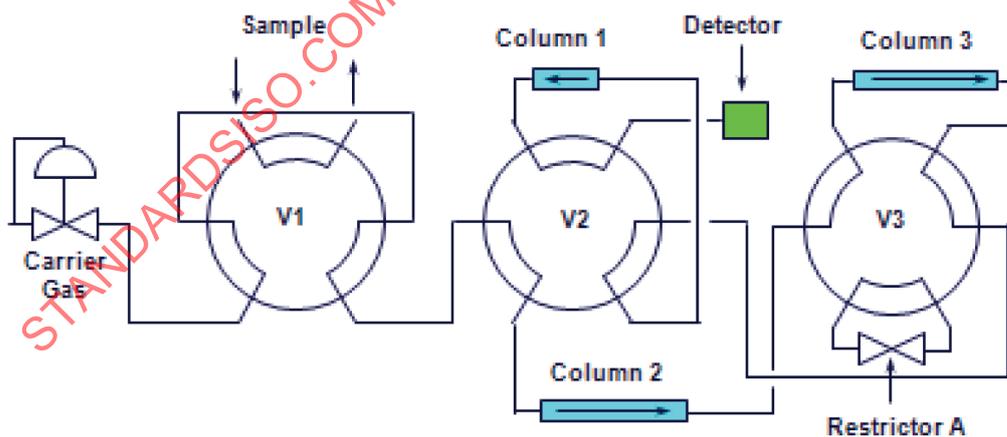
## 6 Scheme of the configuration



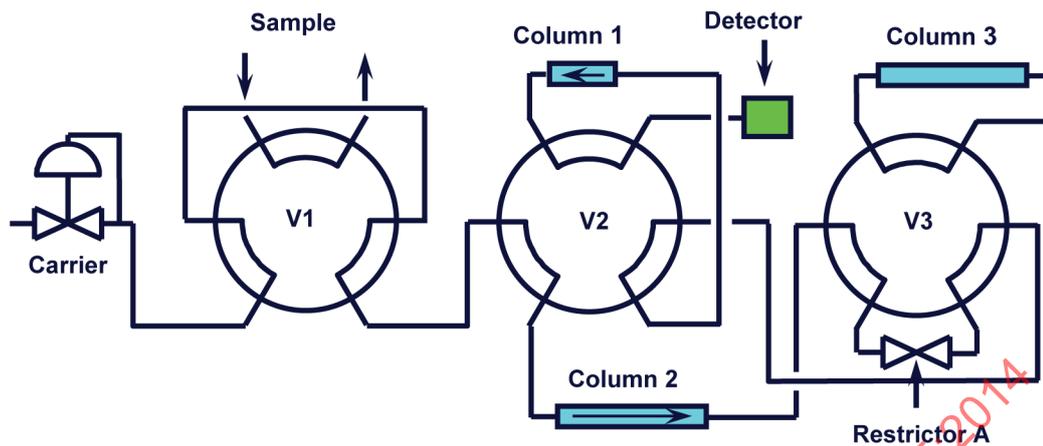
a) Initial configuration: all valves in position 1



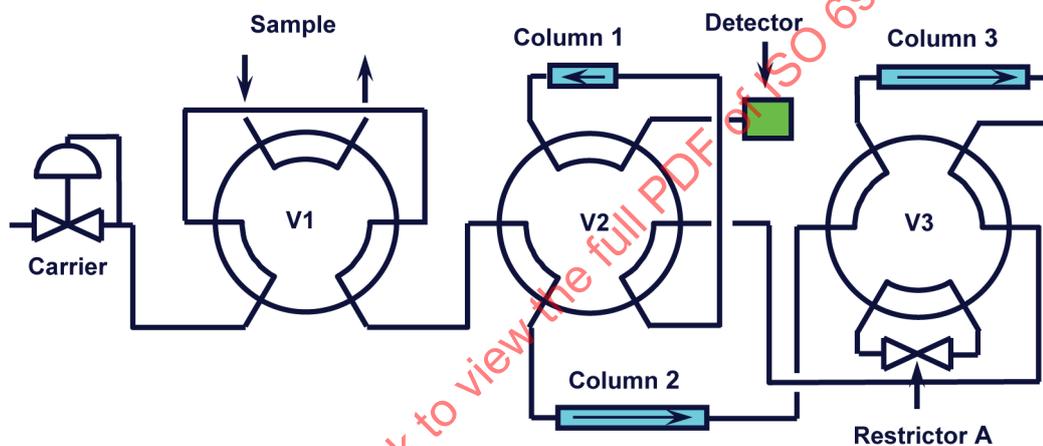
b) Sample injection: V1 to position 2



c) Backflush C<sub>6</sub>+: V2 to position 2



d) Isolate  $N_2$ ,  $C_1$ ,  $CO_2$ ,  $C_2$ ; measure  $C_3$  to  $C_5$ : V3 to position 2



e) Reconnect column 3 - measure  $N_2$ ,  $C_1$ ,  $CO_2$ ,  $C_2$ : V3 to position 1

Figure 2 — Scheme of the configuration

## 7 Procedure

### 7.1 Control of the apparatus

Set up the gas chromatograph according to the manufacturer's instructions.

#### 7.1.1 Column Conditioning

The columns described in 5.8 and 5.9 do not need conditioning or activation, and are generally being used well within their temperature limits. However a small amount of column bleed due to lower-boiling impurities may be evident on first use, and result in unstable baselines. Operation of the analyser overnight with carrier gas flowing but no sample injections, at a temperature 20 °C to 40 °C above the recommended operating temperature should eliminate this effect.

Residual adsorbed moisture in the lines supplying carrier gas or sample gas can give rise to unexplained peaks over and above those expected. Operation overnight under the recommended conditions with sample injection should eliminate these effects.

## 7.2 Operation of the apparatus

### 7.2.1 Analytical method

Examples of the operating conditions for configurations 1 (5.9.1) and 2 (5.9.2) are given in [Tables 2](#) and [3](#).

**Table 2 — Example of instrument conditions, configuration 1**

	Column 1	Column 2	Column 3
Stationary phase	Silicone oil DC 200/500	Silicone oil DC 200/500	Silicone oil DC 200/500
Loading %	28 %	28 %	15 %
Support	Chromosorb P-AW	Chromosorb P-AW	
Active solid			Porapak N
ASTM mesh size	45/60	45/60	50/80
Column length	0,75 m	5,2 m	2,4 m
Column i.d.	2 mm	2 mm	2 mm
material	Stainless steel		
Temperature	100 °C		
Carrier gas	Helium		
Supply pressure	4 bar		
Flowrate	28 ml/min		
Detector	Thermal Conductivity		
Detector Temp.	100 °C minimum		
Injection device	Valve		
Injector Temp.	100 °C		
Sample size	1,0 ml		

Table 3 — Example of instrument conditions, configuration 2

	Column 1	Column 2	Column 3
Stationary phase	Oxy-dipropionitrile	Silicone oil SF-96	Silicone oil DC 200/500
Loading %	28 %	20 %	15 %
Support		Chromosorb P-AW	
Active solid	Porasil C		HayeSep N
ASTM mesh size	80/100	80/100	80/100
Column length	0,3 m	2,1 m	2,1 m
Column i.d.	0,75 mm	0,75 mm	0,75 mm
material	Stainless steel		
Temperature	80 °C		
Carrier gas	Helium		
Supply pressure	4 bar		
Flowrate	28 ml/min		
Detector	Thermal Conductivity		
Detector Temp.	80 °C minimum		
Injection device	Valve		
Injector Temp.	80 °C		
Sample size	0,25 ml		

### 7.2.2 Sample introduction

Purge the sample valve with the gas to be analysed, using at least 20 times the volume of the valve and associated pipe work.

Stop the purge to enable the gas to reach the temperature of the valve and ambient pressure, and then start the analytical cycle, injecting the sample and switching the valves as required.

If this volume of sample is not enough to purge the valve, then contamination by air or by the previous sample will be evident. If either occurs, then use a larger volume of sample for purging.

NOTE The sample loop should be purged with gas for a precise time, at a defined rate, and the sample should then be allowed to equilibrate to ambient pressure before injection. In the absence of equipment which can confirm the latter, there should be a defined time between sample valve shut off and injection.

### 7.2.3 Analysis

The analytical system shown in [Figure 2](#) consists of one six-port sample injection valve, V1, one six-port backflush valve, V2, and one six-port by-pass valve V3. Restrictor A maintains the pneumatic balance of the system when column 3 is isolated. The detailed setting-up procedure is given in [Annex B](#). (One 10-port valve may be used in place of the six-port valves V1 and V2, controlling both sample injection and backflushing of column 1.).

The timings of the valve switching operations must ensure that:

- V2 is returned to the backflush position (position 2) after all the n-pentane leaves column 1 but before the lowest C<sub>6</sub> isomer leaves column 1 on its way to column 2.
- V3 is switched to isolate column 3 (position 2) before any propane leaves column 2 (on its way to column 3) and after all the ethane has left column 2 and entered column 3.

- c) V3 is not returned to reconnect column 3 (position 1) until all the n-pentane has been detected, having emerged from column 2 via column 1.

A typical chromatogram is shown in [Figure 3](#).

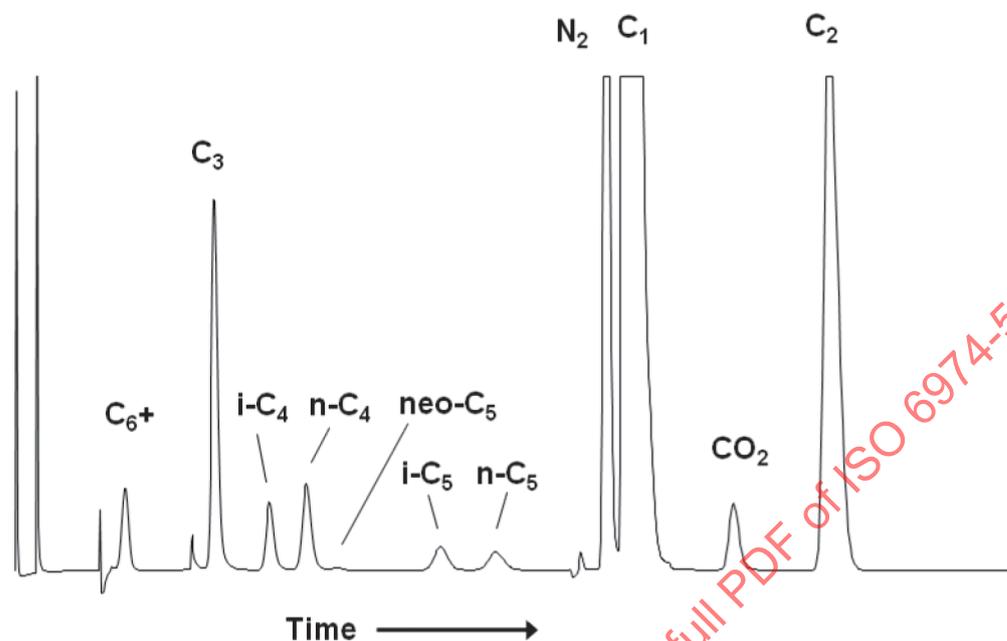


Figure 3 — A typical chromatogram

#### 7.2.4 Peak resolution

It is important that all components are measured without interference from others. The resolution between neighbouring peaks can be assessed according to ISO 7504:2001, 3.3.4.4. Although the resolutions of all peaks are important, there are particular pairs of peaks which are critical: their satisfactory resolution ensures that of other pairs (see [Table 4](#)).

The resolution required is likely to vary with the component uncertainties which are deemed to be acceptable for particular applications. Two values are quoted below - medium resolution, which should be available if the procedure is implemented normally, and high resolution, which may require modifications to column sizes, temperature and flow rate, and is likely to involve a longer analysis time.

NOTE A resolution of 1.5 or higher indicates baseline separation between symmetrical peaks. A resolution of 1.0 is taken to be the minimum value for quantitative measurement.

Table 4 — Peak resolution

Component 1	Component 2	Medium Resolution	High Resolution
i-Butane	n-Butane	1,5	2,0
Nitrogen	Methane	1,25	2,0
Carbon dioxide	Ethane	2,0	2,3

#### 7.2.5 Calibration

Calibrate the equipment according to the procedures in ISO 6974-1.

## **8 Expression of results**

Refer to ISO 6974-1.

### **8.1 Uncertainty**

Refer to ISO 6974-2.

### **8.2 Test report**

Refer to ISO 6974-1.

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## Annex A (informative)

### Example of application

#### A.1 General considerations

In this example, the analysis is considered to be of Type 2. Instrument response for all components is assumed to be first order with zero intercept. All components are measured directly against the same component in the Working Measurement Standard (WMS). No Other Components were determined. No pressure correction was employed, either during calibration or analysis of sample. Multiple operation methods (with or without bridging) were not employed.

Performance evaluation of the instrument according to ISO 10723 was carried out prior to calibration and analysis using seven test gases, each containing 11 components. From prior knowledge of the intended application and likely composition of gases to be presented to the analyser, the Working Range of the instrument is decided to be that given in [Table A.1](#).

**Table A.1 — Working range of the analyser**

	minimum/mol%	maximum/mol%
nitrogen	0,20	12,20
carbon dioxide	0,05	8,00
methane	63,50	98,50
ethane	0,10	14,20
propane	0,05	7,80
i-butane	0,01	1,20
n-butane	0,01	1,20
neo-pentane	0,01	0,34
i-pentane	0,01	0,34
n-pentane	0,01	0,34
C <sub>6</sub> +	0,05	0,34

Performance evaluation resulted in the outputs shown in [Table A.2](#). The mean errors,  $\bar{\delta}_1$ , shown in the second column of [Table A.2](#), are deemed to be sufficiently close to zero that correction is unnecessary (see ISO 6974-1:2012, 6.9.4).

Table A.2 — Output from performance evaluation of the analyser

	$\bar{\delta}_l$ /mol%	$u^2(\bar{\delta}_l)$ /mol% <sup>2</sup>	$\overline{u^2(\delta_l)}$ /mol% <sup>2</sup>
nitrogen	-0,026 937 130	0,000 309 597	0,002 748 444
carbon dioxide	-0,011 354 153	0,000 186 463	0,000 916 820
methane	0,085 980 308	0,000 945 917	0,028 341 745
ethane	-0,030 749 812	0,000 600 030	0,004 667 961
propane	-0,014 790 556	0,000 180 682	0,001 797 461
i-butane	-0,003 493 228	0,000 009 708	0,000 049 207
n-butane	0,001 375 116	0,000 009 970	0,000 021 061
neo-pentane	-0,000 373 260	0,000 000 099	0,000 000 146
i-pentane	0,000 278 938	0,000 001 441	0,000 004 245
n-pentane	0,000 395 722	0,000 000 985	0,000 001 035
C <sub>6</sub> +	-0,000 332 000	0,000 000 756	0,000 002 234

In the example given the mole fractions and their (standard) uncertainties are expressed to a large number of significant figures purely to aid checking of calculations and software. Reporting of results should follow the guidelines indicated in [Clause A.6](#).

## A.2 Calibration

Calibration was performed using 10 injections of the WMS and resulted in the responses listed in [Table A.3](#).



For each component the assumed analysis function was calculated from the mean of the 10 responses of the instrument to that component according to Formula (6) of ISO 6974-1. The coefficients of the calibration function  $b_{1,j}$  and their uncertainties are shown in [Table A.4](#).

**Table A.4 — Mean responses, coefficients of the assumed analysis function and their uncertainties**

	$\overline{y_{i,c}}$	$(\overline{y_{i,c}})$	$b_{1,i}$	$u(b_{1,i})$	$u(\overline{b_{1,i}})$
N <sub>2</sub>	45 389 461	14 233	9,894 4E-08	1,724 6E-10	5,453 6E-11
CO <sub>2</sub>	40 542 542	14 490	8,191 4E-08	1,581 3E-10	5,000 4E-11
CH <sub>4</sub>	668 202 532	165 777	1,204 5E-07	6,030 4E-11	1,907 0E-11
C <sub>2</sub> H <sub>6</sub>	95 516 238	33 698	7,298 2E-08	1,180 1E-10	3,731 7E-11
C <sub>3</sub> H <sub>8</sub>	57 701 475	17 268	5,694 8E-08	1,896 7E-10	5,997 9E-11
i-C <sub>4</sub> H <sub>10</sub>	10 032 792	4 385	4,971 7E-08	2,005 3E-10	6,341 2E-11
n- C <sub>4</sub> H <sub>10</sub>	10 701 339	8 544	4,736 8E-08	1,934 3E-10	6,116 8E-11
neo-C <sub>5</sub> H <sub>12</sub>	2 458 844	5 191	4,485 8E-08	3,899 0E-10	1,233 0E-10
i-C <sub>5</sub> H <sub>12</sub>	2 592 878	1 861	4,308 0E-08	1,762 8E-10	5,574 6E-11
n-C <sub>5</sub> H <sub>12</sub>	2 667 420	1 837	4,131 3E-08	1,673 9E-10	5,293 3E-11
n-C <sub>6</sub> H <sub>14</sub>	2 889 713	919	3,779 4E-08	1,907 1E-10	6,030 8E-11

### A.3 Calculation of mole fractions

#### A.3.1 Mean normalization method (see ISO 6974-1:2012, 6.9.2)

Analysis was performed using 10 injections of unknown sample and the responses are shown in [Table A.5](#).



The mean responses, calculated according to ISO 6974-1:2012, Formula (7) are shown in [Table A.6](#).

**Table A.6 — Analysis of the unknown sample — Mean responses and uncertainties**

Component	$\bar{y}_i$	$u(\bar{y}_i)$	$u(y_{i,l})$
N <sub>2</sub>	10 232 949	3 425	10 830
CO <sub>2</sub>	18 408 482	6 919	21 881
CH <sub>4</sub>	757 934 636	226 900	717 522
C <sub>2</sub> H <sub>6</sub>	34 388 732	12 816	40 527
C <sub>3</sub> H <sub>8</sub>	26 398 346	8 572	27 106
i-C <sub>4</sub> H <sub>10</sub>	3 010 900	1 404	4 441
n- C <sub>4</sub> H <sub>10</sub>	1 092 217	1 910	6 040
neo-C <sub>5</sub> H <sub>12</sub>	2 238 304	2 087	6 600
i-C <sub>5</sub> H <sub>12</sub>	6 506 026	3 061	9 678
n-C <sub>5</sub> H <sub>12</sub>	6 817 882	3 746	11 844
C <sub>6+</sub>	7 502 615	2 606	8 240

The raw mole fractions were calculated according to ISO 6974-1:2012, Formula (9) and are shown in [Table A.7](#).

The mole fractions were calculated according to ISO 6974-1:2012, Formula (11) (note that  $x_{oc}$  is zero in this instance) and are shown in [Table A.7](#).

**Table A.7 — Raw mole fractions, mole fractions and their uncertainties**

Component	$x_i^*$	$u(x_i^*)$	$x_i$	$u(x_i)$
N <sub>2</sub>	1,012	0,019 463	1,023	0,019 479
CO <sub>2</sub>	1,508	0,011 153	1,524	0,011 160
CH <sub>4</sub>	91,291	0,067 999	92,239	0,034 802
C <sub>2</sub> H <sub>6</sub>	2,510	0,024 978	2,536	0,024 668
C <sub>3</sub> H <sub>8</sub>	1,503	0,014 914	1,519	0,014 886
i-C <sub>4</sub> H <sub>10</sub>	0,150	0,002 675	0,151	0,002 701
n- C <sub>4</sub> H <sub>10</sub>	0,052	0,001 818	0,052	0,001 836
neo-C <sub>5</sub> H <sub>12</sub>	0,100	0,000 351	0,101	0,000 363
i-C <sub>5</sub> H <sub>12</sub>	0,280	0,000 852	0,283	0,000 886
n-C <sub>5</sub> H <sub>12</sub>	0,282	0,000 610	0,285	0,000 653
C <sub>6+</sub>	0,284	0,000 724	0,286	0,000 763
total	98,972		100,000	

### A.3.2 Run-by-run normalization method (see ISO 6974-1:2012, 6.9.3)

For each injection of unknown the raw mole fractions were calculated according to ISO 6974-1:2012, Formula (13) and are shown in [Table A.8](#).

For each injection the mole fractions were calculated according to ISO 6974-1:2012, Formula (15) of . Note that  $x_{oc}$  is zero in this instance and are shown in [Table A.9](#).

Table A.8 — Raw mole fractions and their uncertainties for each run

Component	Run 1		Run 2		Run 3		Run 4		Run 5	
	$x_{i,l}^*$	$u(x_{i,l}^*)$								
N <sub>2</sub>	1,013	0,061 5	1,013	0,061 5	1,014	0,061 5	1,014	0,061 5	1,013	0,061 5
CO <sub>2</sub>	1,508	0,035 2	1,509	0,035 2	1,510	0,035 2	1,510	0,035 2	1,509	0,035 2
CH <sub>4</sub>	91,359	0,212 6	91,348	0,212 6	91,419	0,212 6	91,386	0,212 6	91,319	0,212 6
C <sub>2</sub> H <sub>6</sub>	2,511	0,079 0	2,512	0,079 0	2,513	0,079 0	2,513	0,079 0	2,512	0,079 0
C <sub>3</sub> H <sub>8</sub>	1,504	0,047 1	1,504	0,047 1	1,505	0,047 1	1,505	0,047 1	1,504	0,047 1
i-C <sub>4</sub> H <sub>10</sub>	0,150	0,008 4	0,150	0,008 4	0,150	0,008 4	0,150	0,008 4	0,150	0,008 4
n-C <sub>4</sub> H <sub>10</sub>	0,052	0,005 7	0,052	0,005 7	0,052	0,005 7	0,051	0,005 7	0,052	0,005 7
neo-C <sub>5</sub> H <sub>12</sub>	0,100	0,001 1	0,101	0,001 1	0,101	0,001 1	0,100	0,001 1	0,100	0,001 1
i-C <sub>5</sub> H <sub>12</sub>	0,280	0,002 7	0,280	0,002 7	0,280	0,002 7	0,280	0,002 7	0,280	0,002 7
n-C <sub>5</sub> H <sub>12</sub>	0,282	0,001 9	0,282	0,001 9	0,282	0,001 9	0,282	0,001 9	0,282	0,001 9
C <sub>6+</sub>	0,284	0,002 1	0,284	0,002 1	0,284	0,002 1	0,284	0,002 1	0,284	0,002 1
total	99,042		99,035		99,110		99,075		99,006	
Component	Run 6		Run 7		Run 8		Run 9		Run 10	
	$x_{i,l}^*$	$u(x_{i,l}^*)$								
N <sub>2</sub>	1,012	0,061 5	1,012	0,061 5	1,011	0,061 5	1,011	0,061 5	1,011	0,061 5
CO <sub>2</sub>	1,508	0,035 2	1,507	0,035 2	1,506	0,035 2	1,506	0,035 2	1,505	0,035 2
CH <sub>4</sub>	91,255	0,212 5	91,242	0,212 5	91,221	0,212 5	91,192	0,212 5	91,169	0,212 5
C <sub>2</sub> H <sub>6</sub>	2,510	0,079 0	2,509	0,079 0	2,506	0,079 0	2,506	0,079 0	2,505	0,079 0
C <sub>3</sub> H <sub>8</sub>	1,504	0,047 1	1,503	0,047 1	1,502	0,047 1	1,501	0,047 1	1,501	0,047 1
i-C <sub>4</sub> H <sub>10</sub>	0,150	0,008 4	0,150	0,008 4	0,149	0,008 4	0,149	0,008 4	0,150	0,008 4
n-C <sub>4</sub> H <sub>10</sub>	0,052	0,005 7	0,052	0,005 7	0,051	0,005 7	0,052	0,005 7	0,052	0,005 7
neo-C <sub>5</sub> H <sub>12</sub>	0,101	0,001 1	0,100	0,001 1	0,100	0,001 1	0,101	0,001 1	0,101	0,001 1
i-C <sub>5</sub> H <sub>12</sub>	0,280	0,002 7	0,280	0,002 7	0,281	0,002 7	0,280	0,002 7	0,280	0,002 7
n-C <sub>5</sub> H <sub>12</sub>	0,282	0,001 9	0,281	0,001 9	0,282	0,001 9	0,281	0,001 9	0,281	0,001 9
C <sub>6+</sub>	0,284	0,002 1	0,284	0,002 1	0,283	0,002 1	0,283	0,002 1	0,283	0,002 1
total	98,937		98,920		98,893		98,862		98,838	

Table A.9 — Mole fractions and their uncertainties for each run

Component	Run 1		Run 2		Run 3		Run 4		Run 5	
	$x_{i,l}$	$u(x_{i,l})$								
N <sub>2</sub>	1,023	0,061 6	1,023	0,061 6	1,023	0,061 5	1,023	0,061 5	1,023	0,061 6
CO <sub>2</sub>	1,523	0,035 3	1,524	0,035 3	1,524	0,035 2	1,524	0,035 3	1,524	0,035 3
CH <sub>4</sub>	92,242	0,110 0	92,238	0,110 0	92,239	0,109 9	92,239	0,109 9	92,236	0,110 0
C <sub>2</sub> H <sub>6</sub>	2,535	0,078 0	2,537	0,078 0	2,536	0,077 9	2,536	0,077 9	2,537	0,078 0
C <sub>3</sub> H <sub>8</sub>	1,519	0,047 0	1,519	0,047 0	1,519	0,047 0	1,519	0,047 0	1,519	0,047 1
i-C <sub>4</sub> H <sub>10</sub>	0,151	0,008 5	0,151	0,008 5	0,151	0,008 5	0,151	0,008 5	0,151	0,008 5
n-C <sub>4</sub> H <sub>10</sub>	0,052	0,005 8	0,052	0,005 8	0,053	0,005 8	0,052	0,005 8	0,052	0,005 8
neo-C <sub>5</sub> H <sub>12</sub>	0,101	0,001 1	0,102	0,001 1	0,102	0,001 1	0,101	0,001 1	0,101	0,001 1
i-C <sub>5</sub> H <sub>12</sub>	0,283	0,002 8	0,283	0,002 8	0,283	0,002 8	0,283	0,002 8	0,283	0,002 8
n-C <sub>5</sub> H <sub>12</sub>	0,284	0,002 1	0,285	0,002 1	0,284	0,002 1	0,285	0,002 1	0,285	0,002 1
C <sub>6+</sub>	0,286	0,002 4	0,287	0,002 4	0,286	0,002 4	0,286	0,002 4	0,287	0,002 4
total	100,000		100,000		100,000		100,000		100,000	
Component	Run 6		Run 7		Run 8		Run 9		Run 10	
	$x_{i,l}$	$u(x_{i,l})$								
N <sub>2</sub>	1,023	0,061 6	1,023	0,061 6	1,023	0,061 6	1,023	0,061 7	1,023	0,061 7
CO <sub>2</sub>	1,524	0,035 3	1,524	0,035 3	1,523	0,035 3	1,523	0,035 3	1,523	0,035 3
CH <sub>4</sub>	92,236	0,110 1	92,238	0,110 1	92,242	0,110 1	92,242	0,110 2	92,241	0,110 2
C <sub>2</sub> H <sub>6</sub>	2,537	0,078 0	2,537	0,078 0	2,534	0,078 1	2,535	0,078 1	2,535	0,078 1
C <sub>3</sub> H <sub>8</sub>	1,520	0,047 1	1,519	0,047 1	1,519	0,047 1	1,518	0,047 1	1,519	0,047 1
i-C <sub>4</sub> H <sub>10</sub>	0,151	0,008 5	0,151	0,008 5	0,151	0,008 5	0,151	0,008 6	0,151	0,008 6
n-C <sub>4</sub> H <sub>10</sub>	0,053	0,005 8	0,052	0,005 8	0,052	0,005 8	0,053	0,005 8	0,053	0,005 8
neo-C <sub>5</sub> H <sub>12</sub>	0,102	0,001 2	0,101	0,001 1	0,101	0,001 1	0,102	0,001 2	0,102	0,001 2
i-C <sub>5</sub> H <sub>12</sub>	0,283	0,002 8	0,283	0,002 8	0,284	0,002 8	0,283	0,002 8	0,283	0,002 8
n-C <sub>5</sub> H <sub>12</sub>	0,285	0,002 1	0,284	0,002 1	0,285	0,002 1	0,284	0,002 1	0,285	0,002 1
C <sub>6+</sub>	0,287	0,002 4	0,287	0,002 4	0,286	0,002 4	0,286	0,002 4	0,286	0,002 4
total	100,000		100,000		100,000		100,000		100,000	