
**Cigarettes — Determination of
selected carbonyls in the mainstream
smoke of cigarettes with an intense
smoking regime — Method using high
performance liquid chromatography**

*Cigarettes — Dosage de carbonyles sélectionnés dans le courant
principal de la fumée de cigarette avec un régime de fumage intense
— Méthode par chromatographie liquide haute performance*

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ISO copyright office
CP 401 • Ch. de Blandonnet 8
CH-1214 Vernier, Geneva
Phone: +41 22 749 01 11
Email: copyright@iso.org
Website: www.iso.org

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see www.iso.org/patents).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation of the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT), see www.iso.org/iso/foreword.html.

This document was prepared by Technical Committee ISO/TC 126, *Tobacco and tobacco products*.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

Introduction

At the outset of this work, discussions in the CORESTA (www.coresta.org) Special Analytes Sub-Group (since 2017 the Sub-Group changed its name to Smoke Analytes) determined that most laboratories used a method involving derivatization of carbonyls with 2,4-dinitrophenylhydrazine (DNPH) because they considered it the most suitable. This was chosen as the basis of the CORESTA Recommended Method (CRM). The CRM comprised smoke collection in impinger traps, derivatization of carbonyls with DNPH followed by their determination using reversed phase High Performance Liquid Chromatography with Ultra Violet or Diode Array Detection (HPLC-UV or HPLC-DAD).

This document was produced from a 2012 collaborative study involving 19 laboratories from 11 countries and included 10 samples with different tar yields^{[1][2]}. This method includes recommendations about critical steps that should be controlled to provide data as robust and consistent as the repeatability and reproducibility data provided. Cigarettes were smoked using the intense smoking regime parameters given in Health Canada Official Method T-115. At the time the collaborative study was conducted, the study protocol stipulated the use of Health Canada Official Method (T-115) for intense conditions as there was not an International Standard that defined intense smoking conditions. ISO 20778 was published in 2018 and is equivalent to Health Canada Intense conditions. Statistical evaluations were carried out according to ISO 5725-1^[3] and ISO 5725-2^[4].

No machine smoking regime can represent all human smoking behaviour.

- It is recommended that cigarettes also be tested under conditions of a different intensity of machine smoking than those specified in this document.
- Machine smoking testing is useful to characterize cigarette emissions for design and regulatory purposes, but communication of machine measurements to smokers can result in misunderstandings about differences in exposure and risk across brands.
- Smoke emission data from machine measurements may be used as inputs for product hazard assessment, but they are not intended to be nor are they valid as measures of human exposure or risks. Communicating differences between products in machine measurements as differences in exposure or risk is a misuse of testing using International Standards.

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WARNING — The use of this document can involve hazardous materials, operations and equipment. This document does not purport to address all the safety problems associated with its use. It is the responsibility of the user of this document to establish appropriate safety and health practices and determine the applicability of any other restrictions prior to use.

1 Scope

This document specifies a method for the determination of selected carbonyls (formaldehyde, acetaldehyde, acetone, acrolein, propionaldehyde, crotonaldehyde, 2-butanone and *n*-butyraldehyde) as their 2,4-dinitrophenylhydrazones by reversed phase HPLC-UV/DAD in mainstream smoke using an intense smoking regime.

2 Normative references

The following documents are referred to in the text in such a way that some or all of their content constitutes requirements of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 3402, *Tobacco and tobacco products — Atmosphere for conditioning and testing*

ISO 8243, *Cigarettes — Sampling*

ISO 20778, *Cigarettes — Routine analytical cigarette smoking machine — Definitions and standard conditions with an intense smoking regime*

3 Terms and definitions

No terms and definitions are listed in this document.

ISO and IEC maintain terminological databases for use in standardization at the following addresses:

- ISO Online browsing platform: available at <https://www.iso.org/obp>
- IEC Electropedia: available at <http://www.electropedia.org/>

4 Principle

Cigarettes are smoked on a standard smoking machine as specified in ISO 20778 that has been fitted with impingers, but without a glass fibre filter pad as described in ISO 20778 [e.g. Cambridge filter pad¹⁾ (CFP), for example of equivalent product] and the filter pad holder, with the ISO 20778 smoking regime.

The carbonyls in mainstream tobacco smoke are trapped by passing each puff through an impinger device containing an acidified solution of 2,4-dinitrophenylhydrazine (DNPH) in 1:1 acetonitrile:water.

1) Cambridge filter pad (CFP) is an example of a suitable product available commercially. This information is given for the convenience of users of this document and does not constitute an endorsement by ISO of this product.

An aliquot of the smoke extract is then syringe-filtered and diluted with tris-(hydroxymethyl)-aminomethane dilution solution.

The samples are subjected to analysis using reversed phase HPLC-UV or HPLC-DAD.

5 Apparatus

The usual laboratory apparatus and equipment for use in preparation of samples and standards is needed in addition to the list provided below.

5.1 Equipment for conditioning of tobacco products.

5.2 Equipment for butt length marking.

5.3 Equipment for smoking of tobacco products, complying with ISO 20778.

5.4 Impingers for trapping mainstream smoke.

5.5 Polyvinylchloride (PVC) tubing, suitable for connection of the trapping system.

5.6 Analytical balance, capable of measuring to four decimal places.

5.7 Amber glass flasks, of capacities 10 ml, 25 ml, 200 ml, 1 l and 2 l.

5.8 Pipettes, of appropriate volumes.

5.9 Glass graduated measuring cylinders, of capacities 25 ml, 50 ml and 100 ml.

5.10 Dispenser, capable of delivering 35 ml.

5.11 Hot plate/stirrer.

5.12 Syringe filter, 0,45 µm PVDF or equivalent.

5.13 Disposable syringes, 5 ml.

5.14 Disposable glass Pasteur pipettes.

5.15 Amber Autosampler vials, caps and PTFE faced septa.

5.16 HPLC system, consisting of:

- tertiary gradient pump;
- auto-sampler with appropriate sampling loop;
- UV and/or DAD detector;
- data collection system;
- LC column: 250 mm × 4 mm, 100 Å, Reversed Phase (RP) C18 endcapped (5 µm), or equivalent;
- disposable guard column: 4 mm × 4 mm RP C18 endcapped (5 µm), or equivalent;
- vacuum filter;

- amber glass bottles 1 l and 4 l;
- desiccator.

6 Reagents

- 6.1 **Acetonitrile**, MeCN, HPLC grade.
- 6.2 **Isopropanol**, IPA, HPLC grade.
- 6.3 **Ethyl acetate**, HPLC grade.
- 6.4 **Tetrahydrofuran**, THF, HPLC grade.
- 6.5 **Ethanol**, HPLC grade.
- 6.6 **Phosphoric acid**, 85 %.
- 6.7 **Deionized water**, resistivity >18,0 M Ω .cm at 25 °C.
- 6.8 **Formaldehyde-DNPH**, min. 99 %.
- 6.9 **Acetaldehyde-DNPH**, min. 99 %.
- 6.10 **Acetone-DNPH**, min. 99 %.
- 6.11 **Acrolein-DNPH**, min. 99 %.
- 6.12 **Propionaldehyde-DNPH**, min. 98 %.
- 6.13 **Crotonaldehyde-DNPH**, min. 99 %.
- 6.14 **2-Butanone-DNPH**, min. 98 %; methyl ethyl ketone-DNPH derivative.
- 6.15 ***n*-Butyraldehyde-DNPH**, min. 99 %.
- 6.16 **Tris-(hydroxymethyl)-aminomethane**, ACS reagent grade²⁾.
- 6.17 **2,4-Dinitrophenylhydrazine (DNPH) (approximately 30 % water)**.

7 Preparation

7.1 Preparation of glassware

Glassware shall be cleaned and dried in such a manner as to ensure that contamination from glassware does not occur.

All possible sources of contamination shall be removed from the work area (e.g. acetone solvent wash bottles).

2) A reagent that meets the requirements of the American Chemical Society (ACS) Committee on Analytical Reagents.

7.2 Preparation of solutions

7.2.1 DNPH solution (using phosphoric acid)

Add approximately 150 ml deionized water to a 200 ml volumetric flask, then carefully add 28 ml of 85 % phosphoric acid (approximately 2,05 mol/l) and mix the solution.

Make up the solution to volume with deionized water.

Weigh approximately 6,8 g (24,0 mmol should be achieved) of DNPH (approximately 30 % water) into a 2 l volumetric flask and add 1 l of acetonitrile. Dissolve DNPH by alternately gently swirling the flask. Make sure there are no crystals remaining.

WARNING — Do not sonicate as a precipitation of DNPH may occur.

If using re-crystallized DNPH, weigh 4,8 g to achieve the same molarity (see [Annex A](#)).

After the DNPH is dissolved, add 58 ml of the diluted phosphoric acid solution while gently mixing. Dilute to volume with deionized water. The colour of the solution will become bright orange upon addition of the deionized water.

The addition of acid or water will cool the solution and may initiate the precipitation of the DNPH. Add the acid or water slowly. Gentle swirling may be required to maintain the solution at room temperature and to prevent the precipitation of DNPH. If crystals appear, do not sonicate.

Store the solution in an appropriately sized amber bottle at room temperature. This solution has been shown to be stable for one week. Stability shall be assessed by each laboratory.

7.2.2 Tris-(hydroxymethyl)-aminomethane dilution solution, 80:20 (volume fraction), MeCN: aqueous solution.

Dissolve 2,00 g of tris-(hydroxymethyl)-aminomethane in 200 ml of deionized water in a 1 l volumetric flask. Dilute to volume with acetonitrile.

Store in a 1 l amber bottle with PTFE-lined cap or equivalent at ambient temperature.

7.3 Preparation of standards

7.3.1 HPLC calibration standards and working solutions

The calibration should cover the concentration range of interest.

7.3.1.1 Primary carbonyl standards

Where available, certified reference solutions of the selected hydrazones can be used.

Weigh the hydrazones as described in [Annex B](#) into individual 25 ml volumetric flasks and dissolve in acetonitrile. Record the concentrations of the free aldehyde equivalents in µg/ml.

These solutions have been shown to be stable for up to one year when stored at approximately 4 °C. Stability and storage time should be checked by the laboratory.

7.3.1.2 Secondary carbonyl standards

Pipette predetermined volumes ([Annex B](#)) of each primary hydrazone standard into a 25 ml volumetric flask and dilute to the mark with acetonitrile.

Store at approximately 4 °C. Stability and storage time should be checked by the laboratory.

7.3.2 Carbonyl working standards

Take appropriate volumes (0,050 ml to 10 ml) of the secondary carbonyl standard (7.3.1.2) and dilute to 10 ml with acetonitrile to prepare calibration standards with approximate carbonyl concentrations (see Annex B).

Transfer to auto-sampler vials.

The calibration range described in Annex B has been shown to be suitable; however, it can be necessary to adjust the calibration range depending on factors such as the number of cigarettes smoked and the carbonyl yields of the test cigarettes. The user shall ensure the low calibration standard has a sufficient signal to noise ratio for accurate quantitation ($\geq 10:1$) and that the calibration curve is linear.

These solutions have been shown to be stable for 20 days when stored at approximately 4 °C. Stability and storage time should be checked by the laboratory.

8 Sampling

Carry out sampling in accordance with ISO 8243.

9 Tobacco product preparation

Condition the cigarettes in accordance with ISO 3402.

10 Sample generation — Smoking of cigarettes

10.1 General

Cigarettes are smoked in accordance with ISO 20778.

10.2 Smoking machine setup

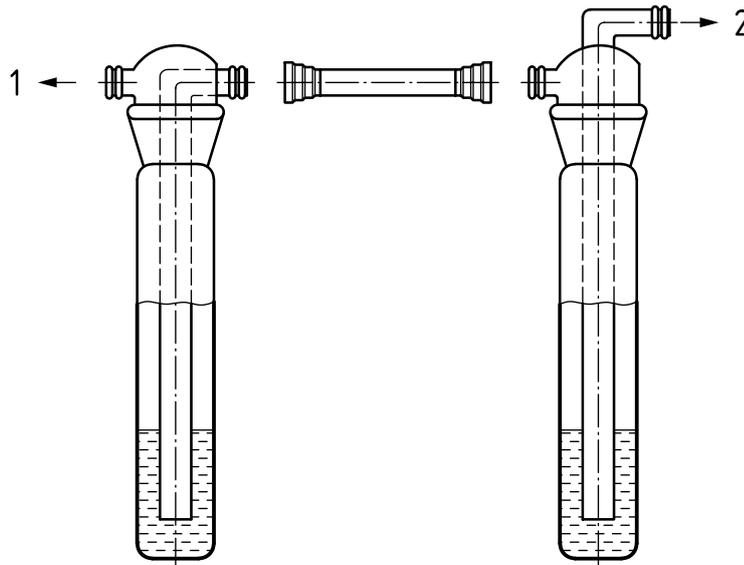
An analytical cigarette smoking machine complying with the requirements of ISO 20778 is required with the following modifications as detailed below.

No filter pad is required in the set up and therefore puff count information is the only means of monitoring whether the smoking process is controlled.

Add 35 ml of DNPH solution to each impinger. Assemble the carbonyl mainstream apparatus on the smoking machine without using the filter pads and filter holders (Figure 1).

A volume other than 35 ml of DNPH solution may need to be added to each impinger depending on the particular style of impinger used.

Check and adjust the puff volume drawn by the smoking machine at all channels at the cigarette end of the port as described in ISO 20778 with the impingers and DNPH in line.



Key

- 1 attached to tubing leading to smoking machine puffing piston
- 2 back of cigarette holder

Figure 1 — Example of a suitable trapping system

Since there is no standard impinger design, it is recommended to verify the trapping efficiency when validating this method. To check the trapping efficiency of the method, add an additional backup impinger and follow the method accordingly. Analyse each impinger individually for the compounds of interest. If the impinger trapping system effectively traps 95 % of the analytes of interest, then only the prescribed number of impingers is required to trap all the carbonyls effectively. Breakthrough or poor trapping efficiency can be due to the design of the impinger or cigarettes with high carbonyl yields.

To determine whether a leak has occurred in the smoking machine impinger setup, use a leak tester. If the fluid column does not maintain its position but drops then there is a leak in the system.

10.3 Smoking

10.3.1 General

The cigarettes are smoked according to ISO 20778 with the modifications given in [10.3.2](#) and [10.3.3](#).

10.3.2 Linear smoking

Two cigarettes are smoked per replicate.

10.3.3 Rotary smoking

Five cigarettes are smoked per replicate.

The number of cigarettes smoked may be adjusted to bring the samples within the calibration range.

11 Sample analysis

11.1 Preparation of mainstream smoke extract solution

After collection of the smoke extract, the contents of the two impingers should be combined, by adding the contents of the second impinger into the first, and mixed. This solution may be used to rinse the gas transfer lines if needed.

There can be sample losses in the gas transfer lines leading from the smoking machine to the impinger as well as the gas transfer lines connecting the two impingers. The laboratory shall determine the amount of losses in these transfer lines. If these losses are significant, the contents of the impingers shall be used to rinse the lines.

Allow the DNPH smoke extract solution to sit for five to 30 min^{[5][6]} before continuing with sample preparation.

Pipette 6 ml of tris-(hydroxymethyl)-aminomethane solution into an appropriately sized glass flask.

Pipette 4 ml of syringe-filtered DNPH smoke extract into the flask.

Ratios may be adjusted according to laboratory preparation procedures.

Transfer a portion of this solution to an amber auto-sampler vial.

Cap the vials with PTFE faced septa and store at ambient temperature until analysed.

Repeat above steps for each smoke extract sample.

11.2 Reversed phase high performance liquid chromatography

Set up and operate the HPLC-UV or HPLC-DAD system in accordance with the manufacturer's instruction.

The following suggested parameters have been found to be suitable for separation.

11.2.1 Chromatographic conditions

EXAMPLE

- Column temperature: 30 °C
- Auto-sampler tray temperature: ambient
- Injection volume: 20 µl
- UV or DAD detection: at 365 nm

11.2.2 Mobile phase reagents

- **Solvent A:** Prepare 2 l of 30 % acetonitrile, 10 % THF, 1 % IPA in deionized water, filter and degas (UHP helium sparged)
- **Solvent B:** Prepare 2 l of 65 % acetonitrile, 1 % THF, 1 % IPA in deionized water, filter and degas (UHP helium sparged)
- **Solvent C:** Acetonitrile
- Sample wash: Solvent A

11.2.3 HPLC separation conditions

Standards and samples are analysed by HPLC operated at a flow rate of 1,5 ml/min (Table 1). The injection volume is 20 µl.

Table 1 — HPLC Mobile phase gradient

Time (min)	Composition		
0,0	100 % A	0 % B	0 % C
8,0	70 % A	30 % B	0 % C
20,0	47 % A	53 % B	0 % C
27,0	0 % A	100 % B	0 % C
30,0	0 % A	0 % B	100 % C
32,0	0 % A	0 % B	100 % C
34,0	95 % A	5 % B	0 % C
Method end	—	—	—
Equilibrate for 10 min	100 % A	0 % B	0 % C

The chromatographic conditions can be different for different instrument configurations and columns. However, as a general system suitability check, the elution pattern should be similar to the example chromatograms shown in Annex C (Figure C.1) and Annex D (Figure D.1).

11.3 Calculation

11.3.1 Calibration curve

Generate a calibration curve for each individual carbonyl by plotting standard peak areas against their respective concentrations.

11.3.2 Sample quantification

The concentration of selected carbonyls in smoke samples is quantified by the external standard method. An example of a typical chromatogram is shown in Annex D. The identification of peaks is by comparison of retention times with standards, and the spiking of smoke samples.

Carbonyl concentrations are reported in micrograms per millilitre by the chromatography software.

The amount of carbonyl yields in the mainstream smoke of cigarettes, m_c , expressed in micrograms per cigarette, is given by Formula (1):

$$m_c = [A]d \frac{V}{N_{\text{cig}}} \quad (1)$$

where

[A] is the concentration of the analyte, in micrograms per millilitre, from the linear regression reported by the software;

d is the dilution factor (final volume/aliquot volume);

V is the extraction solvent volume, in millilitres;

N_{cig} is the number of cigarettes smoked.

NOTE It was observed that under the conditions chosen for the derivatization an isomerization of acetaldehyde hydrazone occurs. The resulting additional isomer can be separated by HPLC and elutes under the described separation conditions in front of the main isomer (Annex D).

For the calculation of acetaldehyde yield, the area of both isomers should be calculated to obtain correct results.

If the carbonyl yields are above the top calibration standard, and the trapping efficiency was demonstrated to be sufficient, the sample should be diluted to fit into the calibration curve and re-analysed. It is recommended to use the same batch of tris-(hydroxymethyl)-aminomethane dilution solution and DNPH trapping solution when diluting the sample. Dilute the sample with the appropriate volumes of sample and tris-(hydroxymethyl)-aminomethane dilution solution so that the sample is within the calibration range.

The expression of the laboratory data depends on the purpose for which the data are required, and the level of laboratory precision. Any further statistical analyses should be calculated and expressed on the basis of the laboratory data before any rounding has taken place.

Carbonyl yields in the mainstream smoke of cigarette in units of microgram per cigarette ($\mu\text{g}/\text{cig}$) shall be reported rounded to the nearest 0,1 μg .

12 Repeatability and reproducibility

12.1 General

In 2012, mean yield, r and R data were obtained from a collaborative study involving 19 laboratories. This provided data on the measurement of the selected carbonyls in five replicate analyses of 10 samples (seven commercial products, KR 3R4F, KR 1R5F, and CORESTA Monitor CM6) performed under the smoking regime Health Canada Method T-115, that using the same parameters as the intense smoking regime, ISO 20778. The collaborative study samples are identified in [Table 2](#), and calculated statistical data of the individual selected carbonyl compounds are summarised in [Tables 3](#) to [10](#). The statistical evaluation was performed according to ISO 5725-2.

The Carbonyl analysis does not generate total particulate matter (TPM) values, for this reason, the TPM yields for Benzo[a]pyrene (B[a]P) from the same 2012 collaborative study are included in [Table 2](#). These TPM values are included because they demonstrate the range of cigarette deliveries for the products used in this study.

Table 2 — Cigarette test samples of the 2012 collaborative study

Sample	Intense TPM yield ^a (mg/cig)	Product/Blend type
CM6	42,0	CORESTA Monitor 6 Test Piece
1R5F	26,7	Kentucky Reference 1R5F
3R4F	39,9	Kentucky Reference 3R4F
Sample 1	37,1	Dark air-cured
Sample 2	35,3	American blended
Sample 3	30,7	American blended
Sample 4	24,7	Virginia blended
Sample 5	17,0	Virginia blended
Sample 6	32,8	Virginia blended
Sample 7	21,3	Charcoal filtered

^a TPM (total particulate matter): The TPM values are the collaborative study means from the B[a]P analysis after removal of outliers^[4].

12.2 Results from the 2012 collaborative study

Results from the collaborative study including number of laboratories included in the study, mean yields obtained from Health Canada T-115 smoking regime, and calculated repeatability and reproducibility data for the individual carbonyls are given in [Tables 3](#) to [10](#).

Table 3 — Formaldehyde

Sample description	N ^a	Mean	<i>r</i>	<i>R</i>
		(µg/cigarette)		
CM6	17	104,2	21,1	50
1R5F	19	38,7	11,6	27
3R4F	19	76,0	15,5	38
1	15	39,0	7,5	21
2	15	64,0	15,5	39
3	14	71,9	18,4	37
4	14	140,3	33,6	107
5	15	87,9	24,0	87
6	15	164,2	35,4	84
7	14	72,8	16,0	34

^a N = number of data sets taken for statistical analysis after removal of outliers.

Table 4 — Acetaldehyde

Sample description	N ^a	Mean	<i>r</i>	<i>R</i>
		(µg/cigarette)		
CM6	17	1 309,3	159,3	365
1R5F	16	1 363,6	225,4	434
3R4F	18	1 605,8	185,5	389
1	14	1 191,9	150,8	221
2	14	1 341,5	200,3	315
3	13	1 361,3	137,5	325
4	14	1 054,6	157,6	444
5	15	836,5	169,4	595
6	14	1 209,6	112,9	222
7	14	1 175,4	159,0	321

^a N = number of data sets taken for statistical analysis after removal of outliers.

Table 5 — Acetone

Sample description	N ^a	Mean	r	R
		(µg/cigarette)		
CM6	18	517,3	64,8	172
1R5F	19	488,6	103,1	258
3R4F	18	596,8	81,3	180
1	15	473,2	69,8	149
2	14	490,8	67,4	179
3	13	496,3	50,9	192
4	14	369,1	71,3	179
5	14	281,9	51,2	227
6	15	475,6	62,1	161
7	13	415,5	57,5	156

^a N = number of data sets taken for statistical analysis after removal of outliers.

Table 6 — Acrolein

Sample description	N ^a	Mean	r	R
		(µg/cigarette)		
CM6	17	133,4	16,6	37
1R5F	17	121,6	23,6	53
3R4F	18	155,4	21,9	41
1	14	95,4	15,0	21
2	14	125,9	20,4	41
3	13	125,6	15,1	39
4	14	127,6	25,7	66
5	14	95,3	20,6	80
6	14	138,8	19,8	31
7	14	125,6	21,7	36

^a N = number of data sets taken for statistical analysis after removal of outliers.

Table 7 — Propionaldehyde

Sample description	N ^a	Mean	<i>r</i>	<i>R</i>
		(µg/cigarette)		
CM6	18	104,5	16,1	39
1R5F	18	98,5	17,9	47
3R4F	18	122,5	15,8	39
1	15	87,3	12,1	31
2	14	99,4	16,9	35
3	13	100,5	10,3	26
4	14	80,5	13,7	34
5	15	60,0	13,5	44
6	15	95,3	13,0	43
7	15	83,7	12,4	55

^a N = number of data sets taken for statistical analysis after removal of outliers.

Table 8 — Crotonaldehyde

Sample description	N ^a	Mean	<i>r</i>	<i>R</i>
		(µg/cigarette)		
CM6	17	47,9	8,1	17
1R5F	18	35,6	8,6	24
3R4F	18	49,9	10,4	22
1	14	40,2	6,8	12
2	14	41,6	8,6	18
3	14	43,9	9,0	19
4	14	40,2	10,0	21
5	14	30,4	8,1	25
6	15	46,0	8,1	13
7	14	35,8	8,2	15

^a N = number of data sets taken for statistical analysis after removal of outliers.

Table 9 — 2-butanone

Sample description	N ^a	Mean	r	R
		(µg/cigarette)		
CM6	18	131,4	27,7	69
1R5F	19	111,6	25,7	85
3R4F	18	147,0	30,3	79
1	14	112,7	21,6	57
2	14	118,5	22,8	62
3	14	121,2	22,2	65
4	15	89,9	22,7	66
5	14	69,6	18,2	32
6	15	116,2	24,6	66
7	13	95,7	19,3	47

^a N = number of data sets taken for statistical analysis after removal of outliers.

Table 10 — 2-butyraldehyde

Sample description	N ^a	Mean	r	R
		(µg/cigarette)		
CM6	16	69,8	14,1	33
1R5F	17	58,9	15,1	40
3R4F	17	71,8	11,3	31
1	13	52,4	7,0	14
2	13	62,0	13,1	26
3	14	65,7	9,4	30
4	14	48,9	12,3	29
5	13	40,2	9,0	34
6	14	62,4	11,3	21
7	14	46,8	9,2	34

^a N = number of data sets taken for statistical analysis after removal of outliers.

13 Test report

The test report shall state all tested product(s) each with unique identification, reference to the smoking regime used for sample generation, the yield of selected carbonyls in micrograms per cigarette smoked, and the method used. The test report shall include all conditions not specified in this document and deviations which can affect the result. All information should be recorded in fully traceable manner.

Annex A (informative)

Recrystallization of 2,4-dinitrophenylhydrazine

The supplied DNPH can contain contaminants or impurities. In this case, recrystallization of DNPH is recommended.

Weigh approximately 35 g of DNPH into a weighing boat. Transfer the DNPH into a clean 2 l Erlenmeyer flask and add a stirrer.

Add 750 ml of anhydrous reagent grade ethanol to the flask. Place the flask on a hot plate equipped with a stirrer. Gently heat the solution with constant stirring.

When the solution is warm, slowly add 1 000 ml of ethyl acetate. Continue to heat and stir (making sure not to boil) until all of the DNPH is completely dissolved. The solution should be clear and a very dark red.

Vacuum filter the hot solution.

Transfer the filtrate to a 2 l Erlenmeyer flask.

If crystallization does not start to occur, scratch the inside of the flask with a glass rod. Cover the Erlenmeyer flask with a watch glass and allow the solution to cool overnight in a cupboard.

Vacuum filter the recrystallized DNPH.

Transfer the crystals into a clean weighing boat that is labelled with the date of recrystallization and the lot of the DNPH. Weigh the recrystallized DNPH. Place the crystals in a desiccator to remove any moisture.

The filtrate can be evaporated down with a rotovap and vacuum filtered again to recover more crystals.

If recrystallizing a larger quantity of DNPH (requirement of more than 2 days), the DNPH shall be hydrated to approximately 30 % with water. After adding the water, place in an airtight container and label it as containing 30 % water.

Other methods for recrystallization may be used.