
**Gaseous fire-extinguishing systems —
Physical properties and system design —**

**Part 6:
HCFC Blend A extinguishant**

*Systèmes d'extinction d'incendie utilisant des agents gazeux —
Propriétés physiques et conception des systèmes —*

Partie 6: Agent extincteur HCFC, mélange A

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Foreword

ISO (the International Organization for Standardization) is a worldwide federation of national standards bodies (ISO member bodies). The work of preparing International Standards is normally carried out through ISO technical committees. Each member body interested in a subject for which a technical committee has been established has the right to be represented on that committee. International organizations, governmental and non-governmental, in liaison with ISO, also take part in the work. ISO collaborates closely with the International Electrotechnical Commission (IEC) on all matters of electrotechnical standardization.

International Standards are drafted in accordance with the rules given in the ISO/IEC Directives, Part 2.

The main task of technical committees is to prepare International Standards. Draft International Standards adopted by the technical committees are circulated to the member bodies for voting. Publication as an International Standard requires approval by at least 75 % of the member bodies casting a vote.

Attention is drawn to the possibility that some of the elements of this document may be the subject of patent rights. ISO shall not be held responsible for identifying any or all such patent rights.

ISO 14520-6 was prepared by Technical Committee ISO/TC 21, *Equipment for fire protection and fire fighting*, Subcommittee SC 8, *Gaseous media and firefighting systems using gas*.

This second edition cancels and replaces the first edition (ISO 14520-6:2000), which has been technically revised.

ISO 14520 consists of the following parts, under the general title *Gaseous fire-extinguishing systems — Physical properties and system design*:

- *Part 1: General requirements*
- *Part 2: CF₃I extinguishant*
- *Part 5: FK-5-1-12 extinguishant*
- *Part 6: HCFC Blend A extinguishant*
- *Part 8: HFC 125 extinguishant*
- *Part 9: HFC 227ea extinguishant*
- *Part 10: HFC 23 extinguishant*
- *Part 11: HFC 236fa extinguishant*
- *Part 12: IG-01 extinguishant*
- *Part 13: IG-100 extinguishant*
- *Part 14: IG-55 extinguishant*
- *Part 15: IG-541 extinguishant*

Parts 3, 4 and 7, which dealt with FC-2-1-8, FC-3-1-10 and HCFC 124 extinguishants, respectively, have been withdrawn, as these types are no longer manufactured.

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Gaseous fire-extinguishing systems — Physical properties and system design —

Part 6: HCFC Blend A extinguishant

1 Scope

This part of ISO 14520 gives specific requirements for gaseous fire-extinguishing systems, with respect to the HCFC Blend A extinguishant. It includes details of physical properties, specification, usage and safety aspects and is applicable to systems operating at nominal pressures of 25 bar and 42 bar, superpressurized with nitrogen. This does not preclude the use of other systems.

2 Normative references

The following referenced documents are indispensable for the application of this document. For dated references, only the edition cited applies. For undated references, the latest edition of the referenced document (including any amendments) applies.

ISO 14520-1:2006, *Gaseous fire-extinguishing systems — Physical properties and system design — Part 1: General requirements*

3 Terms and definitions

For the purposes of this document, the terms and definitions given in ISO 14520-1 apply.

4 Characteristics and uses

4.1 General

Extinguishant HCFC Blend A shall comply with the specification according to Table 1. The tolerances of its components shall be in accordance with Table 2.

HCFC Blend A is a colourless, electrically non-conductive gas with a citrus-like odour and a density approximately 11 times that of air.

The physical properties are given in Table 3.

HCFC Blend A extinguishes fires mainly by physical means, but also by some chemical means.

Table 1 — Specification for HCFC Blend A

Property	Requirement
Purity	99,6 % by mass, min.
Acidity	3×10^{-4} % by mass (3 ppm), max.
Water content	10×10^{-4} % by mass (10 ppm), max.
Non-volatile residue	0,01 % by mass, max.
Suspended matter or sediment	None visible

Table 2 — HCFC Blend A components and tolerances

Component	Tolerance (by mass)
CHCl ₂ CF ₃	± 0,5
CHClF ₂	± 0,8
CHClFCF ₃	± 0,9
C ₁₀ H ₁₆	± 0,5

Table 3 — Physical properties of HCFC Blend A

Property	Unit	Value
Molecular mass	—	92,9
Boiling point at 1,013 bar (absolute) ^a	°C	−38,3
Freezing point	°C	< −107,2
Critical temperature	°C	125
Critical pressure	bar abs ^a	66,50
Critical volume	cm ³ /mol	170
Critical density	kg/m ³	580
Vapour pressure 20 °C	bar abs ^a	8,25
Liquid density 20 °C	kg/m ³	1 200
Saturated vapour density 20 °C	kg/m ³	31
Specific volume of superheated vapour at 1,013 bar and 20 °C	m ³ /kg	0,259
Chemical formulae	Component	%
	CHCl ₂ CF ₃	4,75
	CHClF ₂	82
	CHClFCF ₃	9,5
	C ₁₀ H ₁₆	3,75

^a 1 bar = 0,1 MPa = 10⁵ Pa; 1 MPa = 1 N/mm².

4.2 Use of HCFC Blend A systems

HCFC Blend A total flooding systems may be used for extinguishing fires of all classes within the limits specified in ISO 14520-1:2006, Clause 4.

The extinguishant requirements per volume of protected space are given in Table 4 for various levels of concentration. These are based on methods given in ISO 14520-1:2006, 7.6.

The extinguishing concentrations and design concentrations for heptane and Surface class A hazards are given in Table 5, and inerting concentrations in Table 6.

Table 4 — HCFC Blend A total flooding quantity

Temperature <i>T</i> °C	Specific vapour volume <i>S</i> m ³ /kg	HCFC Blend A mass requirements per unit volume of protected space, m/V (kg/m ³) This information refers only to HCFC Blend A, and may not represent any other products containing dichlorotrifluoroethane, chlorodifluoromethane, chlorotetrafluoroethane or isopropenyl-1-methylcyclohexane as components.									
		Design concentration (by volume)									
		7 %	8 %	9 %	10 %	11 %	12 %	13 %	14 %	15 %	16 %
−35	0,210	0,358	0,413	0,470	0,528	0,588	0,648	0,710	0,774	0,839	0,906
−30	0,215	0,351	0,405	0,461	0,517	0,576	0,635	696	0,758	0,822	0,887
−25	0,219	0,343	0,397	0,451	0,507	0,564	0,622	682	0,743	0,805	0,869
−20	0,224	0,337	0,389	0,442	0,497	0,553	0,610	668	0,728	0,790	0,852
−15	0,228	0,330	0,381	0,434	0,487	0,542	0,598	0,655	0,714	0,774	0,835
−10	0,232	0,324	0,374	0,426	0,478	0,532	0,587	0,643	0,700	0,760	0,819
−5	0,237	0,318	0,367	0,418	0,469	0,522	0,576	0,631	0,687	0,745	0,804
0	0,241	0,312	0,360	0,410	0,461	0,512	0,565	0,619	0,675	0,731	0,789
5	0,246	0,306	0,354	0,403	0,452	0,503	0,555	0,608	0,663	0,718	0,775
10	0,250	0,301	0,348	0,396	0,444	0,494	0,545	0,598	0,651	0,706	0,762
15	0,254	0,296	0,342	0,389	0,437	0,486	0,536	0,587	0,640	0,693	0,748
20	0,259	0,291	0,336	0,382	0,429	0,477	0,527	0,577	0,629	0,682	0,736
25	0,263	0,286	0,330	0,376	0,422	0,469	0,518	0,568	0,618	0,670	0,723
30	0,268	0,281	0,325	0,369	0,415	0,462	0,509	0,558	0,608	0,659	0,711
35	0,272	0,277	0,320	0,363	0,408	0,454	0,501	0,549	0,598	0,648	0,700
40	0,277	0,272	0,314	0,358	0,402	0,447	0,493	0,540	0,589	0,638	0,689
45	0,281	0,268	0,310	0,352	0,395	0,440	0,485	0,532	0,579	0,628	0,678
50	0,285	0,264	0,305	0,347	0,389	0,433	0,478	0,524	0,570	0,618	0,667
55	0,290	0,260	0,300	0,341	0,383	0,427	0,471	0,516	0,562	0,609	0,657
60	0,294	0,256	0,296	0,336	0,378	0,420	0,463	0,508	0,553	0,600	0,647
65	0,299	0,252	0,291	0,331	0,372	0,414	0,457	0,500	0,545	0,591	0,638
70	0,303	0,248	0,287	0,326	0,367	0,408	0,450	0,593	0,537	0,582	0,628
75	0,307	0,245	0,283	0,322	0,361	0,402	0,444	0,486	0,529	0,573	0,620
80	0,312	0,241	0,279	0,317	0,356	0,396	0,437	0,479	0,522	0,566	0,611
85	0,317	0,238	0,275	0,313	0,351	0,391	0,432	0,472	0,515	0,558	0,602
90	0,321	0,235	0,271	0,308	0,346	0,385	0,425	0,466	0,508	0,550	0,594
95	0,325	0,232	0,267	0,304	0,342	0,380	0,419	0,460	0,501	0,543	0,586

m/V is the agent mass requirement (in kilograms per cubic metre); i.e. mass, m , in kilograms of agent required per cubic metre of protected volume V to produce the indicated concentration at the temperature specified;

V is the net volume of hazard (in cubic metres); i.e. the enclosed volume minus the fixed structures impervious to extinguishant

$$m = \left(\frac{c}{100 - c} \right) \frac{V}{S}$$

T is the temperature (in degrees Celsius); i.e. the design temperature in the hazard area;

S is the specific volume (in cubic metres per kilogram); the specific volume of superheated HCFC Blend A vapour at a pressure of 1,013 bar may be approximated by

$$S = k_1 + k_2 T$$

where $k_1 = 0,241\ 3$; $k_2 = 0,000\ 88$

c is the concentration (in percent); i.e. the volumetric concentration of HCFC Blend A in air at the temperature indicated, and a pressure of 1,013 bar absolute.

Table 5 — HCFC Blend A reference extinguishing and design concentrations

Fuel	Extinguishment % by volume	Minimum design % by volume
Class B		
Heptane (cup burner)	10,0	13,0
Heptane (room test)	9,9	
Surface Class A		
Wood crib	6,0	a
PMMA	—	
PP	—	
ABS	—	
Higher Hazard Class A	b	12,4
<p>The extinguishment values for the Class B and the Surface Class A fuels are determined by testing in accordance with ISO 14520-1:2006, Annexes B and C.</p> <p>The minimum design concentration for the Class B fuel is the higher value of the heptane cup burner or room test heptane extinguishment concentration multiplied by 1,3.</p> <p>See ISO 14520-1:2006, 7.5.1.3, for guidance on Class A fuels.</p> <p>The extinguishing and design concentrations for room-scale test fires are for informational purposes only. Lower and higher extinguishing concentrations than those shown for room-scale test fires may be achieved and allowed when validated by test reports from internationally recognized laboratories.</p> <p>^a The minimum design concentration for Surface Class A fuel is the highest value of the wood crib, PMMA, PP or ABS extinguishment concentrations multiplied by 1,3. In the absence of any of the 4 extinguishment values, the minimum design concentration for Surface Class A shall be that of Higher Hazard Class A.</p> <p>^b The minimum design concentration for Higher Hazard Class A fuels shall be the higher of the Surface Class A or 95 % of the Class B minimum design concentration.</p>		

Table 6 — HCFC Blend A inerting and design concentrations

Fuel	Inertion % by volume	Minimum design % by volume
Methane	18,6	20,5
Propane	18,3	20,1
1,1-Difluoroethane (HFC-152a)	13,6	15,0
Difluoromethane (HFC-32)	8,6	9,5
Isobutane	18,4	20,2
Inerting concentrations derived in accordance with ISO 14520-1:2006, 7.5.2 and Annex D.		

5 Safety of personnel

Any hazard to personnel created by the discharge of HCFC Blend A shall be considered in the design of the system.

Potential hazards can arise from the following:

- the extinguishant itself;
- the combustion products of the fire;
- breakdown products of the extinguishant resulting from exposure to fire.

For minimum safety requirements, see ISO 14520-1:2006, Clause 5.

Toxicological information for HCFC Blend A is given in Table 7.

When the design concentrations exceed the LOAEL under normal design conditions, HCFC Blend A shall only be used for total flooding in normally unoccupied areas.

Table 7 — Toxicological information for HCFC Blend A

Property	Value % by volume
4 h LC ₅₀	64
No observed adverse effect level (NOAEL)	10
Lowest observed adverse effect level (LOAEL)	> 10
4 h LC ₅₀ is the concentration lethal to 50 % of a rat population during a 4 h exposure.	

6 System design

6.1 Fill density

The fill density of the container shall not exceed the values given in Tables 8 and 9 for 25 bar and 42 bar systems, respectively.

Table 8 — Storage container characteristics for HCFC Blend A — 25 bar

Property	Unit	Value
Maximum fill density	kg/m ³	900
Maximum container working pressure at 50 °C	bar ^a	35
Superpressurization at 20 °C	bar ^a	25
Reference should be made to Figure 1 for further data on pressure/temperature relationships.		
^a 1 bar = 0,1 MPa = 10 ⁵ Pa; 1 MPa = 1 N/mm ² .		

Table 9 — Storage container characteristics for HCFC Blend A — 42 bar

Property	Unit	Value
Maximum fill density	kg/m ³	900
Maximum container working pressure at 50 °C	bar ^a	53
Superpressurization at 20 °C	bar ^a	42
Reference should be made to Figure 2 for further data on pressure/temperature relationships.		
^a 1 bar = 0,1 MPa = 10 ⁵ Pa; 1 MPa = 1 N/mm ² .		

Exceeding the maximum fill density may result in the container becoming “liquid full”, with the effect that an extremely high rise in pressure occurs with small increases in temperature, which could adversely affect the integrity of the container assembly.

The relationships between pressure and temperature are shown in Figures 1 and 2 for maximum fill density.

6.2 Superpressurization

Containers shall be superpressurized with nitrogen with a moisture content of not more than 60×10^{-6} % by mass to an equilibrium pressure of $(25^{+1,25}_0)$ bar and $(42^{+2,1}_0)$ bar at a temperature of 20 °C (see Clause 1 for exception).