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**Animal feeding stuffs — Determination of  
phytase activity**

*Aliments des animaux — Détermination de l'activité phytasique*

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# Contents

Page

Foreword.....	iv
Introduction .....	v
1 Scope .....	1
2 Terms and definitions.....	1
3 Principle.....	1
4 Reagents .....	2
5 Apparatus .....	3
6 Sampling.....	4
7 Sample preparation .....	4
8 Procedure .....	4
8.1 Blank solutions .....	4
8.2 Standards .....	4
8.3 Standard curve.....	5
8.4 Phytase level control.....	5
8.5 Feed test portions.....	6
9 Calculations.....	7
9.1 Formulation standard curve .....	7
9.2 Calculation phytase activity .....	8
9.3 Correction for phytic acid purity and water content .....	9
9.4 Interference with blank values .....	9
10 Precision.....	9
10.1 Limit of detection and limit of quantification.....	9
10.2 Interlaboratory test .....	10
10.3 Repeatability.....	10
10.4 Reproducibility.....	10
11 Test report .....	10
Annex A (informative) Interlaboratory study results .....	11
Bibliography .....	12

## Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

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The main task of technical committees is to prepare International Standards. Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

ISO 30024 was prepared by the European Committee for Standardization (CEN) Technical Committee CEN/TC 327, *Animal feeding stuffs — Methods of sampling and analysis*, in collaboration with Technical Committee ISO/TC 34, *Food products*, Subcommittee SC 10, *Animal feeding stuffs*, in accordance with the Agreement on technical cooperation between ISO and CEN (Vienna Agreement).

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## Introduction

This International Standard has been developed to quantify phytase products in feed samples to enable the European Commission to control the phytase content of animal feed products. However, the method cannot be used to evaluate the *in vivo* efficacy of the phytase products.

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# Animal feeding stuffs — Determination of phytase activity

## 1 Scope

This International Standard specifies the determination of phytase activity in feed samples.

The method does not distinguish between phytase added as a feed additive and endogenous phytase already present in the feed materials.

The method cannot be used to evaluate or compare the *in vivo* efficacy of the phytase product. It is not a predictive method of the *in vivo* efficacy of phytases present on the market as they can develop different *in vivo* efficacy per unit of activity.

The method is suitable and validated exclusively for the determination of phytase activity and exclusively in complete feeds.

**NOTE** The harmonized method was developed on the basis of the presently existing phytase products [E1600 (EC 3.1.3.8, 3-phytase), E1614 (EC 3.1.3.26, 4-phytase), and E1640 (EC 3.1.3.26, 4-phytase)]. Therefore, it might not necessarily be suitable as such for phytase products that are developed in the future. The harmonized method is thus a tool which is useful only to evaluate the total phytase activity in feed samples.

## 2 Terms and definitions

For the purposes of this document, the following terms and definitions apply.

### 2.1

#### phytase unit

U

amount of enzyme that releases 1  $\mu\text{mol}$  of inorganic phosphate from phytate per minute under the reaction conditions specified in this International Standard

## 3 Principle

Phytase releases phosphate from the substrate *myo*-inositol hexakisphosphate (phytate). The released inorganic phosphate is determined by forming a yellow complex with an acidic molybdate/vanadate reagent. The optical density (OD) of the yellow complex is measured at a wavelength of 415 nm and the inorganic phosphate released is quantified from a phosphate standard calibration curve.

## 4 Reagents

During the analysis, unless otherwise stated, use only reagents of recognized analytical grade and distilled or demineralized water or water of equivalent purity.

**WARNING — This method requires the handling of hazardous substances. Observe local regulations for potentially hazardous chemicals to minimize risks to organizational, technical, and personal safety.**

- 4.1 Ammonia solution**, 25 % mass fraction;  $\text{NH}_3$ .
- 4.2 Ammonium heptamolybdate tetrahydrate**,  $(\text{NH}_4)_6\text{Mo}_7\text{O}_{24}\cdot 4\text{H}_2\text{O}$ .
- 4.3 Ammonium monovanadate**,  $\text{NH}_4\text{VO}_3$ .
- 4.4 Hydrochloric acid**, 25 % mass fraction;  $\text{HCl}$ .
- 4.5 Nitric acid**, 65 % mass fraction;  $\text{HNO}_3$ .
- 4.6 Potassium dihydrogenphosphate**,  $\text{KH}_2\text{PO}_4$ .
- 4.7 Phytate**, phytic acid, dodecasodium salt,  $\text{C}_6\text{H}_6\text{Na}_{12}\text{O}_{24}\text{P}_6\cdot x\text{H}_2\text{O}$ , from rice, Sigma® P0109<sup>1)</sup>.
- 4.8 Sodium acetate trihydrate**,  $\text{CH}_3\text{COONa}\cdot 3\text{H}_2\text{O}$ .
- 4.9 Polysorbate 20**<sup>2)</sup>.
- 4.10 Dilute nitric acid**. Dilute 1 volume nitric acid 65 % mass fraction (4.5) with 2 volumes water. Store at room temperature. The maximum storage time is indefinite.
- 4.11 Ammonium heptamolybdate reagent**. Dissolve 100,0 g ammonium heptamolybdate tetrahydrate (4.2) in approximately 800 ml water. Add 10 ml 25 % mass fraction ammonia solution (4.1) and make up with water to 1 000 ml. Store at room temperature in the dark. The maximum storage time is 2 months.
- 4.12 Ammonium vanadate reagent**. Dissolve completely 2,35 g of ammonium monovanadate (4.3) in approximately 400 ml water (50 °C to 60 °C). Add 20 ml dilute nitric acid (4.10) and make up with water to 1 000 ml. Store at room temperature in the dark. The maximum storage time is 2 months.
- 4.13 Molybdate/vanadate STOP reagent**. Mix 1 volume ammonium vanadate reagent (4.12) with 1 volume ammonium heptamolybdate reagent (4.11) and add 2 volumes dilute nitric acid (4.10). Mix and store at room temperature. The maximum storage time is 1 day.
- 4.14 Polysorbate 20, 10 % mass fraction**. Dissolve 10,0 g of polysorbate 20 (4.9) with water and make up to 100 ml. Store at room temperature. The maximum storage time is 6 months.
- 4.15 Acetate buffer**, pH 5,5; 0,25 mol/l. Dissolve 34,0 g of sodium acetate trihydrate (4.8) in approximately 900 ml water. Adjust the pH with 25 % mass fraction hydrochloric acid (4.4) to  $5,50 \pm 0,02$  and make up to 1 000 ml with water. Store at room temperature. The maximum storage time is 2 weeks.
- 4.16 Acetate buffer with 0,01 % mass fraction polysorbate 20**, pH 5,5; 0,25 mol/l. Dissolve 34,0 g of sodium acetate trihydrate (4.8) in approximately 900 ml water. Adjust the pH with 25 % mass fraction hydrochloric acid (4.4) to  $5,50 \pm 0,02$ . Add 1 ml 10 % mass fraction polysorbate 20 (4.14) and make up to 1 000 ml with water. Store at room temperature. The maximum storage time is 2 weeks.

1) Example of a suitable product available commercially. This information is given for the convenience of users of this International Standard and does not constitute an endorsement by ISO of this product.

2) Tween 20 is an example of a suitable product available commercially. This information is given for the convenience of users of this International Standard and does not constitute an endorsement by ISO of this product.

**4.17 Acetate buffer with 0,01 % mass fraction polysorbate 20**, pH 5,5; 0,50 mol/l. Dissolve 68,0 g of sodium acetate trihydrate (4.8) in approximately 900 ml water. Adjust the pH with 25 % mass fraction hydrochloric acid (4.4) to  $5,50 \pm 0,02$ . Add 1 ml 10 % mass fraction polysorbate 20 (4.14) and make up to 1 000 ml with water. Store at room temperature. The maximum storage time is 2 weeks.

**4.18 Phytate substrate solution**, 7,5 mmol/l (5 mmol/l end-concentration in the reaction). Dissolve 2,00 g of dodecasodium phytate (4.7) whose inorganic phosphorus content is  $\leq 0,1$  % mass fraction (see 9.3) in approximately 200 ml acetate buffer (4.15). Adjust the pH with 25 % mass fraction hydrochloric acid (4.4) to  $5,50 \pm 0,02$  and make up with acetate buffer (4.15) to 250 ml. The maximum storage time is 2 weeks at 4 °C.

**4.19 Phosphate stock standard solution**, 50 mmol/l. Dry approximately 10 g of potassium dihydrogenphosphate (4.6) at 105 °C for 2 h and store it in a dessicator. Weigh approximately 682 mg of dried potassium dihydrogenphosphate, transfer it quantitatively to a 100 ml volumetric flask and make up to 100 ml with 0,25 mol/l acetate buffer with 0,01 % mass fraction polysorbate 20 (4.16). Calculate the exact concentration of the phosphate stock standard solution. Store at room temperature. The maximum storage time is 2 weeks.

**4.20 Phytase stock standard solution**. Weigh 100,0 mg to 300,0 mg of a certified phytase standard, transfer it quantitatively to a 100 ml volumetric flask and dissolve it in 100 ml 0,25 mol/l acetate buffer with 0,01 % mass fraction polysorbate 20 (4.16). Stir it for 15 min to 45 min. Store at room temperature. The maximum storage time is 1 day.

## 5 Apparatus

Usual laboratory apparatus, in particular, the following.

- 5.1 **Water bath**, thermostatically controlled (with inserts for 2 ml tubes).
- 5.2 **pH-meter**, capable of being read to at least two places of decimals.
- 5.3 **Magnetic stirrers** ( $\geq 20$  W power).
- 5.4 **Egg-shaped stirring bars** (40 mm  $\times$  20 mm).
- 5.5 **Analytical balance**, capable of being read to at least 0,1 mg.
- 5.6 **Balance**, capable of being read to at least 0,01 g.
- 5.7 **Vortex mixer**.
- 5.8 **Centrifuge** for microcentrifuge tubes (5.12), capable of 11 000g to 20 000g.
- 5.9 **Electronic dispenser**.
- 5.10 **Pipettes** (electronic and manual), in the range 10  $\mu$ l to 2 000  $\mu$ l.
- 5.11 **Spectrophotometer** double beam or microplate reader.
- 5.12 **Microcentrifuge tubes**, capacity 2 ml.

## 6 Sampling

A representative sample should have been sent to the laboratory. It should not have been damaged or changed during transport or storage.

Sampling is not part of the method specified in this International Standard. A recommended sampling procedure is given in ISO 6497 [1].

## 7 Sample preparation

Perform two weighings for each sample.

Weigh two portions of pellets or mash, of about 50 g each, into 500 ml conical flasks. Add 500 ml water and 0,5 ml of 10 % mass fraction polysorbate 20 (4.14) to the feed and mix vigorously for 45 min on a magnetic stirrer (5.3) with egg-shaped stirring bars (5.4). Transfer 2 ml of the feed extract to a microcentrifuge tube (5.12) and centrifuge (5.8) for 3 min at 11 000g to 20 000g.

Inhomogeneity in the sample can lead to high coefficients of variation (CVs). For feed samples showing CVs > 15 %, such inhomogeneity can derive from inhomogeneous particle size distribution in products or inhomogeneous feed preparation. If feed samples show high CVs, grind the feed samples using an Ultra centrifugal mill<sup>3)</sup> with a sieve of nominal size of openings 1 mm. Grind 150 g feed and extract as described in this clause.

## 8 Procedure

### 8.1 Blank solutions

Inorganic phosphate in the sample contributes to colour formation. Therefore, blanks are included for each sample. For calculation of phytase activity, subtract blank values from the test values.

### 8.2 Standards

#### 8.2.1 Phosphate standard solutions

The phosphate stock standard solution (4.19) is diluted with 0,25 mol/l acetate buffer containing 0,01 % mass fraction polysorbate 20 (4.16) according to Table 1.

**Table 1 — Dilution steps to obtain standard colorimetric solutions for the phosphate curve**

Standard solution	Volumes phosphate stock standard solution (4.19)	Volumes 0,25 mol/l acetate buffer with 0,01 % mass fraction polysorbate 20 (4.16)	Dilution factor	Concentration $\mu\text{mol/ml}^a$
A	1	1	2	25
B	1	3	4	12,5
C	1	7	8	6,25
D	1	15	16	3,125

<sup>a</sup> Calculate the exact concentrations (4.19).

3) Example of a suitable product available commercially. This information is given for the convenience of users of this International Standard and does not constitute an endorsement by ISO of this product.

### 8.2.2 Phytase level control

For each incubation of samples, a phytase level control is included. A phytase stock standard solution (4.20) with known activity is diluted to a final activity of 0,15 U/ml to 0,25 U/ml and the exact activity is determined as specified in 8.4.

### 8.3 Standard curve

Perform three determinations for each phosphate dilution and two blanks, and average the results. The procedure is specified in Table 2.

For the phosphate standard solutions, pipette 360  $\mu$ l 0,25 mol/l acetate buffer with 0,01 % mass fraction polysorbate 20 (4.16) into a 2 ml tube (5.12). Add 40  $\mu$ l phosphate standard solution (Table 1).

For the phosphate standard blanks, pipette 400  $\mu$ l 0,25 mol/l acetate buffer with 0,01 % mass fraction polysorbate 20 (4.16) into a 2 ml tube (5.12).

In both cases, add 0,8 ml phytate substrate solution (4.18) and 0,8 ml STOP reagent (4.13). Mix the contents of the tubes and maintain them for 10 min at room temperature. Centrifuge (5.8) the tubes and their contents for 3 min at 11 000g to 20 000g and measure the OD of the clear supernatant at 415 nm,  $D(415)$ .

Table 2 — Procedure for standard curve

Assay steps	Standard colorimetric solutions	Blank
Acetate buffer 0,25 mol/l with 0,01 % mass fraction polysorbate 20 (4.16)	360 $\mu$ l	400 $\mu$ l
Phosphate standard solution (8.2.1)	40 $\mu$ l	0 $\mu$ l
Phytate substrate solution (4.18)	0,8 ml	0,8 ml
STOP reagent (4.13)	0,8 ml	0,8 ml
Mix	Yes	Yes
Room temperature	10 min	10 min
Centrifugation	3 min at 11 000g to 20 000g	3 min at 11 000g to 20 000g
Spectrophotometer (5.11)	415 nm (against water)	415 nm (against water)

### 8.4 Phytase level control

Perform three determinations for each dilution and two blanks, and average the results. The procedure is specified in Table 3.

For the phytase level control determination solutions, pipette 360  $\mu$ l 0,25 mol/l acetate buffer with 0,01 % mass fraction polysorbate 20 (4.16) into a 2 ml tube (5.12). Add 40  $\mu$ l dilute phytase level control solution (8.2.2). Mix the sample. Pre-incubate the solutions for 5 min at 37 °C. Add 0,8 ml phytate substrate solution (4.18) preheated to 37 °C. Incubate for exactly 30 min at 37 °C. After 30 min, add 0,8 ml STOP reagent (4.13) and mix. Maintain the solutions for 10 min at room temperature and then centrifuge them for 3 min at 11 000g to 20 000g. Measure the  $D(415)$  of the clear supernatant.

For the phytase level control blank solutions, pipette 360  $\mu$ l acetate buffer (4.15) into a 2 ml tube (5.12). Add 40  $\mu$ l dilute phytase level control solution (8.2.2). The order of addition of solutions differs from that used for the determinations. Pre-incubate blanks for 5 min at 37 °C. Then, as step 1, add STOP reagent (4.13); as step 2, add phytate substrate solution (4.18) preheated to 37 °C. Then proceed with Table 3, column 3, row 9, onwards.

**Table 3 — Procedure for level control**

Assay steps	Level control samples	Blank
Acetate buffer 0,25 mol/l with 0,01 % mass fraction polysorbate 20 (4.16)	360 µl	360 µl
Dilute phytase level control solution (8.2.2)	40 µl	40 µl
Mix	Yes	Yes
Pre-incubation at 37 °C	5 min	5 min
Phytate substrate solution (4.18) at 37 °C	0,8 ml	0,8 ml: Step 2
Mix	No	No
Incubation at 37 °C	30 min	No
STOP reagent (4.13)	0,8 ml	0,8 ml: Step 1
Mix	Yes	Yes
Room temperature	10 min	10 min
Centrifugation	3 min at 11 000g to 20 000g	3 min at 11 000g to 20 000g
Spectrophotometer (5.11)	415 nm (against water)	415 nm (against water)

**8.5 Feed test portions**

Perform three determinations for each extraction (Clause 7) and two blanks, and average the results. The procedure is specified in Table 4.

For the determinations, pipette 300 µl 0,25 mol/l acetate buffer with 0,01 % mass fraction polysorbate 20 (4.16) into a 2 ml tube (5.12). Add 100 µl feed extract (Clause 7) as the test portion. Mix the contents of the tube. Pre-incubate for 5 min at 37 °C. Add 0,8 ml phytate substrate solution (4.18) preheated to 37 °C. Incubate for exactly 30 min at 37 °C. After 30 min, add 0,8 ml STOP reagent (4.13) and mix. Maintain the tube and its contents for 10 min at room temperature and then centrifuge for 3 min at 11 000g to 20 000g. Measure the *D*(415) of the clear supernatant.

For the blanks, pipette 300 µl 0,25 mol/l acetate buffer with 0,01 % mass fraction polysorbate 20 (4.16) into a 2 ml tube (5.12). Add 100 µl feed extract (Clause 7). The order of addition of solutions differs from that used for the test portion. Pre-incubate blanks for 5 min at 37 °C. Then, as step 1, add STOP reagent (4.13); as step 2, add phytate substrate solution (4.18) preheated to 37 °C. Then proceed with Table 4, column 3, row 9, onwards.

**Table 4 — Procedure for feed test portions**

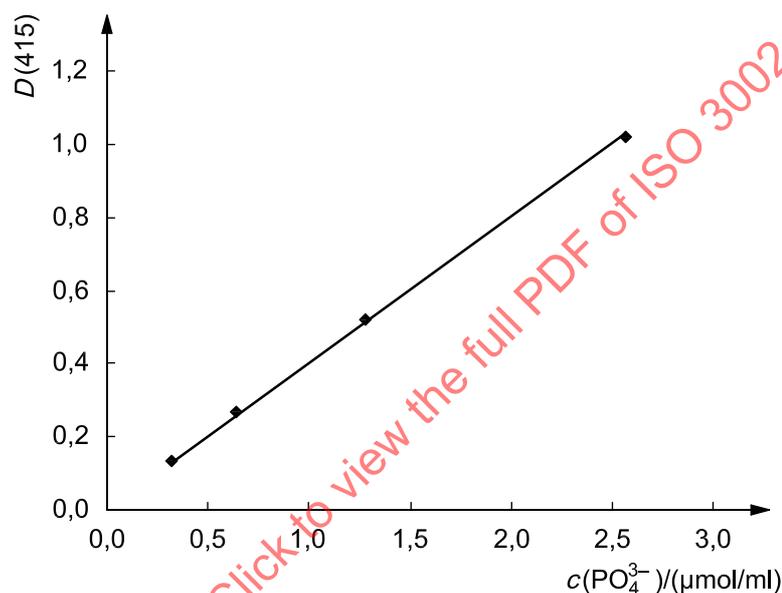
Assay steps	Feed samples	Blank
Acetate buffer 0,25 mol/l with 0,01 % mass fraction polysorbate 20 (4.16)	300 µl	300 µl
Test portion <sup>a</sup>	100 µl	100 µl
Mix	Yes	Yes
Pre-incubation at 37 °C	5 min	5 min
Phytate substrate solution (4.18) at 37 °C	0,8 ml	0,8 ml: Step 2
Mix	No	No
Incubation at 37 °C	30 min	No
STOP reagent (4.13)	0,8 ml	0,8 ml: Step 1
Mix	Yes	Yes
Room temperature	10 min	10 min
Centrifugation	3 min at 11 000g to 20 000g	3 min at 11 000g to 20 000g
Spectrophotometer (5.11)	415 nm (against water)	415 nm (against water)

<sup>a</sup> Test portions with ≤ 200 phytase U/kg feed, 200 µl sample extract and 200 µl 0,50 mol/l acetate buffer with 0,01 % mass fraction polysorbate 20 (4.17) are taken for the assay (1→2 dilution). Samples with > 2 000 phytase U/kg feed have to be appropriately diluted with 0,25 mol/l acetate buffer with 0,01 % mass fraction polysorbate 20 (4.16).

## 9 Calculations

### 9.1 Formulation standard curve

Plot a standard curve with the  $\Delta D(415)$  [ $D(415)_s - D(415)_b$ ], where  $D(415)_s$  and  $D(415)_b$  are the mean standard and blank optical densities, respectively, obtained with the phosphate standards (8.2.1 and 8.3), on the ordinate and the calculated phosphate concentration on the abscissa {the 1→10 dilution in the microcentrifuge tube [40  $\mu$ l diluted standard (Table 1) plus 360  $\mu$ l buffer] for reaction should take the phosphate concentration of the standard into account for the calculation}. The best fitting line is calculated by linear regression,  $\Delta D(415) = kc(\text{PO}_4^{3-})$ , (set intercept at 0). An example is shown in Figure 1.



#### Key

$D(415)$  optical density at 415 nm  $D(415) = 0,400\ 8\ c(\text{PO}_4^{3-})$

$c(\text{PO}_4^{3-})$  concentration of phosphate  $r_{c(\text{PO}_4^{3-}) D(415)}^2 = 0,999\ 6$

**Figure 1 — Example of plot of optical density versus concentration for phosphate standard colorimetric solutions buffered with 0,25 mol/l acetate containing 0,01 % mass fraction polysorbate 20 (4.16)**

## 9.2 Calculation phytase activity

The phytase activity,  $a_p$ , is calculated as follows:

$$a_p = \frac{\Delta D(415) V_d}{k m t} \quad (1)$$

where

$\Delta D(415)$  is the net optical density at 415 nm, calculated from

$$D(415)_t - D(415)_b$$

in which  $D(415)_t$  and  $D(415)_b$  are the mean test portion and blank optical densities (8.5), respectively;

$k$  is the slope of the standard curve, in reciprocal micromole millilitres, at  $D(415)$ ;

$V_d$  is the volume, in millilitres, corrected for dilution (extraction volume times dilution of the extract);

$m$  is the mass, in grams or kilograms, of the sample;

$t$  is the incubation time, in minutes.

### EXAMPLE 1 Phytase level control

$$\Delta D(415) = 0,216$$

$$k = 0,375 \text{ 7 } \mu\text{mol}^{-1} \text{ ml}$$

$$V_d = 30 \text{ 000 ml (100 ml extraction volume } \times 30 \text{ [1} \rightarrow 30 \text{ dilution of phytase stock standard solution (4.20)] } \times 10 \text{ [40 } \mu\text{l diluted phytase stock standard solution (4.20) + 360 } \mu\text{l buffer = 1} \rightarrow 10\text{]})}$$

$$m = 0,107 \text{ 4 g}$$

$$t = 30 \text{ min}$$

$$a_p = \frac{0,216 \times 30 \text{ 000}}{0,375 \text{ 7} \times 0,107 \text{ 4} \times 30} = 5 \text{ 353 } \mu\text{mol min}^{-1} \text{ g}^{-1} = 5 \text{ 353 U g}^{-1} \quad (2)$$

### EXAMPLE 2 Feed test portion

$$\Delta D(415) = 0,183$$

$$k = 0,375 \text{ 7 } \mu\text{mol}^{-1} \text{ ml}$$

$$V_d = 2 \text{ 000 ml (500 ml extraction volume } \times 4 \text{ [100 } \mu\text{l extract + 300 } \mu\text{l buffer = 1} \rightarrow 4\text{]})}$$

$$m = 0,050 \text{ kg}$$

$$t = 30 \text{ min}$$

$$a_p = \frac{0,183 \times 2 \text{ 000}}{0,375 \text{ 7} \times 0,050 \times 30} = 650 \mu\text{mol min}^{-1} \text{ kg}^{-1} = 650 \text{ U kg}^{-1} \quad (3)$$

### 9.3 Correction for phytic acid purity and water content

The purity and water content of phytic acid varies from batch to batch and has therefore to be included in the calculation for the substrate preparation.

EXAMPLE Phytic acid dodecasodium salt (0,008 % mass fraction inorganic phosphorus)<sup>4</sup>).

The correction is given by

$$\frac{c(\text{phytate}) M}{w_p (1 - w_{\text{H}_2\text{O}})} \quad (4)$$

where

$M$  is the molar mass, in grams per mole, of dodecasodium phytate;

$w_p$  is the purity, as a mass fraction, of dodecasodium phytate;

$w_{\text{H}_2\text{O}}$  is the water content, as a mass fraction, of dodecasodium phytate.

If:  $c(\text{phytate}) = 7,5$  mmol/l;  $M_r = 923,8$  g/mol;  $w_p = 97$  % mass fraction; and  $w_{\text{H}_2\text{O}} = 12,6$  % mass fraction, the correction is 8,17 g/l.

### 9.4 Interference with blank values

High levels of, for example, MCP [monocalciumphosphate,  $\text{Ca}(\text{H}_2\text{PO}_4)_2$ ] in feed samples can lead to high blank values [ $D(415)_b > 1,6$ ]. Some spectrophotometers may not be linear in the high OD range (close to 2). Phytase values obtained from feed samples with such high blank values have to be interpreted carefully since the high OD values may lead to an over- or underestimation of the phytase content. A 1→2 or 1→4 dilution of the feed extract may help to reduce the OD value. However,  $\Delta D(415)$  should be  $> 0,04$ .

## 10 Precision

### 10.1 Limit of detection and limit of quantification

According to the IUPAC nomenclature (Reference [2]), the detection limit is defined as  $L_D = 3s$ , and the quantification limit as  $L_Q = 10\sigma$ , where  $s$  and  $\sigma$  are the estimated and absolute standard deviations, respectively.

In terms of  $\Delta D(415)$ , the detection limit is given by

$$L_D = 0,011 \Delta D(415)$$

or 20 U/kg feed; the quantification limit is given by

$$L_Q = 0,036 \Delta D(415)$$

or 60 U/kg feed.

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4) P0109 from Sigma® Lot 057K0049 is an example of a suitable product available commercially. This information is given for the convenience of users of this International Standard and does not constitute an endorsement by ISO of this product.

## 10.2 Interlaboratory test

The values derived from this interlaboratory test may not be applicable to concentration ranges and matrices other than those given.

## 10.3 Repeatability

The coefficient of variation of repeatability,  $CV(r)$ , is the mean coefficient of variation from two independent results obtained from the same sample on the same day, from the same technician, with the same equipment and method.

The estimated coefficient of variation of repeatability,  $\widehat{CV}(r)$ , is 10 %.

## 10.4 Reproducibility

The coefficient of variation of reproducibility,  $CV(R)$ , is the mean coefficient of variation from results obtained with the same sample using the same method, but received from different laboratories, measured on different days, with different equipment and technicians.

The estimated coefficient of variation of reproducibility,  $\widehat{CV}(R)$ , is 12 %.

## 11 Test report

The test report shall contain at least the following information:

- a) all information necessary for the complete identification of the sample;
- b) the sampling method used, if known;
- c) the test method used, with reference to this International Standard;
- d) operating details not specified in this International Standard, or regarded as optional, together with details of any incidents which may have influenced the test result(s);
- e) the test result obtained;
- f) if the repeatability has been checked, the final quoted result obtained.