

First edition
2019-10

AMENDMENT 1
2021-08

**Gaseous hydrogen — Fuelling
stations —**

Part 8:
Fuel quality control

**AMENDMENT 1: Alignment with Grade D
of ISO 14687**

Hydrogène gazeux — Stations de remplissage —

Partie 8: Contrôle qualité du carburant

AMENDMENT 1: Alignement avec le Grade D de l'ISO 14687



Reference number
ISO 19880-8:2019/Amd.1:2021(E)

© ISO 2021

STANDARDSISO.COM : Click to view the full PDF of ISO 19880-8:2019/Amd 1:2021



COPYRIGHT PROTECTED DOCUMENT

© ISO 2021

All rights reserved. Unless otherwise specified, or required in the context of its implementation, no part of this publication may be reproduced or utilized otherwise in any form or by any means, electronic or mechanical, including photocopying, or posting on the internet or an intranet, without prior written permission. Permission can be requested from either ISO at the address below or ISO's member body in the country of the requester.

ISO copyright office
CP 401 • Ch. de Blandonnet 8
CH-1214 Vernier, Geneva
Phone: +41 22 749 01 11
Email: copyright@iso.org
Website: www.iso.org

Published in Switzerland

Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

The procedures used to develop this document and those intended for its further maintenance are described in the ISO/IEC Directives, Part 1. In particular, the different approval criteria needed for the different types of ISO documents should be noted. This document was drafted in accordance with the editorial rules of the ISO/IEC Directives, Part 2 (see www.iso.org/directives).

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights. Details of any patent rights identified during the development of the document will be in the Introduction and/or on the ISO list of patent declarations received (see www.iso.org/patents).

Any trade name used in this document is information given for the convenience of users and does not constitute an endorsement.

For an explanation of the voluntary nature of standards, the meaning of ISO specific terms and expressions related to conformity assessment, as well as information about ISO's adherence to the World Trade Organization (WTO) principles in the Technical Barriers to Trade (TBT) see www.iso.org/iso/foreword.html.

This document was prepared by Technical Committee TC 197, *Hydrogen technologies*.

A list of all parts in the ISO 19880 series can be found on the ISO website.

Any feedback or questions on this document should be directed to the user's national standards body. A complete listing of these bodies can be found at www.iso.org/members.html.

STANDARDSISO.COM : Click to view the full PDF of ISO 19880-8:2019/Amd 1:2021

Gaseous hydrogen — Fuelling stations —

Part 8: Fuel quality control

AMENDMENT 1: Alignment with Grade D of ISO 14687

Clause 5, first paragraph

Replace the paragraph with the following:

The quality requirements of hydrogen fuel dispensed to PEM fuel cells for road vehicles are listed in Grade D of ISO 14687.

8.4, first paragraph

Replace the paragraph with the following:

It is necessary to evaluate the possible consequences on a fuel cell vehicle if any impurity exceeds the threshold value of ISO 14687 Grade D.

8.4, second paragraph

Replace the paragraph with the following:

An estimation of the concentration above the ISO 14687 Grade D threshold values at which the severity increases (if applicable) is named “Level 1” and is given in column 5 for each impurity where the “severity class” is not already 4.

Table 4

Replace Table 4 with the following table:

Table 4 — Impact of impurities on fuel cell powertrain

Impurity		ISO 14687 Grade D threshold value ^a [μmol/mol]	Severity class (from ISO 14687 Grade D threshold value to Level 1)	Level 1 value [μmol/mol]	Severity class (greater than Level 1 threshold)
Total non-H ₂ gases		300	UD ^b	UD ^b	4
Helium	He	300	UD ^b	UD ^b	4
Nitrogen	N ₂	300	UD ^b	UD ^b	4
Argon	Ar	300	UD ^b	UD ^b	4
Oxygen	O ₂	5	UD ^c	UD ^c	4
Carbon dioxide	CO ₂	2	1	3	4
Carbon monoxide	CO	0,2	2-3 ^d	1	4
Methane	CH ₄	100	1	300	4
Water	H ₂ O	5	4	N/A	4
Total sulphur compounds	H ₂ S basis	0,004	4	N/A	4
Ammonia	NH ₃	0,1	4	N/A	4
Total hydrocarbons except methane	CH ₄ basis	2	1-4 ^d	N/A	4
Formaldehyde	HCHO	0,2	2-3 ^d	1	4
Formic acid	HCOOH	0,2	2-3 ^d	1	4
Halogens		0,05	4	N/A	4
Maximum particulate concentration (liquid and solid) ^e		1 mg/kg	4	N/A	4

Key

UD: undetermined

N/A: not applicable

^a The threshold value is according to hydrogen specification of Grade D of ISO 14687.

^b The severity class (from ISO 14687 Grade D threshold value to Level 1) and Level 1 value for this impurity is undetermined because no specific study has been conducted yet in alignment with the new threshold value. It needs to be covered in the next edition of this document.

^c The severity class (from ISO 14687 Grade D threshold value to Level 1) and Level 1 value for oxygen are undetermined because data are lacking to confirm those values. It needs to be covered in the next edition of this document.

^d A higher value is to be considered for risk assessment approach until more specific data is available.

^e Particulates are based upon mass density mg/kg.

A.15 first paragraph

Replace “ISO 14687-2” with “Grade D of ISO 14687” in the second last sentence.

Table B.1

Replace Table B.1 with the following table:

Table B.1 — Probability of occurrence for off-site SMR

Impurity	Threshold μmol/mol	Possible causes For the source studied	Typical barriers employed in this process	Probability with barriers
Inert gas N₂	300	Present in natural gas and syngas PSA malfunction	— PSA — Double analysis PSA outlet <100 μmol/mol	UD ^a
Inert gas Ar	300	Only ATR and POx present in O ₂ typical 0,6 % in syngas from ATR	— PSA. Not sized to remove Ar. Ar content may be higher if H ₂ comes from ATR, POX or feeds with high Ar content	UD ^a
O₂	5	Not present in syngas. O ₂ is unstable in the condition of reforming and shift reactions. Combines with H ₂ , CO, and CH ₄	— PSA cannot be used with significant O ₂ content for safety reasons	0
CO₂	2	Present in syngas (%)	— PSA adsorption strength of MS, activated carbon, silicagel higher for CO ₂ than CO. A CO content lower than 10 μmol/mol insures a CO ₂ content lower than 2 μmol/mol	0
CO	0,2	Normal operation below threshold. Occasional peaks at μmol/mol level	— Double analysis at the PSA outlet + trip if the CO > 1-10 μmol/mol at PSA outlet	4
CH₄	100	Present in syngas at % level	— In most cases CO is sizing the PSA, therefore CO < 10 μmol/mol ==> CH ₄ < 100 μmol/mol depending on users' specification (Europe pipeline 2 μmol/mol).	2
H₂O	5	Syngas saturated in H ₂ O	— PSA adsorbed in alumina and MS adsorption strength higher than CO ₂ . A CO content lower than 10 μmol/mol insures a H ₂ O content lower than 5 μmol/mol.	0
Key				
UD: undetermined				
^a The probability of occurrence for this impurity is undetermined because no specific study has been conducted yet.				

Table B.1 (continued)

Impurity	Threshold µmol/mol	Possible causes For the source studied	Typical barriers employed in this process	Probability with barriers
TS	0,004	TS from natural gas	— Desulphuration upstream reformer (typical values: normal < 10 ppb, maximum < 20 ppb, guarantee < 50 ppb)	0
			— Typical dilution factor 2,5 (1 mole natural gas produces 2,5 mole H ₂)	
			— Pre-reformer catalyst poisoning by sulphur is irreversible. Sulphur trapped at this stage. In case of breakthrough, process condition cannot be achieved	
			— Reformer catalyst poisoning by sulphur is irreversible. Sulphur trapped at this stage. In case of breakthrough, process condition cannot be achieved	
			— Shift catalyst poisoning by sulphur is irreversible. Sulphur trapped at this stage. In case of breakthrough, process condition cannot be achieved	
			— PSA adsorption of H ₂ S before CO, CO ₂ , species	
NH₃	0,1	Traces present in syngas	— H ₂ S adsorption in pipe and vessels. Strong affinity with steel	0
			— PSA adsorption strength of alumina and molecular sieve higher than CO. A CO content lower than 10 µmol/mol insures a NH ₃ content lower than 0,1 µmol/mol	
THC	2	Traces of C2+ after reforming reaction	— PSA C2 C3, C4, C5+ adsorbed by activated carbon layer. A CO content lower than 10 µmol/mol insures a THC (C ₂ H ₄ excluded) content lower than 2 µmol/mol	0
			— PSA. Formaldehyde adsorption strength of alumina and molecular sieve higher than CO. A CO content lower than 10 µmol/mol insures a HCHO content lower than 0,1 µmol/mol. To guarantee 0,01 µmol/mol would require more experience of measuring at those levels	
HCHO	0,2	May be present in syngas. essentially liquid		UD ^a
HCOOH	0,2	May be present in syngas essentially liquid	— PSA. Formic adsorption strength of alumina and molecular sieve higher than CO. A CO content lower than 10 µmol/mol insures a HCOOH content lower than 0,2 µmol/mol	0
Key				
UD: undetermined				
^a The probability of occurrence for this impurity is undetermined because no specific study has been conducted yet.				

Table B.1 (continued)

Impurity	Threshold $\mu\text{mol/mol}$	Possible causes For the source studied	Typical barriers employed in this process	Probability with barriers
Halogens	0,05	Present in natural gas	<ul style="list-style-type: none"> — Any Cl present in natural gas would be stopped by HDS — Pre-reformer catalyst poisoning by Cl irreversible. Cl trapped at this stage. If breakthrough, process condition cannot be achieved — Reformer catalyst poisoning by Cl irreversible. Cl trapped at this stage. I breakthrough, process condition cannot be achieved — Shift catalyst poisoning by Cl irreversible. Cl trapped at this stage. I breakthrough, process condition cannot be achieved — PSA adsorption of Cl before CO, CO₂, species 	0
He	300	Not present in natural gas in N Europe (<10 $\mu\text{mol/mol}$). Passes through the whole process. Dilution factor 2,5		0
Key				
UD: undetermined				
^a The probability of occurrence for this impurity is undetermined because no specific study has been conducted yet.				

Table B.2

Replace Table B.2 with the following table:

Table B.2 — Probability of occurrence for pipeline

Impurity	Threshold μmol/mol	Causes possible For the item studied	Typical barriers employed in this process	Probability with barriers
Inert gas N ₂	300	Air intake if some areas are at negative pressure From seal gas or purge gas Wrong purging after maintenance	Inlet pressure PSL trip on compressors	UD ^a
Inert gas Ar	300	No potential	1 % Ar in the air. 100 μmol/mol would mean 1 % air in the pipe Never been observed	UD ^a
O₂	5	Air intake if some areas are at negative pressure	Inlet pressure PSL trip on compressors	1
CO₂	2	From Air: CO ₂ at 400 μmol/mol in the air	2 μmol/mol of C O ₂ would mean 0.5 % air in the pipe Never been observed	0
CO	0,2	No potential		0
CH₄	100	No potential		0
H₂O	5	Wrong drying after pressure hydraulic test	H ₂ > 40 bar ==> leak from H ₂ O to H ₂ unlikely during operation.	0
TS	0,004	No potential		0
NH₃	0,1	No potential		0
THC	2	No potential		0
HCHO	0,2	No potential		UD ^a
HCOOH	0,2	No potential		0
Halogens	0,05	From cleaning material after maintenance		1
He	300	No potential		0

Key

UD: undetermined

^a The probability of occurrence for this impurity is undetermined because no specific study has been conducted yet. It needs to be covered in the next edition of this document.

Table B.3

Replace Table B.3 with the following table:

Table B.3 — Probability of occurrence for fuelling station to be source of impurities

Impurity	Threshold μmol/mol	Causes possible For the source studied	Existing barriers	Probability
Inert gas N ₂	300	N ₂ purging operation, air intake during normal operation or maintenance		UD ^a
Inert gas Ar	300	Air intake during normal operation or maintenance	1 % Ar in the air. 100 μmol/mol would mean 1 % air in the fuelling station Never been observed	UD ^a
O ₂	5	Air intake during normal operation or maintenance		2
CO ₂	2	Air intake during normal operation or maintenance	2 μmol/mol CO ₂ would mean 0,5 % air in the fuelling station. Never been observed	0
CO	0,2	No potential at fuelling station level		0
CH ₄	100	No potential at fuelling station level		0
H ₂ O	5	Maintenance, leaks from compressor exchangers, improper pressure vessel drying after periodic inspection, H ₂ O coming from the vent in case of check valve malfunction, depending on fuelling station/compressor technology		2
TS	0,004	Materials gaskets, valve seats and tubing	Material specifications	1
NH ₃	0,1	No potential		0
THC	2	Oil carryover from compressor (depending on compressor technology)		2
HCHO	0,2	No potential		UD ^a
HCOOH	0,2	No potential		0
Halogens	0,05	From degreasing material		1
He	300	No potential at fuelling station level	If pure He is not used for maintenance	0
Key				
UD: undetermined				
^a The probability of occurrence for this impurity is undetermined because no specific study has been conducted yet. It will be covered in the next edition of this document.				

Table B.4

Replace Table B.4 with the following table:

Table B.4 — Combined risk assessment

ISO specification		Supply chain probability			Compounded probability	Severity	Criticality	Additional risk reduction measures	Residual			
Impurity	Threshold µmol/mol	Production SMR	Pipeline distribution	Fuelling station					Severity reduction measures	Pro-ability	Se-verity	Criti-cality
Inert gas N ₂	300	UD ^a	UD ^a	UD ^a	UD ^a	UD ^a	UD ^a	Systematic N ₂ analysis after shutdown before resuming operation or specific purging procedure	None	UD ^a	UD ^a	UD ^a
Inert gas Ar	300	UD ^a	UD ^a	UD ^a	UD ^a	UD ^a	UD ^a		None	UD ^a	UD ^a	UD ^a
O ₂	5	0	1	2	2	UD ^b	UD ^b		None	2	UD ^b	UD ^b
CO ₂	2	0	0	0	0	1	+		None	0	1	+
CO	0,2	4	0	0	4	2	*	CO absorber at fuelling station design margin 100 % + operation procedure for replacement when H ₂ quantity purified = 50 % of design capacity.	None	1	2	+
CH ₄	100	2	0	0	2	1	+		None	2	1	+
H ₂ O	5	0	0	1	1	4	*	Check H ₂ O at commissioning and after maintenance involving opening of vessels or piping. Measurement shall be done at appropriate location downstream of the considered vessel or piping	None	0	4	+
TS	0,004	0	0	1	1	4	*	Check TS at commissioning and after maintenance involving parts modification (piping, valves, seals, gaskets). Not required for part replaced by identical component	None	0	4	+
NH ₃	0,1	0	0	0	0	4	+		None	0	4	+

Key

UD: undetermined

^a The probability of occurrence, the severity and the criticality for this impurity are undetermined because no specific study has been conducted yet. It needs to be covered in the next edition of this document.

^b The severity and the criticality for oxygen are undetermined because data are lacking to confirm those values.

Table B.4 (continued)

ISO specification	Supply chain probability			Severity	Criticality	Additional risk reduction measures	Residual			
	Production SMR	Pipeline distribution	Fuelling station				Severity reduction measures	Probability	Criticality	
Impurity Threshold µmol/mol										
THC	0	0	2	4	+	Oil/grease cleaning at commissioning and after maintenance. Compressor surveillance depending on compressor technology (coalescing filter) THC analysis or commissioning and after maintenance	None	0	4	+
HCHO	UD ^a	UD ^a	UD ^a	UD ^a	UD ^a		None	UD ^a	UD ^a	UD ^a
HCOOH	0	0	0	2	+		None	0	2	+
Halogens	0	0	1	4	*	Halogenated analysis at commissioning (species shall be defined) or after maintenance	None	0	4	+
He	0	0	0	4	+		None	0	4	+
Key				+		Acceptable risk area: existing controls sufficient	0		Unacceptable risk; additional control or barriers required	
Key										
UD: undetermined										
^a The probability of occurrence, the severity and the criticality for this impurity are undetermined because no specific study has been conducted yet. It needs to be covered in the next edition of this document.										
^b The severity and the criticality for oxygen are undetermined because data are lacking to confirm those values.										

Table B.5

Replace Table B.5 with the following table:

Table B.5 — Probability of occurrence for alkaline electrolysis

Impurity	Threshold μmol/mol	Possible causes For the source studied	Typical barriers employed in this process	Probability
Inert gas N ₂	300	Insufficient purging after shutdown	Electrolyser process control	UD ^a
Inert gas Ar	300	Insufficient purging after shutdown	If N ₂ below 1 %, Ar < 100 μmol/mol	UD ^a
O ₂	5	Insufficient purging after shutdown (does not apply to all systems) O ₂ permeation through the membrane	Process control Deoxo O ₂ sensor	2
CO ₂	2			0
CO	0,2			0
C H ₄	100			0
H ₂ O	5	H ₂ from electrolyser is saturated.	Dryer plus water sensor downstream of electrolyser	2
TS	0,004			0
NH ₃	0,1			0
THC (excluding CH ₄)	2			0
HCHO	0,2			UD ^a
HCOOH	0,2			0
Halogens	0,05	Cl from water?	Any trace of halogenat- ed compounds would be trapped in the dryer which has a stronger adsorption capacity for Cl than for H ₂ O	0
He	300			0
Key				
UD: undetermined				
^a The probability of occurrence for this impurity is undetermined because no specific study has been conducted yet in alignment with the new threshold value. It needs to be covered in the next edition of this document.				

C.2 first paragraph

Replace “ISO 14687-2” with “Grade D of ISO 14687”.

C.10 first paragraph

Replace “ISO 14687-2” with “Grade D of ISO 14687”.

Table C.4

Replace Table C.4 with the following table:

Table C.4 — Routine analysis work

Category: Distribution				
Facility type	Sampling/ Monitoring point	Contaminant	Threshold ($\mu\text{mol/mol}$)	Reduced frequency
Production of hydrogen from hydrocarbons utilizing steam reforming, catalytic reforming, partial oxidation, or ATR, purification using refining equipment, and distribution	Downstream of the purifier	TS ^a	0,004	Annual ^b
		THC as C1	2	Annual ^b
		CO	0,2	Annual ^b
		N ₂	300	Annual ^{b, c}
		Ar	300	Annual ^{b, c}
		H ₂ O	5	Annual ^b
		O ₂	5	Annual ^{b, c}
Electrolysis of NaCl for hydrogen, purification, and distribution	Downstream of the purifier	Halogens	0,05	Annual ^b
		N ₂	300	Annual ^b
		Ar	300	Annual ^b
		H ₂ O	5	Annual ^b
		O ₂	5	Annual ^b
Purification of coke-oven gas, and distribution	Downstream of the purifier	TS	0,004	Annual ^b
		THC as C1	2	Annual ^b
		CO	0,2	Annual ^b
		Halogens	0,05	Annual ^b
		N ₂	300	Annual ^b
		Ar	300	Annual ^b
		H ₂ O	5	Annual ^b
		O ₂	5	Annual ^b
		NH ₃	0,1	Annual ^b
		HCHO	0,2	Annual ^b
		HCOOH	0,2	Annual ^b
Purification of by-product hydrogen from ethylene plants, and distribution	Downstream of the purifier	TS	0,004	Annual ^b
		THC as C1	2	Annual ^b
		CO	0,2	Annual ^b
		N ₂	300	Annual ^b
		Ar	300	Annual ^b
		H ₂ O	5	Annual ^b
		O ₂	5	Annual ^b

^a Since sulfur in steam reforming is mostly found as hydrogen sulfide (H₂S), H₂S analysis may be performed in lieu of total sulfur analysis.

^b The minimum frequency of the analysis is premised on the combination of continuous analysis of the indicator species (CO for hydrocarbon cracking and O₂/H₂O for water electrolysis) and proper operation controls to make sure no mixing of impurities will occur. If that premise is not satisfied, the analysis shall be carried out once a day.

^c If a container or piping maintenance has not been opened to air or purged since the last hydrogen analysis, it is not necessary for analysis

^d If the 5 ppm (dew point of -66 °C or less) are satisfied by routine management, there is no need for analysis.

^e If the hydrogen on a pipeline contains odorant, the odorant shall be analysed with minimum frequency of once a year.

^f The batch analysis once a year is needed even if a continuous analysis is performed.

Table C.4 (continued)

Category: Fuelling station				
Facility type	Sampling/ Monitoring point	Contaminant	Threshold (µmol/mol)	Reduced frequency
With off-site supply of transported compressed or liquid hydrogen	End of nozzle	Those not analysed by the distributor		Annual ^b
		N ₂	300	Annual ^{b, c}
		Ar	300	Annual ^{b, c}
		H ₂ O	5	Annual ^{b, d}
		O ₂	5	Annual ^{b, c}
With off-site supply from hydrogen pipelines	Downstream of the deodorant equipment	(Those listed for the odorant)		Annual ^{b, e}
	End of nozzle	Those not analysed by the distributor		Annual ^b
		N ₂	300	Annual ^{b, c}
		Ar	300	Annual ^{b, c}
		H ₂ O	5	Annual ^{b, d}
O ₂	5	Annual ^{b, c}		
With on-site supply of hydrogen produced from hydrocarbons utilizing steam reforming, catalytic reforming, partial oxidation, or ATR and purification using refining equipment	Downstream of the purifier	CO	0,2	Continuous and Annual ^{b, f}
	End of nozzle	TS ^a	0,004	Annual ^b
		THC as C1	2	Annual ^b
		CO	0,2	Annual ^b
		N ₂	300	Annual ^{b, c}
		Ar	300	Annual ^{b, c}
		H ₂ O	5	Annual ^{b, d}
		O ₂	5	Annual ^{b, c}
With on-site supply from hydroelectrolysis and purification using refining equipment	Downstream of the purifier	N ₂	300	Annual ^{b, c}
		Ar	300	Annual ^{b, c}
		H ₂ O	5	Continuous and Annual ^{b, f}
		O ₂	5	Continuous and Annual ^{b, f}
	End of nozzle	Halogens	0,05	Annual ^b
		N ₂	300	Annual ^b
		Ar	300	Annual ^{b, c}
		H ₂ O	5	Annual ^b
O ₂	5	Annual ^b		

^a Since sulfur in steam reforming is mostly found as hydrogen sulfide (H₂S), H₂S analysis may be performed in lieu of total sulfur analysis.

^b The minimum frequency of the analysis is premised on the combination of continuous analysis of the indicator species (CO for hydrocarbon cracking and O₂/H₂O for water electrolysis) and proper operation controls to make sure no mixing of impurities will occur. If that premise is not satisfied, the analysis shall be carried out once a day.

^c If a container or piping maintenance has not been opened to air or purged since the last hydrogen analysis, it is not necessary for analysis

^d If the 5 ppm (dew point of -66 °C or less) are satisfied by routine management, there is no need for analysis.

^e If the hydrogen on a pipeline contains odorant, the odorant shall be analysed with minimum frequency of once a year.

^f The batch analysis once a year is needed even if a continuous analysis is performed.