
**Comparison of toxic gas data from
different tests —**

**Part 1:
Guidance and requirements**

*Recommandations pour la comparaison de données de gaz toxiques
provenant de différents essais —*

Partie 1: Lignes directrices et exigences

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Published in Switzerland

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see www.iso.org/patents).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation of the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT) see www.iso.org/iso/foreword.html.

This document was prepared by Technical Committee ISO/TC 92, *Fire safety*, Subcommittee SC 3, *Fire threat to people and environment*.

This first edition of ISO 29903-1 cancels and replaces the first edition (ISO 29903:2012), which has been technically revised.

The main changes compared to the previous edition are as follows:

- ISO 29903 has been divided into two parts: ISO 29903-1 (this document) and ISO 29903-2.
- Subclause 4.4 has been revised to include requirements on the identity and properties of test specimens.
- Annex C from ISO 29903 (previous edition) (application examples) has been deleted. Application examples will instead be put in a separate document in the ISO 29903 series.

A list of all parts in the ISO 29903 series can be found on the ISO website.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

Introduction

The production of toxic gases in fires can be a significant factor in determining whether people escape from a fire or not. Estimation of the time available for escape and the time required for escape each require values of the concentrations of toxic gases along possible escape paths. Typically, the yields of the gases from burning finished products are estimated or measured prior to conducting such calculations. In some rare cases toxic species production can be calculated during modelling of the fire development. Typically, spread of the gases and their dilution with air is then simulated using equations or computational models.

The yields of these gases can be measured in a real-scale laboratory test of the entire finished product (e.g. a chair) or in a bench-scale test (using a physical fire model) of a specimen cut from the product or a component of the product. Since there are thousands of different combustibles, routine real-scale testing is both costly and impractical. Thus, there is a need to develop reliable methods to use physical fire models, conducted in less than real-scale, for the estimation of real-scale emissions.

The yields of the gases from the real-scale test are often considered to be the accurate values for the particular test conditions. In tests involving a portion of the finished product in a physical fire model, the specimen characteristics and the combustion conditions differ from those in the real-scale test. In most cases the physical fire model reproduces one part of the entire real-scale scenario, e.g. initial well-ventilated conditions or later vitiated conditions. The yields of combustion products in a fire test depend on apparatus conditions such as: the fuel/air equivalence ratio, whether the decomposition is flaming or non-flaming, the persistence of flaming of the sample, the temperature of the specimen and the effluents produced, the stability of the decomposition conditions, and the interaction of the apparatus with the decomposition process, with the effluents and with the flames.

It is, therefore, important to have a standardised methodology for comparing the toxic gas yields generated in tests of different scales to determine the appropriateness of using the data from individual physical fire models in fire hazard and risk assessment. It is also valuable to be able to compare the yield data from different physical fire models to determine whether or when they generate comparable results.

This document concerns the comparison of toxic gas data between small-scale (physical fire models) and large-scale tests and between different small-scale tests, i.e. it covers

- a) the comparison of toxic gas data from fire tests of different physical scales and characteristics in terms of a methodology to identify whether the data are comparable and (provided it is comparable) how to make relevant comparisons, and
- b) the prediction of large-scale results based on small-scale test data or vice versa.

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Comparison of toxic gas data from different tests —

Part 1: Guidance and requirements

1 Scope

This document provides principles for characterizing the measured production of toxic gases from a laboratory fire test and provides bases for comparing the results between different types and scales of such tests. It also includes consideration of the uncertainties in the gas determinations. The combined uncertainty is a key factor in the ability to establish similarity or difference of test results.

The sufficiency of the agreement between a bench-scale test and a real-scale test depends on the precision needed in the fire hazard or risk assessment, which is not covered by this document.

This document defines the relevance and significance of toxic gas data from measurements in different fire tests. With such a definition it is possible to provide generic guidance on how such data can be compared between different sizes and types of fire tests.

The combustion conditions represented by the fire test, other specific characteristics of the test and the test specimen, the sampling strategy of the fire effluents, and the analysis technique for the toxic gas species are the most important factors when defining the significance of the toxic gas data.

This document is intended to serve as a tool for the

- a) definition of the relevance and significance of toxic gas data from fire tests,
- b) comparison of toxic gas data from fire tests of different scales and characteristics, and
- c) prediction of toxic gas data from a large-scale test based on small-scale data or vice versa.

This document gives general guidance regarding comparison of toxic gas data between physical fire models of different scales, but is principally developed for the gases listed in ISO 13571, i.e. carbon dioxide (CO₂), carbon monoxide (CO), hydrogen halides (HCl, HBr, HF), sulfur dioxide (SO₂), hydrogen cyanide (HCN), nitrogen oxides (NO, NO₂), formaldehyde (CH₂O) and acrolein (C₃H₄O).

This document is not applicable to characterization and comparisons of the toxicity of the effluents from fire tests.

2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 5725-1, *Accuracy (trueness and precision) of measurement methods and results — Part 1: General principles and definitions*

ISO 13344, *Estimation of the lethal toxic potency of fire effluents*

ISO 13571, *Life-threatening components of fire — Guidelines for the estimation of time to compromised tenability in fires*

ISO 13943, *Fire safety — Vocabulary*

ISO 29903-1:2020(E)

ISO 16312-1, *Guidance for assessing the validity of physical fire models for obtaining fire effluent toxicity data for fire hazard and risk assessment — Part 1: Criteria*

ISO 16730-1, *Fire safety engineering — Procedures and requirements for verification and validation of calculation methods — Part 1: General*

ISO 19703, *Generation and analysis of toxic gases in fire — Calculation of species yields, equivalence ratios and combustion efficiency in experimental fires*

ISO 19706, *Guidelines for assessing the fire threat to people*

3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 5725-1, ISO 13943, and the following apply.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

— ISO Online browsing platform: available at <https://www.iso.org/obp>

— IEC Electropedia: available at <http://www.electropedia.org/>

3.1

small-scale fire test

bench-scale test

fire test performed on a test specimen of small dimensions

Note 1 to entry: In these tests, the typical maximum length of a linear test specimen is less than 1 m. The typical maximum dimensions of a rectangular specimen are approximately 0,1 m.

[SOURCE: ISO 13943:2017, definition 3.346, modified — Note 1 to entry has been redrafted and the term "bench-scale test" has been added as a preferred term because of its use in this document.]

3.2

medium-scale fire test

fire test performed on a test specimen of small-medium size dimensions

Note 1 to entry: A fire test performed on a test specimen of which the maximum dimension is between 0,5 m and 1,0 m is here called a medium-scale fire test.

3.3

intermediate-scale fire test

fire test performed on a test specimen of medium dimensions

Note 1 to entry: A fire test performed on a test specimen for which the maximum dimension is between 1 m and 3 m is usually called an intermediate-scale fire test.

[SOURCE: ISO 13943:2017, definition 3.233, modified — Notes 2 and 3 to entry have been removed.]

3.4

large-scale fire test

fire test, that cannot be carried out in a typical laboratory chamber, performed on a test specimen of large dimensions

Note 1 to entry: A fire test performed on a test specimen of which the maximum dimension is greater than 3 m is usually called a large-scale fire test.

[SOURCE: ISO 13943:2017, definition 3.239]

3.5

real-scale fire test

fire test that simulates a given application, taking into account the real scale, the real way the item is installed and used, and the environment

Note 1 to entry: Such a fire test normally assumes that the products are used in accordance with the conditions laid down by the specifier and/or in accordance with normal practice.

[SOURCE: ISO 13943:2017, definition 3.325]

3.6

matrix effect

combined effect of all components of the sample other than the analyte on the measurement of the quantity

Note 1 to entry: Matrix effect (in analytical chemistry) as defined in IUPAC Compendium of Chemical Terminology^[1].

Note 2 to entry: The matrix effect in analysis of toxic gases in a fire effluent will be the combined effect from the components of the effluent on the analyte.

Note 3 to entry: If a specific component can be identified as causing an effect then this is referred to as interference.

4 Combustion conditions

4.1 General

The yields and nature of the fire effluent component from a fire test of any scale are determined by the involved fuels and the prevalent thermal and oxidative conditions in the current stage of the fire. These conditions also determine the burning rate of the products/materials and thus the rate of effluent generation. See ISO 16312-1.

During a fire test of a finished product, the combustion conditions are likely to change. These changes include the chemistry of the combustible item and the sufficiency of the ventilation.

Whether decomposition is flaming or non-flaming is a dominant factor in the production of toxic gases.

The combustion conditions under which toxic gas data are developed shall be as close to equivalent as possible between the physical fire models or test scales compared (see [Clause 6](#)).

NOTE 1 A large change in the rate of combustion can affect the degree of oxidation of the emitted effluent. Smaller changes in combustion rate can have no significant effect.

NOTE 2 Fire stages and the corresponding combustion conditions are described in ISO 19706.

4.2 Thermal environment

The thermal boundary conditions in a test include the external applied heat flux and the heat flux from any flaming combustion. Also of importance is the heat flux distribution among radiation, convection, and conduction.

The thermal environment sensed by the test specimen during combustion includes both gas temperature and the temperature of the sample material, as defined by the thermal boundary conditions.

4.3 Ventilation

The oxygen availability (ventilation) in the physical fire models compared determines the combustion conditions. Comparison among different methods requires characterization of the ventilation conditions in order to assess the degree of similarity.

For a given experiment, it is necessary to identify how the ventilation is characterized and whether the characterization is local or global.

For a physical fire model in which the fuel gasification rate and the entering oxygen flow and concentration are each controlled independently, the relative oxygen availability can be characterized by a fuel/oxygen equivalence ratio. For other models and real-scale fire tests, one or both of the terms in the equivalence ratio may not be well-known. In those cases, a broader characterization is used. This could be a global equivalence ratio or a term such as “underventilated burning” or “well ventilated burning”.

NOTE 1 Methods for calculating equivalence ratios for physical fire models are given in ISO 19703.

NOTE 2 The local air speed rate can be a significant factor in some fire tests. This applies especially for a tube furnace, where the air speed can affect the results of the combustion.

4.4 Characteristics of test specimens

The test specimens used for comparison of gas yields among physical fire models or between a physical fire model and a larger scale test shall be prepared from a single batch of the finished product or a single batch of each of the component materials. Alternatively, it shall be demonstrated that any differences in composition among the test specimens, tested in the different apparatus, do not affect the test outcome significantly.

For finished products that consist of a single, homogeneous material, the test specimen used in a physical fire model shall be prepared to accommodate the constraints of the test apparatus.

For specimens from non-homogeneous products, the test specimen shall also contain the same portions of the different materials present in the finished product in both tests compared.

For layered commercial products, an ideal physical fire model accommodates specimens that preserve the relationship of the layers. When this is not possible within the constraints of the model, the rationale for the configuration of the layers shall be documented.

NOTE The yields of toxic gases can depend on the surface exposed, and the timing and extent of penetration of the layers.

5 Toxic gas data

5.1 Identification of toxic species

The minimum set of gases that shall be considered are listed in ISO 13571.

Additional gases shall be appraised as warranted by the chemical composition of the test specimen and the finished product from which it is sampled.

5.2 Different expressions for toxic gas data

5.2.1 General

Subclause 5.2 contains a summary of different expressions typically used for toxic gas data obtained from fire tests and whether the data are suitable for comparison with similar data from other tests or as a basis for the prediction of large-scale results based on small scale data or vice versa.

The experimental data on toxic gases from a fire test can be expressed in several ways. From unrefined measurement data, which is often expressed as gas concentrations from a specific physical fire model, to data in higher degrees of refinement, e.g. yields. What is determined depends in part on the physical fire model used. See [Annex A](#) for information concerning the characteristics of different fire models.

The data can be in the form of scalar data or vector data. Some types of data are suitable for direct quantitative comparison, but others require a model for quantitative comparison. The most common quantities used in presentation of toxic gas data are given in [Table 1](#) below.

Table 1 — Common types of data on toxic gases from fire tests and properties for comparison

| Type of data | Typical units | SI-units | Scalar or vector data | Direct or indirect comparison | Qualitative or quantitative |
|--|-------------------------------------|--------------------------------------|---------------------------------|--------------------------------|---|
| Concentration of toxicants | ppm (v/v), i.e. µL/L | m ³ /m ³ | Scalar / Vector ^a | Indirect (Direct) ^d | Quantitative (Qualitative ^e) |
| The contribution to FED (or FEC) from individual toxicants | — | — | Scalar / Vector ^a | Indirect (Direct) ^d | Quantitative (Qualitative ^e) |
| Lethal toxic potency | g/m ³ | kg/m ³ | Scalar | Direct | Quantitative |
| Total amount of toxicants released | kg | kg | Scalar | Indirect (e.g. as yield) | Quantitative |
| Yields | g/g | kg/kg | Scalar (Vector) ^b | Direct | Quantitative |
| Production rates | g/s | kg/s | Vector (Scalar) ^c | Indirect (Direct) ^d | Quantitative (Qualitative ^e) |
| Normalized production rates | g·s ⁻¹ · m ⁻² | kg·s ⁻¹ · m ⁻² | Vector (Scalar) ^c | Direct | Quantitative |

^a Scalar if the model is steady-state or vector if the model studies changes in concentration over time.

^b Typically calculated as scalar data for the whole experimental period but can be calculated as vector data at each point in time.

^c Typically calculated as vector data at each point in time but can be calculated as scalar data for the whole experimental period.

^d Indirect comparison using a model (as shown in [Figure 1](#)) allows quantitative comparison. In some cases, direct comparison can be used for qualitative assessment.

^e Direct comparisons without the use of a model can provide qualitative information.

NOTE The "Type of data" given in [Table 1](#) are explained in [5.2.2](#) to [5.2.7](#).

5.2.2 Yields

Yield is typically the recommended comparison parameter. Yield is the measured mass of a toxicant generated during combustion, per unit mass of test specimen consumed in the fire test (mass loss) or alternatively, per unit mass of specimen exposed (mass-charge). The calculation of yields shall be made according to instructions in ISO 19703.

Yield is a quantitative comparison parameter and is independent of dilution or other apparatus specific parameters which do not impact on the combustion conditions.

NOTE 1 It can be difficult to calculate toxic gas yields in some large-scale physical fire models (e.g. ISO 9705-1^[2]), as the mass loss is normally not measured in these tests.

NOTE 2 Yield can be expressed relative to mass loss rate and thereby provide kinetic information or be a unique value representing an average of the complete test.

5.2.2.1 Mass of a toxic gas generated

A measurement or approximation of the mass of the toxic gas generated is essential to the calculation of toxic gas yields. The typical gas measurement during a fire test is the volumetric concentration of the gas in a volumetric sample of the total effluent. If the effluent from the fire test is not fully mixed, multiple concentration measurements across the effluent flow are necessary.

The concentration measurement shall be converted to a mass of the toxic gas generated during the sampling time interval using the ideal gas law. Corrections for condensation, solution, and deposition of the gas shall be included, as appropriate, in the calculation.

5.2.2.2 Mass of the test specimen consumed

A measurement or approximation of the consumed mass of the specimen is essential to the calculation of toxic gas yields.

The mass consumed shall be calculated in at least one of three ways.

- Mass loss based on continuous measurement of the remaining mass of the test specimen.
- Mass loss based on a final measurement of the remaining specimen mass.
- Estimation of the mass loss, when no gravimetric measurement is possible, using the chemical formulation of the test specimen and a carbon balance of the combustion products.

NOTE The third method can be in significant error if the chemical composition of the specimen residue is not the same as the initial chemical composition. This error can be reduced by determining the chemical composition of the residue.

5.2.3 Concentrations of toxicants

The concentrations measured in a specific physical fire model are a function of the degree of dilution in the sampling point. Concentrations are unique for the specific physical fire model and should not be used for a direct quantitative comparison. The agreement of relative concentrations between different physical fire models, can however be used for comparison.

The CO/CO₂ concentration ratio, for example, can be used as a comparison principle (see [6.2.2](#)).

NOTE Concentrations are normally expressed as volume fractions.

5.2.4 The contribution to FED (or FEC) from individual toxicants

The ranking of the different toxicants measured based on the relative contribution to the total toxicity using the FED (or FEC) concept is a semiquantitative comparison principle. The measured concentrations of toxicants are weighted relative to their toxic effects. See ISO 13344 for rodent lethality limits and ISO 13571 for human incapacitation limits.

5.2.5 Lethal toxic potency

Total lethal toxic potency of the fire effluents measured from the physical fire model is a quantitative comparison parameter. The predicted lethal toxic potency (LC50) has the unit g/m³ and requires data on mass loss or mass charge. The concept of lethal toxic potency referred to here is defined in ISO 13344.

5.2.6 Total amount of toxicant released

The total amount of a toxicant produced from a test is a unique parameter for a specific test only and is not a suitable comparison parameter unless weighted against surface area of sample, sample mass or mass loss.

5.2.7 Production rates

The production rate is temporally resolved data concerning the measured mass of a toxicant generated during combustion, e.g. expressed in g/s.

The production rate can be normalized relative to, e.g. the exposed surface area, and is in that case expressed in g · s⁻¹ · m⁻². Normalized production rates are directly comparable quantitative parameters.

5.3 Significance of analysis data

5.3.1 General

It is important to ascertain that the analytical techniques used for measurement of the toxic gas components compared between fire models, give comparable data. Factors to take into consideration are

- resolution of data,
- response time,
- selectivity of analytical technique, and
- effect of matrix.

Regarding selectivity, the same individual toxicants shall be included in a comparison. This is important if data are expressed as “total values” as is sometimes the case when referring to oxides of nitrogen (NO_x) or to polycyclic aromatic hydrocarbons (PAH).

The measurement matrix can induce errors and interfere with the analytical method. An assessment of possible matrix effects shall always be made as fire effluents is a difficult and variable matrix.

NOTE An example of influence of resolution is the comparison of the analytical results from methods giving peak-value data and methods giving averaged data.

5.3.2 Limit of detection (LoD)

If a specific toxicant is not identified in any of the scales compared, and if the LoD on both scales is toxicologically insignificant (see ISO 13571), then this information concerning the specific toxicant is sufficient for comparison based on LoD alone, i.e. a numerical comparison is not important.

If the LoD-value for a certain toxic gas results in a fractional effective dose/concentration that is insignificant relative to the total fractional effective dose/concentration for the sum of toxic gases present, the maximum contribution (i.e. LoD) from that gas could be regarded as insignificant.

NOTE 1 Information on methods for calculating limit of detection and limit of quantification is given in ISO 12828-1^[3].

NOTE 2 Methods for calculating fractional effective doses for asphyxiants and fractional effective concentrations for irritants are given in ISO 13571.

5.3.3 Measurement uncertainty

The trueness, repeatability, and reproducibility of analytical methods shall be known and taken into consideration in a determination of the measurement uncertainty.

NOTE 1 Methods for determination of the measurement uncertainty are given in ISO/IEC Guide 98-3^[4].

NOTE 2 For definitions of trueness, repeatability and reproducibility see ISO 5725-1.

6 Comparison/prediction of toxic gas data from different physical fire models

6.1 General

The toxic gas data determined according to [Clause 5](#) shall be compared for the same combustion conditions as defined in [Clause 4](#). Different principles can be applied for the comparisons of yields from different physical fire models, i.e. similar fire stages, CO/CO₂ ratios or equivalence ratios, see [6.2](#). Which methodology should be used for comparison of toxic gas data is dependent not only on which comparison principle is used but also on whether the comparison is direct or indirect as described

in 6.3.1. The corresponding methodology for prediction of toxic gas data from one physical fire model based on another is given in 6.4.

Yield is the recommended comparison parameter although other parameters could be compared, see 5.2.

6.2 Comparison principles

6.2.1 Fire stages

The fire stage simulated by the physical fire model shall be determined in accordance with ISO 19706. If the combustion of the sample encompasses more than one fire stage during a test, the time interval for each fire stage shall be identified.

Comparison of toxic gas yields shall be for the same fire stage or for a succession of identical fire stages. For tests in which there is a succession of fire stages, a basis for the contribution of each stage to the toxic gas yields shall be determined, e.g. the equivalence ratio.

NOTE 1 For smouldering combustion, (fire stage 1.a in ISO 19706), the equivalence ratio is difficult to define.

NOTE 2 For oxidative pyrolysis (fire stage 1.b in ISO 19706), there are insufficient experimental data on which to base an estimate of dependence of toxic product yields on equivalence ratio.

NOTE 3 For anaerobic pyrolysis (fire stage 1.c in ISO 19706), there is no oxygen present, and thus the equivalence ratio concept is immaterial.

NOTE 4 For well-ventilated flaming fire conditions (fire stage 2 in ISO 19706), where the fuel/air equivalence ratio is less than 0,5, the yields of some important toxic gases have been shown to be relatively insensitive to the equivalence ratio.

NOTE 5 For underventilated flaming fire conditions (fire stage 3 in ISO 19706), where the fuel/air equivalence ratio is approximately 1 or higher, the yields of some important toxic gases have been shown to be sensitive to the value of the equivalence ratio.

6.2.2 CO/CO₂-ratio

The CO/CO₂-ratio reflects the availability of oxygen for combustion and thus describes the combustion conditions in many cases. The CO/CO₂-ratio measured from physical fire models can, in some cases, be used as the basis for comparison.

This principle is not appropriate for sample materials containing halogens (e.g. PVC materials) as halogens interfere with the gas phase combustion process and promote CO-production irrespective of the prevailing combustion conditions.

The combustion temperature and the upper layer temperature in a large-scale test could influence the production of toxic gases and shall be taken into account.

NOTE 1 ISO 19706 uses the CO/CO₂-ratio in characterization of the different stages of a fire. CO₂/CO-ratio can alternatively be used to avoid working with small decimal numbers with the same result.

NOTE 2 In large-scale fire tests the CO/CO₂-ratio can vary depending on the sampling point, which is an effect of post fire reactions with oxidation of CO to CO₂ downstream of the sampling point.

6.2.3 Equivalence ratio

The fuel/air equivalence ratio describes the amount of oxygen available for the combustion reactions compared to the stoichiometric need for complete combustion to CO₂ and H₂O and is suitable as a basis of comparison. The equivalence ratio can be measured in a fire test using special measurement devices^[5] or calculated from the mass loss of the sample and the oxygen flow to the combustion site.

Boundaries for the equivalence ratio shall be defined for the physical fire model; a local equivalence ratio for the combustion or a global equivalence ratio for the total volume of, e.g. a test compartment.

The combustion temperature and the upper layer temperature in a large-scale test could influence the production of toxic gases and shall be taken into account.

NOTE The calculation of equivalence ratios is described in ISO 19703.

6.3 Comparison methodology

6.3.1 General

The primary basis for comparison of data from different tests is the yield of each toxic gas. The yield data are determined from a test specimen of a material burning under conditions simulating a specific fire stage using a specific bench-scale physical fire model. These data are compared with toxic gas yield data from the same stage of burning using another bench-scale physical fire model, or using a large-scale test of a finished product.

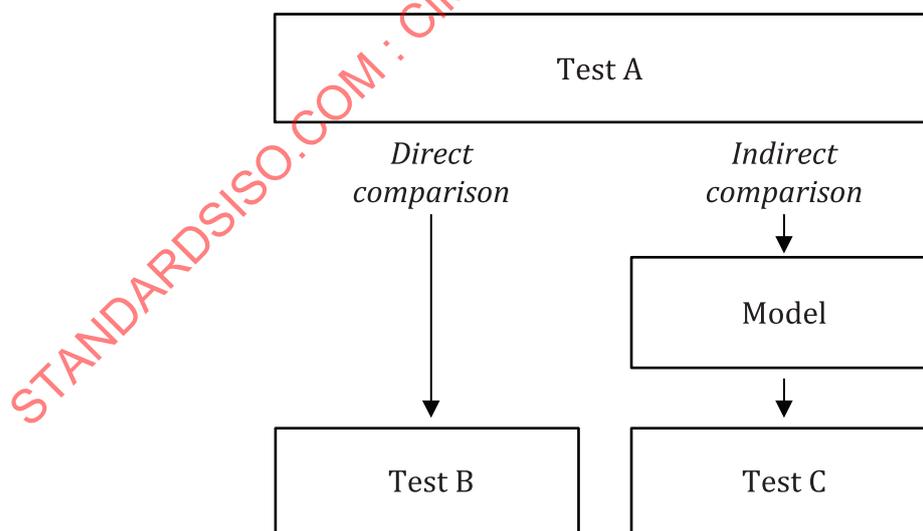
For data concerning toxic gases suitable for a direct comparison (see 5.2.1) the data can be compared directly between fire tests of different scales. For data on toxic gases not suitable for a direct comparison (see 5.2.1), a model is required in order to facilitate an indirect comparison. Such a model may include dilution factors, a factor for normalization of exposed fuel surfaces or other methods for recalculating the original data in a format for comparison.

The two different cases of comparison, direct comparison and indirect comparison, are illustrated in Figure 1.

The comparison methodology is defined in 6.3.2 to 6.3.4 and outlined in Figure 2.

NOTE 1 See ISO 16312-1 for accuracy of physical fire models.

NOTE 2 EN 17084 [6] is an example of a standard including test requirements for toxicity where normalization models (a model for indirect comparison) for toxic gases are used.



NOTE A model can include, e.g. dilution factors, exposed fuel surface.

Figure 1 — Schematic figure of direct and indirect comparison of toxic gas data

6.3.2 Assessment of available data

The first assessment of available data requires determination of the combustion conditions of the two tests compared, for the periods where the gas production data are compared. If the combustion conditions are not comparable, the data on toxic gases cannot be compared between the two tests.

Once it has been established that the combustion conditions are comparable, an assessment of the sampling and analysis techniques used is made. Correction of the data based on sampling and analysis methods should be made as outlined in [Annex B](#).

The type of data available from the two tests selected for comparison (e.g. yields) should be assessed to ascertain that the data are comparable. Finally, it is necessary to make an assessment of the measurement uncertainty for the data compared from the two tests. For a meaningful comparison the measurement uncertainties should not differ in magnitude.

For well-ventilated tests the exact agreement in equivalence ratio is of less importance compared to tests with varying degrees of vitiation where the equivalence ratio is critical for the production of most toxic gases (see [6.2.3](#)). Where possible, comparison of a specific toxic species between different physical fire models or between a physical fire model and a real-scale fire test should be based on the yields of the specific toxic species generated under comparable combustion conditions, as this type of comparison generally provides the best agreement or prediction capability.

6.3.3 Comparison of data

Select the comparison principle for the toxic gas data (see [6.2](#)) based on the assessment made in [6.3.2](#).

For an indirect comparison apply an appropriate scaling model to the data (see [Figure 1](#)).

Compare the data for each selected toxic gas and make an assessment of the numerical agreement based on the overlap between the numerical ranges defined by their respective measurement uncertainties.

For comparison of vector data, the methodology for comparison of vector data as specified in ISO 16730-1 shall be used.

NOTE The ISO 16730 series addresses the assessment, verification and validation of calculation methods for fire safety engineering in general. ISO 16730-1 describes a method to quantify the similarities and differences of two curves such as the time history of the upper layer temperature for a model prediction and an experiment. This is done by treating the curves as infinite dimensional vectors and then using vector analysis to describe the differences. This analysis provides a quantitative method of verifying fire models and quantifying the uncertainty in experimental data. While there are a variety of numbers that could be generated to describe the differences between two curves or a set of curves, two values are focused on as giving appropriate information for determining the differences. The first, relative difference, gives a value for how different the two curves are. It is a positive real valued function that gives a 0 for identical curves and the larger the value the larger the difference. The second, the cosine, gives a measure of how the shapes of the two curves compare. It returns values from 1 to -1 with 1 meaning the curves have the same shape and -1 meaning the curves are mirror images and 0 if the curves have nothing in common. ISO 16730-1 describes the method and appropriate formulae and shows examples of how to do the comparison.

6.3.4 Assessment of agreement

Agreement between two physical fire models or a physical fire model and a real-scale test is affirmed if the yields of all toxic gases are in agreement.

In cases when the comparison of an individual toxic gas component does not agree (outlier) the models/ tests compared could still be in general agreement. An explanation of the reason for the outlying toxic gas component should be found.

If an outlier is due to differences in combustion conditions between the models/tests compared, the comparison between these sets of data shall be regarded as not valid.

Requirements for agreement between a bench-scale fire test and a real-scale fire test given in ISO 16312-1 shall be followed.

NOTE An explanation for an outlier can often be found through re-assessment of the sampling and analysis of that toxic gas component.

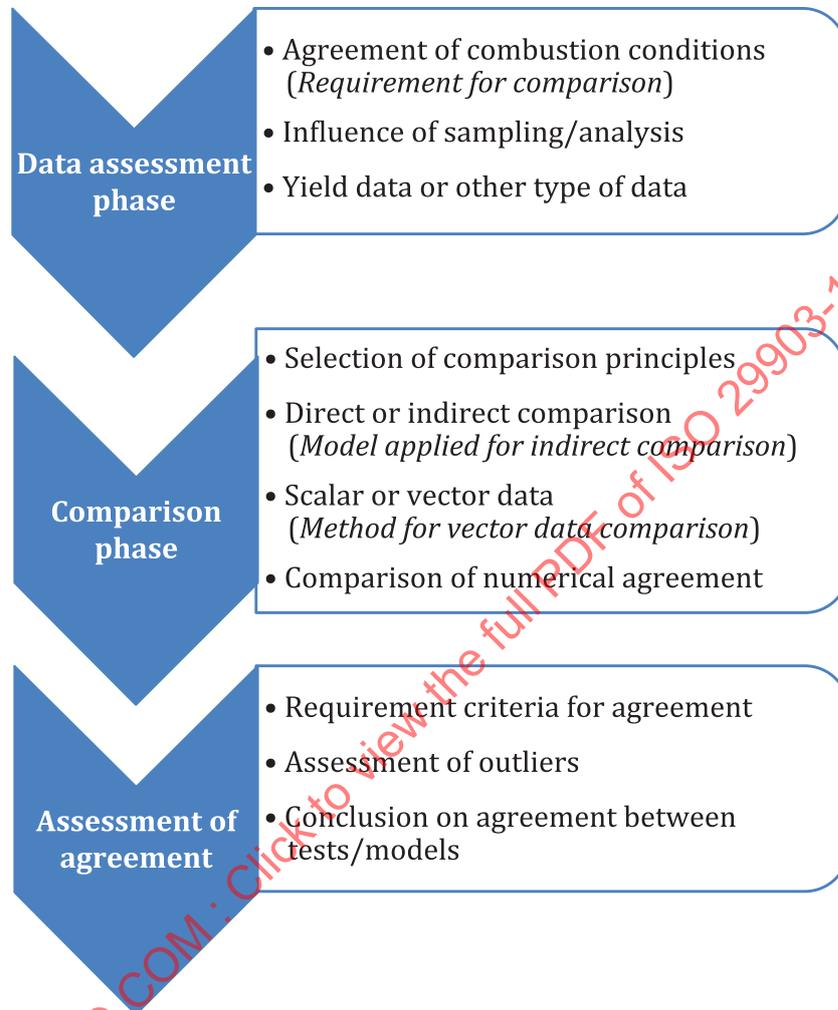


Figure 2 — Flow chart of comparison methodology

6.4 Prediction of data from one fire model to another

Toxic gas data can be predicted in one scale based on results from another if it has been shown that data from these models are comparable using the principles given in 6.3.

If, for example, the fire models of smaller and larger scale of the same fire stage have shown correlation, it is possible to use the results from one such model to predict the results from the other model. The method for prediction will depend on the type of information to be scaled.

7 Documentation

A report or any other documentation shall include the following information for each fire test included in a comparison:

- a) The fire tests for which the toxic gas data are being compared.
- b) The fire stage(s) examined and the prevailing combustion conditions.

- c) The type of data compared (e.g. yields) and any assumptions in calculations of final data used in the comparison.
- d) Any influence of sampling and analysis techniques used.
- e) The uncertainties in the measurements and the cumulative uncertainty in the final data used in the comparison.
- f) The comparison principle used (i.e. fire stage, CO/CO₂ ratio or equivalence ratio).
- g) If direct or indirect comparison is made.
- h) The comparison model used in the case on an indirect comparison.
- i) The result of the comparison and the basis for determining the degree of agreement.

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Annex A (informative)

Characteristics of physical fire models

A.1 General

Bench-scale tests used for toxicity testing are ideally designed to reproduce a single fire stage/combustion condition. In large/real-scale tests, different and variable combustion conditions can be present simultaneously at different locations in the test system. Even if this is not the case, large/real-scale tests will often exhibit different fire stages/combustion conditions at different times during the test.

In a comparison of data concerning toxic gases between different fire tests it is necessary to consider the purpose and aim of the tests compared. [Table A.1](#) summarizes the scales and parameters studied by some common fire test methods.

For example, a large open pool fire with nominally well-ventilated conditions contains different zones of local combustion conditions, from the centre of the pool with a lack of oxygen and a high heat flux to the periphery of the pool where there is good availability of oxygen and effective combustion. During a large-scale enclosure fire test, the fire can pass from one fire stage to another, from well-ventilated pre-flashover to under-ventilated post-flashover conditions. The variability of the combustion conditions in large-scale tests shall be considered in a comparison.

Table A.1 — Scale and parameters of some common fire test methods

| Size / Scale | Parameter studied | Example of fire tests |
|---------------------------------------|--|--|
| Material / Bench-scale | Potency of release of toxic combustion products | — ISO/TS 19700 ^[7] — NF X 70-100 ^[8] |
| Finished material / Bench-scale | (+) Multi-layer thermal / physical / chemical effects | — ISO 5660-1 ^[9] — ISO 12136 ^[10] — ISO 5659-2 ^[11] |
| Semi-finished product / Medium-scale | (+) Fire growth | — ISO 21367 ^[12] |
| Finished product / Intermediate-scale | (+) Joints and assemblies (+) Mounting and fixing (partial) | — EN 13823 ^[13] |
| Finished building / Large-scale | (+) Mounting and fixing in real conditions | — ISO 9705-1 ^[2] — ISO 13784-1 ^[14] |
| Real-scale | (+) Real-scale end-use conditions | — ISO 24473 ^[15] — Real-scale ad hoc tests |

NOTE 1 The different parameters additionally included with an increase in scale are marked with a (+) in [Table A.1](#).

NOTE 2 The test methods given as examples in [Table A.1](#) can allow toxic gas measurement even if this is not included in the standard test protocol at present.

A.2 Bench-scale tests

A.2.1 General

A general characterization of a bench-scale test appropriate for obtaining data on fire effluent toxicity can be found in ISO 16312-1. Individual bench-scale tests are characterized and evaluated for their utility for generating fire effluent toxicity data in ISO/TR 16312-2^[16].

A.2.2 Tests with constant combustion conditions

Physical fire models capable of attaining constant combustion conditions during the complete test or for prolonged steady-state periods are most fitting for producing toxic gas data suitable for comparison and interpolation. Only a few test methods belong to this category.

NOTE The ISO tube furnace method, ISO/TS 19700^[6], the equivalent IEC tube furnace method, IEC 60695-7-50^[17] and the German tube furnace method, DIN 53436^[18], are examples of flow-through tests which are designed for attaining constant combustion conditions throughout the test period. In these tests, the oxygen flow and the fuel flow through the hot zone of the apparatus can be regulated to attain the desired combustion condition.

A.2.3 Tests with non-constant combustion condition

Most bench-scale tests fall into this category. This includes all tests where the test specimen is placed stationary in a closed or semi-closed test apparatus and is exposed to a constant source of heat. The specimen itself will in such cases influence the conditions by its reaction-to-fire behaviour and its finite mass.

This includes stationary tube furnace tests as, e.g. the French test, NF X 70-100^[8], where the combustion conditions will vary by the burning-phases of the test specimen through the test.

This includes closed cabinet tests, e.g. applications with the ISO 5659-2 smoke box^[11], such as that by the International Maritime Organization^[19], where the combustion effluents from the test specimen will vitiate the atmosphere in the cabinet as the test proceeds.

A category of bench-scale tests that can produce partially constant combustion conditions includes the Cone Calorimeter ISO 5660-1^[9], and the Fire Propagation Apparatus ISO 12136^[10]. For both of these tests a stationary sample is heated and ignited under exposure from a radiant source.

NOTE Of these methods, the Cone Calorimeter is the most widespread apparatus, although the Fire Propagation Apparatus is the apparatus most suitable for toxicity testing as the oxygen flow for the combustion can be regulated.

A.3 Large-scale tests

A.3.1 General

Large-scale tests normally involve testing of a finished product, which can contain different materials and layers. Further, the combustion conditions in large-scale tests can include several fire stages as a result of the burning-phases of the test specimen.

A.3.2 Enclosure tests

Enclosure tests can include several fire stages depending on the reaction-to-fire behaviour of the tested product. The combustion conditions can proceed from initial non-flaming oxidative pyrolysis,